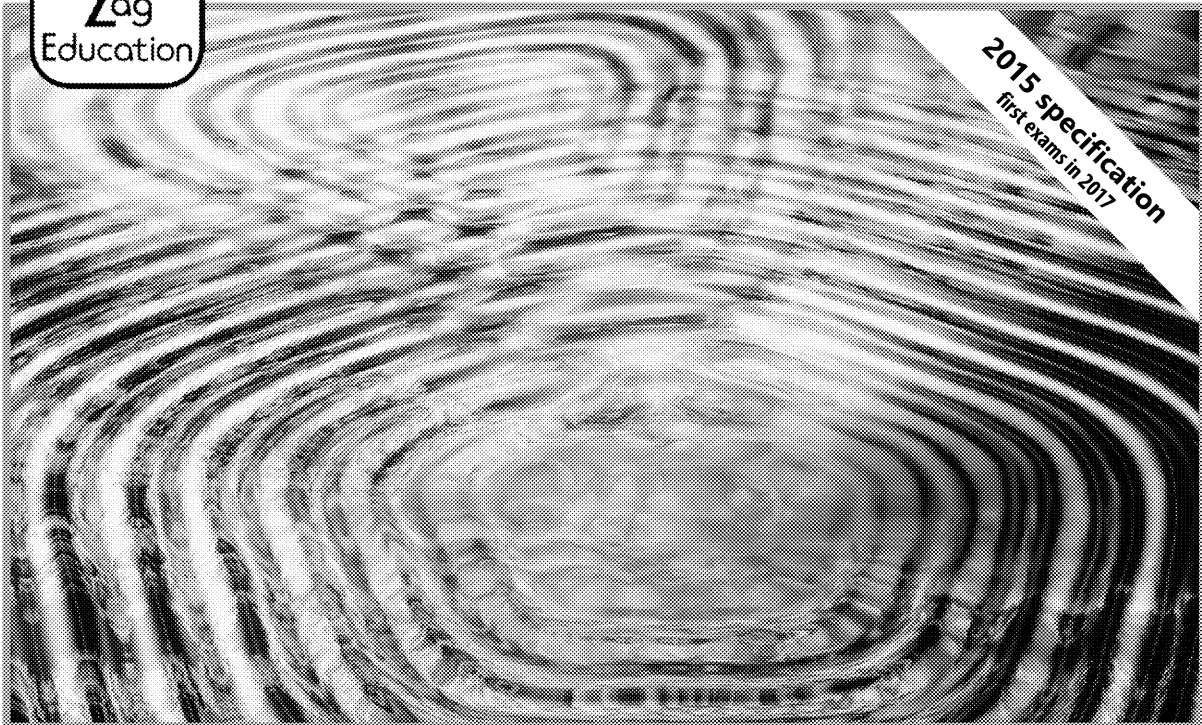




Physics

A Level | AQA | 7408



2015 specification
first exams in 2017

Practice Exams

for A Level AQA Physics

Paper 3B Option D: Turning Points in
Physics

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Teacher's Introduction

This collection of four practice papers has been written to support the AQA A Level physics specification 7408 (first examination 2016). The pack consists of four sets of Paper 3B Option D: Turning Points in Physics.

Each paper consists of 35 marks covering the content in the Turning Points in Physics optional unit, including a 6 mark question testing communication skills. Paper 3 section A (Paper 3A) and Paper 3 section B (Paper 3B) are sat in the same session. Students are given 2 hours to complete both papers 3A and 3B, for a combined total of 80 marks.

Each paper follows a similar format to the AQA papers. Every item listed in the specification is covered, with most aspects visited several times in the pack. Each set of papers matches the weightings of assessment objectives, maths skills and practical skills set out by the exam board.

The mark schemes are written in a similar format to those written by AQA. The individual marking points are on separate lines with additional guidance to clarify points and indicate alternative acceptable answers.

Suggested Uses

1. Set as a mock examination under exam conditions, marked by the teacher. This provides the most reliable summative assessment.
2. Set as a complete paper under exam conditions which is then marked by the student. This provides a good formative assessment as the student gets a good understanding of how the mark schemes work and what they need to do to score. Such a session could be reinforced by a lesson on exam technique.
3. Set as a complete paper under exam conditions which is then peer marked. This could be by the teacher assigning scripts to students to mark or by students swapping amongst themselves. Group marking can be particularly helpful as the students get the chance to develop their ideas by discussing why things do and don't score.
4. Go through a question at a time in a lesson. Get students to discuss their answers before revealing the mark scheme for that question.
5. Set a paper as a homework for the student to answer and mark. This would be an ideal activity for study leave, when the student could come to a tutorial to go through their script. They should be briefed to list questions that need addressing as a result of their marking of their script.

Remember!

Always check the exam board website for new information, including changes to the specification and sample assessment material.

Toby Brown & Samir Khonji, May 2017

Specification Cross-reference: Turning Points in Physics

Turning Points in Physics:	3B (A)	3B (B)	3B (C)	3B (D)
12 Turning points				
12.1 The discovery of the electron				
12.1.1 Cathode rays			1	
12.1.2 Thermionic emission of electrons			1	1
12.1.3 Specific charge of the electron	1, 2		1	
12.1.4 Principle of Millikan's determination of the electronic charge		1		
12.2 Wave-particle duality				
12.2.1 Newton's corpuscular theory		2		3
12.2.2 Significance of Young's double-slit experiment		2	3	3
12.2.3 Electromagnetic waves			2	3, 4
12.2.4 The discovery of photoelectricity				4
12.2.5 Wave-particle duality	2, 3		3	4
12.2.6 Electron microscopes	3			
12.3 Cosmology				
12.3.1 The Michelson-Morley experiment		3		4
12.3.2 Einstein's theory of special relativity		3	4	3, 4
12.3.3 Time dilation		4	4	
12.3.4 Length contraction		4	4	
12.3.5 Mass and energy	4	4	4	3

ZigZag Practice Exam Papers

Supporting A Level AQA Physics

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Practice Exam Paper 3B

Option D: Turning Points in Physics

Name	
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Time allowed

2 hours (for 3A and 3B)

Instructions

Answer **all** of the questions and use the space provided.

Information

The total marks available for this paper is **35**. The number of marks available for each question is shown on the right.

For this paper, you will need:

- Data and formula booklet

Additional materials required

- Pencil
- Electronic calculator
- Ruler (cm/mm)

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1. In 1909, Robert Millikan performed an experiment in which charged drop field generated by parallel charged plates, as seen in **Figure 1**.

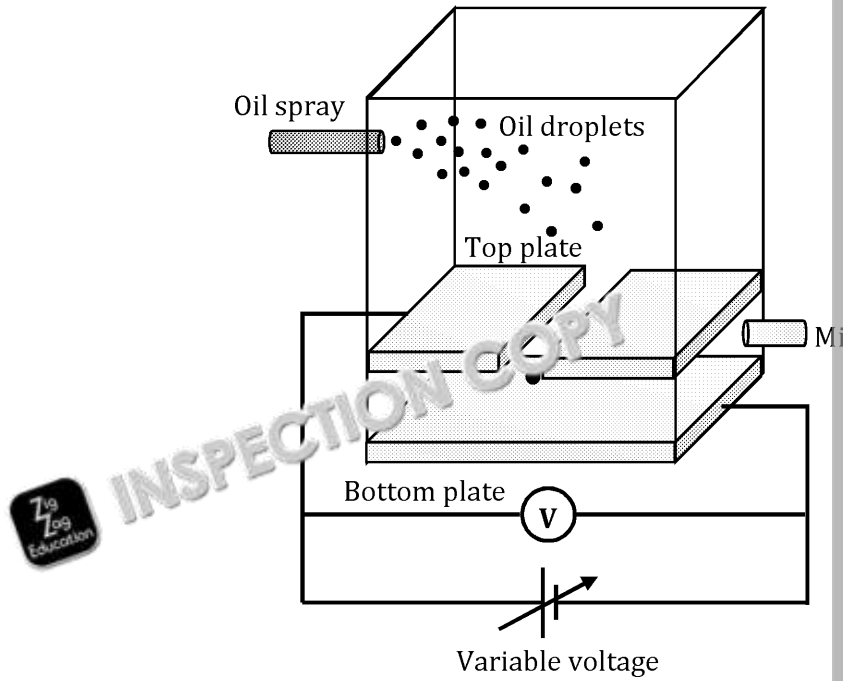


Figure 1

- 1.1 The variable voltage is first switched off and the oil drop falls with terminal velocity.

Show that the radius of the drop is given by

$$r = \left(\frac{9\eta v}{2\rho g} \right)^{\frac{1}{2}}$$

- 1.2 Explain how the radius and density of the oil drop can be used to determine the charge on the drop.

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1.3 The radius of an oil drop is measured to be 1.51×10^{-6} m, and the density is 1.26 g cm^{-3} .

The potential difference between the plates is 1120 V and the distance between the plates is 1.20 cm.

Show that the charge on the oil drop is 4.76×10^{-19} C

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1.4 The error on the value found in 1.3 is $\pm 0.05 \times 10^{-19}$ C. Discuss the accuracy of the value found in 1.3.

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2. In the late seventeenth century, the scientific community was locked in a debate about the nature of light – whether light is a particle or a wave.

Isaac Newton preferred the corpuscular theory of light, in which light is composed of particles, while Christiaan Huygens adopted the theory that light is a wave.

2.1 Discuss the evidence for both the corpuscular theory of light and Huygens' wave theory, including the factors that led to the initial acceptance of the corpuscular theory and the eventual wider acceptance of light as a wave.

Dotted lines for writing the answer to question 2.1.

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2.2 While there are many pieces of evidence of light acting as wave, in 1905 Albert Einstein proposed the photon theory of light, for which he later won the 1921 Nobel Prize.

Explain why light acting as a particle in the way theorised by Einstein is necessary to explain the photoelectric effect.

Dotted lines for writing the answer to question 2.2.

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3. The Michelson–Morley experiment was used to determine the velocity of theorised *aether*, the medium through which light was supposed to propagate.

3.1 Describe the Michelson–Morley experiment, stating the result expected.

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3.2 Why is it significant that the Michelson–Morley did **not** detect the predicted *aether* wind with respect to Earth?

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3.3 Describe how Einstein’s theory of special relativity resolved the issue raised by the Michelson–Morley experiment.

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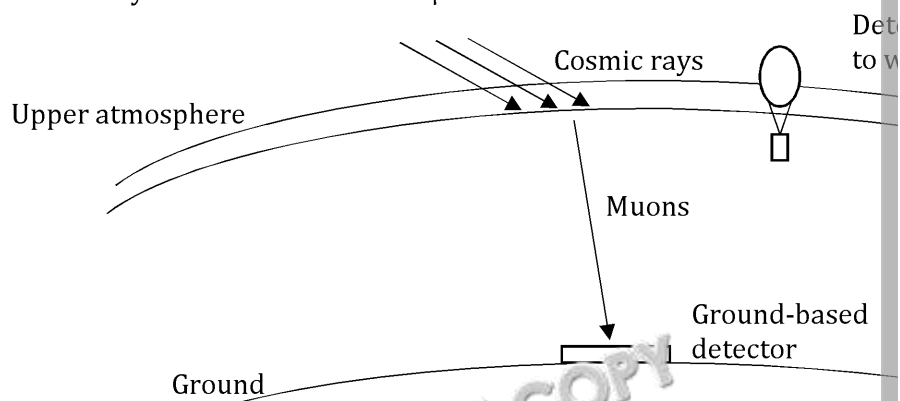
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4. Muons are unstable particles created when cosmic rays bombard particles in the upper atmosphere. Muons decay with a half-life of $1.56 \mu\text{s}$.



- 4.1 Muons are detected in the upper atmosphere travelling at a velocity of $2.96 \times 10^8 \text{ m s}^{-1}$. Show that about $8.7 \times 10^{-9} \%$ of the muons would be expected at ground level if all muons are emitted directly downwards.



- 4.2 The actual amount of muons detected at ground level is significantly greater than expected. Explain why this is the case.

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- 4.3 Show that the percentage of the muons actually detected is 8.69 %.

- 4.4 Calculate the total energy of a muon travelling at $2.96 \times 10^8 \text{ m s}^{-1}$.



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ZigZag Practice Exam Papers

Supporting A Level AQA Physics

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Practice Exam Paper 3B

Option D: Turning Points in Physics

Name	
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Time allowed

2 hours (for 3A and 3B)

Instructions

Answer **all** of the questions.

Information

The total marks available for this paper is **35**. The number of marks available for each question is shown on the right.

For this paper, you will need:

- Data and formula booklet

Additional materials required

- Pencil
- Electronic calculator
- Ruler (cm/mm)
- Graph paper

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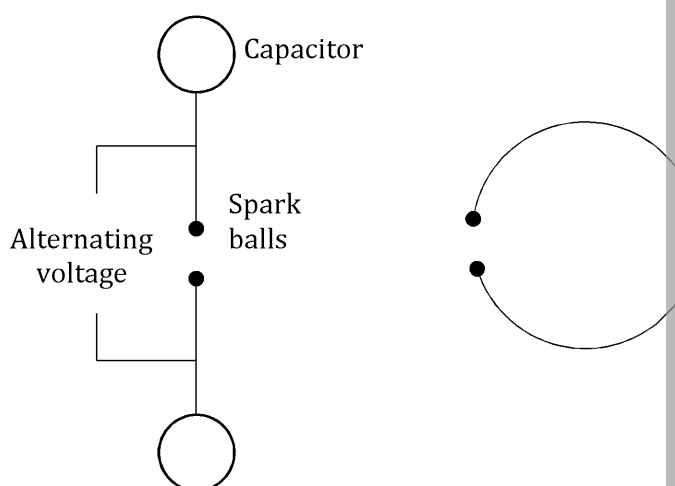
1. Cathode rays produced in low-pressure discharge tubes were used to identify the structure of atoms.
 - 1.1 Describe the structure of a discharge tube, and how cathode rays were produced.
 - 1.2 Discharge tubes were later developed into cathode ray tubes, as used in television sets. Cathode ray tubes accelerated electrons through a ring-shaped anode to a fluorescent screen, producing an image.

Explain why cathode ray tubes were fully evacuated of all air, while television sets required the presence of gas at low pressure.

- 1.3 J J Thomson later performed experiments on cathode rays, in which the beam of cathode rays was bent.

Explain how J J Thomson's work showed the particle nature of electrons.

2. Heinrich Hertz used the set-up below to confirm the nature of light as an electromagnetic wave.



Sparks were created in the gap between spark balls, and an alternating current was induced in the wire loop.

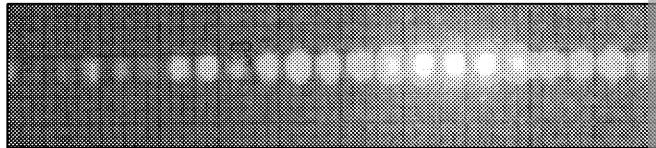
- 2.1 Explain why the results of this experiment confirmed the nature of light as an electromagnetic wave.
- 2.2 Describe how this effect is used in radio communications.
- 2.3 Hertz later used a similar set-up to find the positions of nodes and antinodes in a standing radio wave.

Explain how this allowed Hertz to determine the speed of light.

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3. In Young's double-slit experiment, light is passed through two narrow slits and forms a pattern on a distant screen, as seen below.



3.1 Explain how fringes are produced in Young's double-slit experiment.

3.2 Discuss the shortcomings of the wave theory of light to explain certain phenomena. In your discussion, you should write about:

- the problem with black-body radiation
- the problems with the photoelectric effect
- how the understanding of light changed to overcome these shortcomings

3.3 The classical particle theory of electrons cannot explain electron diffraction.

Explain why electron diffraction cannot be described by the particle theory of electrons.

4. A beam of protons moving at a relativistic speed passes between two detectors.

4.1 State the two postulates of Einstein's theory of special relativity.

4.2 The beam of protons moves at a speed of $0.859c$ in the frame of the detectors. The time between the first and the second detector is 1.55×10^{-7} s after it is detected by the first.

Calculate how long it takes the beam of protons to pass between the detectors in the rest frame of the protons.

4.3 The positions of the detectors are changed so that they are separated by a distance D in the rest frame of the detectors. The distance between the detectors in the rest frame of the protons is 54.1 m. Calculate the distance D .

4.4 Explain why the speed of the beam of protons may not exceed the speed of light. How can the kinetic energy of the beam increase without limit?

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Preview of Questions Ends Here

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Mark Scheme: Set A

Question	Answer	Mark
1.1	Electrons produced and accelerated towards anode ✓ Electrons deflected by electric field (upwards) ✓ Magnetic field varied until deflection of electron path decreases to zero ✓ Electric and magnetic forces equated and rearranged to find e/m ✓	1 1 1 1
1.2	$F_E = Ee$ $F_B = Bev$ $Ee = Bev$ (for forces to balance) $v = \frac{E}{B}$ ✓ $\frac{1}{2}mv^2 = eV$ $\frac{e}{m} = \frac{v^2}{2V}$ $\frac{e}{m} = \frac{(E/B)^2}{2V}$ ✓ $\left(\frac{e}{m} = \frac{E^2}{2VB^2}\right)$	1 1 1
1.3	Factor = $\frac{\text{specific charge of electron}}{\text{specific charge of hydrogen ion}} = \left(\frac{e/m_e}{e/m_H}\right)$ Factor $\sim \frac{10^{11}}{10^8}$ ✓ Factor $\sim 10^3$ ✓	1 1
2.1	$\lambda = \frac{h}{p}$ $E_k = \frac{p^2}{2m}$ E_k comes from the potential difference $E_k = eV$ $eV = \frac{p^2}{2m}$ ✓ $p = \sqrt{2mE_k}$ ✓ $\left(\lambda = \frac{h}{\sqrt{2meV}}\right)$	1 1
2.2	The electron accelerated through a lower potential difference is slower ✓ and so has a longer de Broglie wavelength ✓ so the gaps between fringes will be larger / the electron will be deflected more ✓	1 1 1
2.3	(Determination of $\frac{e}{m}$ for an electron suggests that) the electron has mass and so acts as a particle ✓ Diffraction is a property of waves and so electrons act as waves ✓ Shows wave-particle duality - all particles act as waves; all waves act as particles ✓	1 1 1

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Question	Answer	Mark	
3.1	Wavelength required = size of atom $\sim 10^{-10}$ m ✓	1	
	$\lambda = \frac{h}{\sqrt{2meV}}$		
	$V = \frac{h^2}{\lambda^2 2me}$	1	
	$V = \frac{(6.63 \times 10^{-34})^2}{(10^{-10})^2 \times 2 \times 9.11 \times 10^{-31} \times 1.60 \times 10^{-19}}$ ✓	1	
	$V = 151$ V ✓	1	
3.2	Lighter areas represent less dense areas which electrons can pass through ✓	1	
	Darker areas show denser areas which block electrons ✓	1	
3.3	The guidance below outlines the features of a 1-, 2-, 3-, 4-, 5- and 6-mark answer		
	Mark	Criteria	
	6	A detailed discussion of the operation of an STM, including both modes of operation and the advantages of STMs over TEMs.	Relevant, coherent and clear grammar and handwriting
	5	A detailed discussion of the operation of STMs, including both modes of operation and advantages, with very minor gaps in detail.	
	4	A discussion that includes the majority of main points about the operation, including different modes and advantages	Information is sufficient Spelling and handwriting
	3	A discussion that includes the main points about the operation and at least one point for advantages. There are gaps in knowledge and missing details.	
	2	Only a couple of points made correctly; significant knowledge is missing.	Some present and good derivation
	1	Only one correct point made about the operation, such as quantum tunnelling being utilised.	
0	No relevant information provided.	Presenting serious misunderstandings	
	<p><i>Max 6</i> – The following statements are likely to be included:</p> <p>Operation</p> <ul style="list-style-type: none"> • Tip of probe is short distance (1 nm) above sample • Electrons cross gap from sample to probe via quantum tunnelling • Quantum tunnelling because of the wave nature of electrons • Small voltage between tip of probe and sample to ensure electrons only cross gap • Mode 1: Height of probe kept constant, change in height of sample causes current to change • Mode 2: Height of probe changes to keep the current constant • Variation in current/height converted to map of heights of the sample <p>Advantages of STM over TEM</p> <ul style="list-style-type: none"> • Can be used in air and water • Can be used over a range of temperatures, exceeding 500 K • STM has resolution of 0.01 nm (depth) and 0.1 nm (laterally) • Loss of speed in TEM reduces focusing power • TEM much more affected by vibrations 		

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Question	Answer	Marks
4.1	<p>Correct shape of graph (mass increasing with speed with increasing gradient) ✓ Up to asymptote at c ✓</p>	1 1
4.2	$E = mc^2 = \frac{m_0 c^2}{\sqrt{1 - \frac{v^2}{c^2}}}$ $m = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$ $2m_0 = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}} \checkmark$ $\sqrt{1 - \frac{v^2}{c^2}} = \frac{1}{2}$ $1 - \frac{v^2}{c^2} = \frac{1}{4}$ $\frac{v^2}{c^2} = \frac{3}{4}$ $v^2 = \frac{3}{4} c^2 \checkmark$ $v = \frac{\sqrt{3}}{2} c \checkmark$	1 1 1
4.3	<p>Direct evidence of variation of kinetic energy with speed ✓ Speed of particles limited to below c ✓</p>	1 1



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