

# Topic Tests

for WJEC AS Biology

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# Teacher's Introduction

These topic tests have been designed to help you and your students assess their knowledge of a topic after you have taught each section of the **WJEC AS Biology specification**. Each topic test is closely tied to the WJEC specification, ensuring all aspects of the course will be covered. The units covered in these tests are Unit 1 and Unit 2. The table on the right shows the number of marks given for each topic within the two units.

Each topic test provides a variety of question styles, including:

- **Quick-testing questions** – these test basic understanding and knowledge of terminology, and allow immediate identification of weaker topics
- **Long-answer questions** – these are exam-style questions that require use of comprehensive knowledge and aid practice of writing skills and exam technique
- **Missing information questions** – these allow key knowledge to be tested without it being time-consuming and provide context for further questions
- **Diagram and graph-dependent questions** – these require identification of features, interpretation of data and application of knowledge, as well as testing mathematical skills
- **Practical questions** – cover aspects of practicals from planning and risk awareness to data analysis and evaluation, as well as testing all mathematical skills.
- **Context-dependent questions** – these push students to apply their knowledge to unfamiliar situations, spot key points within provided information and draw on multiple aspects of the course.

Tests have been aimed to take approximately 45 minutes and contain on average between 40 and 50 marks, though please note that this has not been possible where topics are brief and introductory or require more detailed knowledge and assessment. Larger topics have been split into two tests. All information for a question is provided within the test; however, some questions will require use of a calculator and ruler.

Students are able to see the number of marks allocated for each question, allowing them to judge the detail required in their answers as in exam conditions. Full answers are at the end of the resource and are accompanied by marker instructions, providing quick guidelines on what answers would and would not be accepted in exam conditions.

All diagrams and graphs have been designed with black-and-white photocopying in mind, so key features will not be lost.

We hope you find these tests useful during your teaching.

## Remember!

Always check the exam board website for new information, including changes to the specification and sample assessment material.

### AS Unit 1: Basic Biochemistry and Cell Organisation

Topic number	Number of marks
1	36 + 44
2	49
3	45
4	40
5	48 + 45
6	40

### AS Unit 2: Biodiversity and Physiology of Body Systems

Topic number	Number of marks
1	50
2	50
3	50 + 49
4	47

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\* resulting from minor specification changes, suggestions from teachers and peer reviews, or occasional errors reported by customers

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## 1. All organisms are related through their evolution

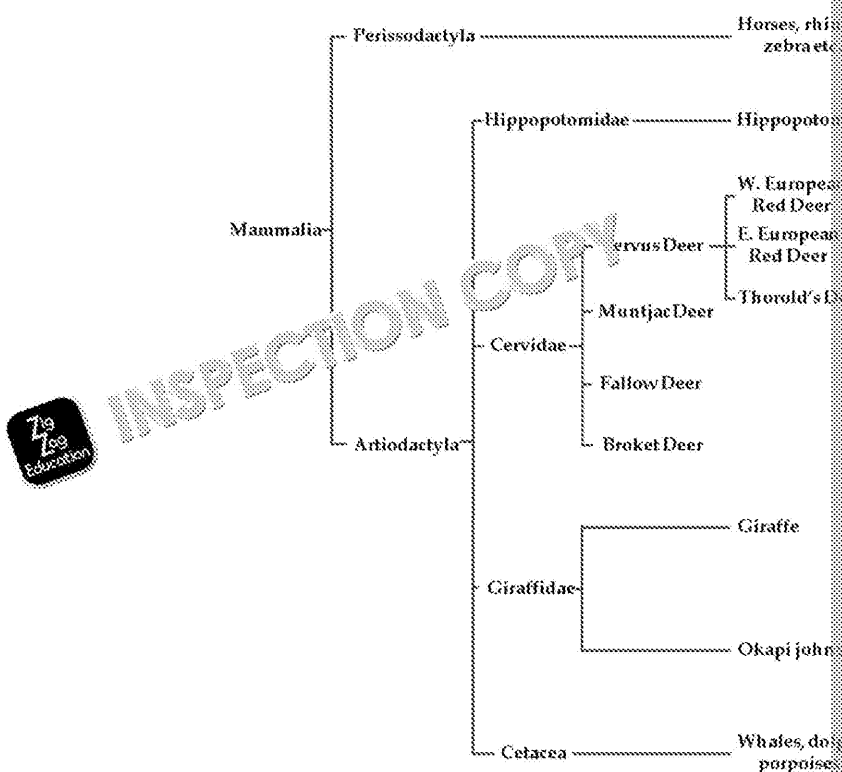
1. a) What is a species?  
b) What is a binomial name? Why is it useful for every species to have a binomial name?
2. Copy and fill in the following taxonomic hierarchy:

Highest – → \_\_\_\_\_  
Domain

3. Copy and fill in the missing information in the table describing features of the three domains of life.

Domain	Bacteria	Archaea	Eukarya
Kingdom	Prokaryotes	Prokaryotes	Animals, Plants, Fungi, Protists
Features	Single-celled prokaryotes	Single-celled prokaryotes	Single-celled eukaryotes
	Small 70S ribosomes	Small 70S ribosomes	Large 80S ribosomes
	Cell wall	Cell wall	Cell membrane

4. Phylogenetic relationships show evolutionary relationships, and are based on the analysis of different organisms.  
Name a method used to assess DNA at the molecular level.  
Use the phylogenetic classification diagram below to answer the following questions:  
a) What is the closest related order to artiodactyla?  
b) Whales, dolphins and porpoises all belong to one family. Identify the family.  
c) What genus do red deer belong to?  
d) What is the closest related species to the giraffes?

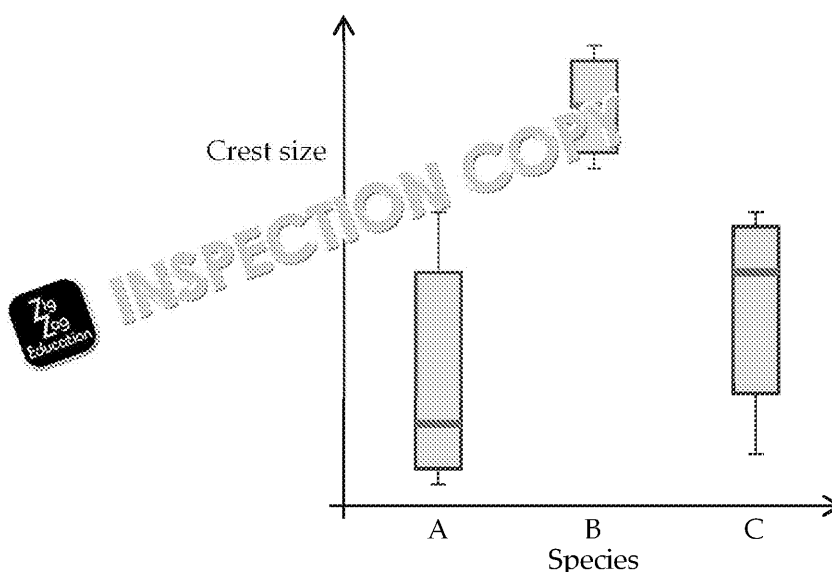


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5. Which three molecules can be used to assess genetic diversity?
6. The graph below shows the mean crest size (solid bar) in three different species. The box shows 90% standard deviation and the extending lines encompass the range of the data.
  - a) Explain the difference between 'mean' and 'standard deviation'.
  - b) Explain why two species could still be considered the same species but one is considered a separate species, based on their crest size.



7. Define 'biodiversity'.
8. Simpson's Diversity Index allows calculation of a habitat's diversity.
  - a) State the formula for Simpson's Diversity Index.
  - b) Calculate the diversity of the two habitats using the species data below.
  - c) Explain which habitat is agricultural land and which is ancient forest.
  - d) What species is the farmer growing?
  - e) What species has become locally extinct on the farm?

Habitat A		
Species	n	n(n-1)
B	63	
C	49	
D	86	
E	45	
F	21	

Species
A
B
C
E
F

9. Conservation is not only concerned about the number of individuals in a population but also about genetic variation. How is genetic diversity of alleles calculated?
10. What is meant by genetic diversity?

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11. a) Identify the four processes of evolution.  
b) Use the four principles of evolution to answer the following scenario.
- A mutation in pine trees caused their needles to have a light-green stripe to reduce water loss so the adaptation spread.
- Explain why, as the leaf-stripe became more common, the caterpillars started to show striped patterns.
12. Identify the three types of adaptation that allow animals to become adapted to their environment.



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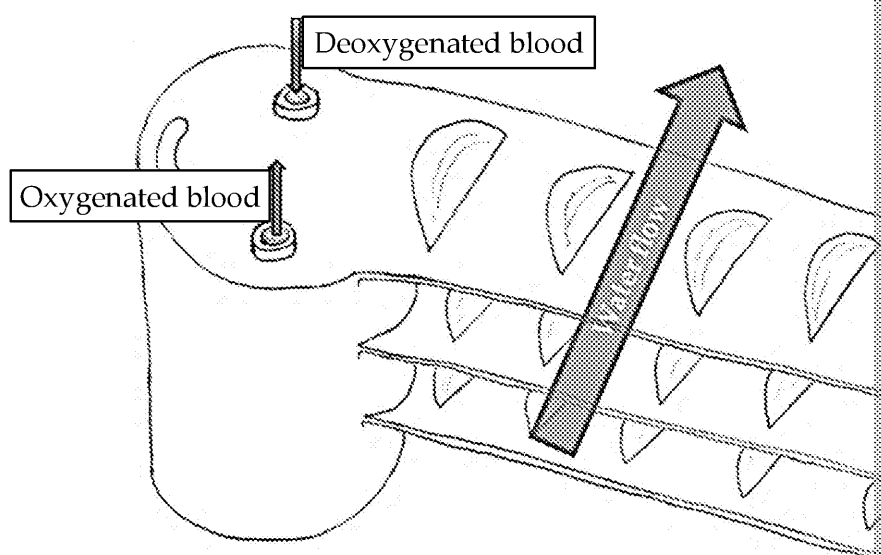
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## 2. Adaptations for gas exchange

1. What method of gas exchange is used by single-celled and small organisms?
2. Why do large, active animals need ventilating mechanisms to obtain oxygen?
3. Compare the gas exchange mechanisms of a flatworm and an earthworm.
4. Insects and plants both have to compromise gas exchange so they do not lose water.
  - a) Describe the structures insects have to allow gas exchange.
  - b) How do gases move through these structures?
  - c) Give two adaptations insects have to reduce water loss.
5. Fish provide a large surface area for gas exchange with their gills.
  - a) Label the passing parts of the gill below and draw on the flow of blood.
  - b) Write the name of the process of gas exchange in the gills?
  - c) Explain why the above process allows more oxygen to enter the blood.



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6. Match each human lung structure to its features and its function.

Bronchi	Minute air sacs lined with elastic fibres
Alveoli	Smaller, branching airways lined with smooth muscle fibres
Trachea	Large flexible airway supported by cartilage
Bronchioles	Two branches supported by cartilage

7. Explain how movement of air and blood in the lungs gas exchange.

8. Copy the text and fill in the missing words in the process below:

During inspiration, the \_\_\_\_\_ intercostal muscles contract and the \_\_\_\_\_ and \_\_\_\_\_ expand. This increases the volume, thereby decreasing the pressure and air enters the lungs. At rest, expiration and inspiration are \_\_\_\_\_ processes. During expiration during exercise, only the \_\_\_\_\_ intercostal muscles and the \_\_\_\_\_ expand.

9. Describe how gas exchange for photosynthesis and respiration takes place in a leaf.

10. An experiment was designed to calculate the number of stomata present on the upper and adaxial (top) surfaces of the leaf of a geranium. A replica of the upper surface of the leaf was made by applying two layers of clear nail varnish to the leaf and gently peeling it off. A microscope was used to count how many stomata were present in 10 fields of view.

Field of view	1	2	3	4	5	6	7
Abaxial surface	5	5	6	7	8	6	7
Adaxial surface	4	3	5	4	3	3	6

- (a) Calculate the mean number of stomata present in the field of view nearest whole number.
- (b) The eyepiece graticule is calibrated so that each eyepiece division represents 1 mm. The diameter of the field of view measures 90 eyepiece divisions. Use this information to calculate the area of the field of view in mm<sup>2</sup> to 1 decimal place.
- (c) Use the equation below to calculate the mean number of stomata per mm<sup>2</sup> of the leaf.

$$\text{Mean number of stomata per mm}^2 = \frac{\text{Mean number of stomata} \times \text{area of field of view}}{\text{area of field of view}}$$

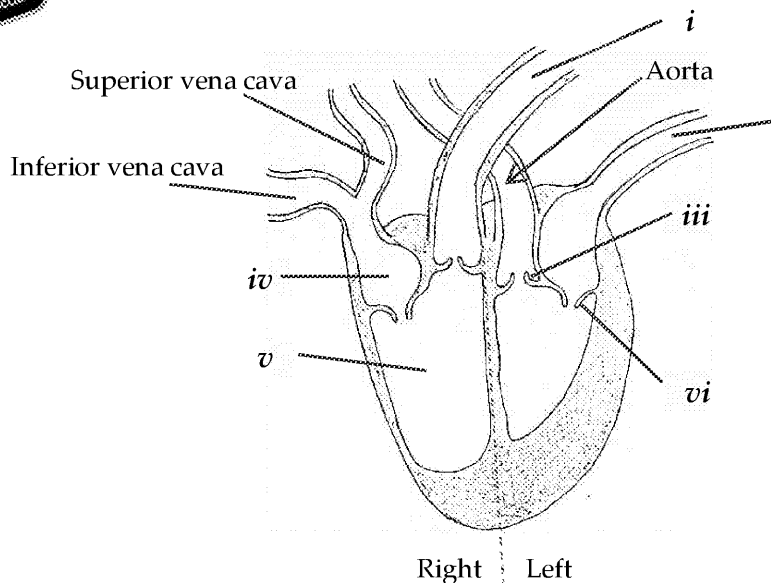
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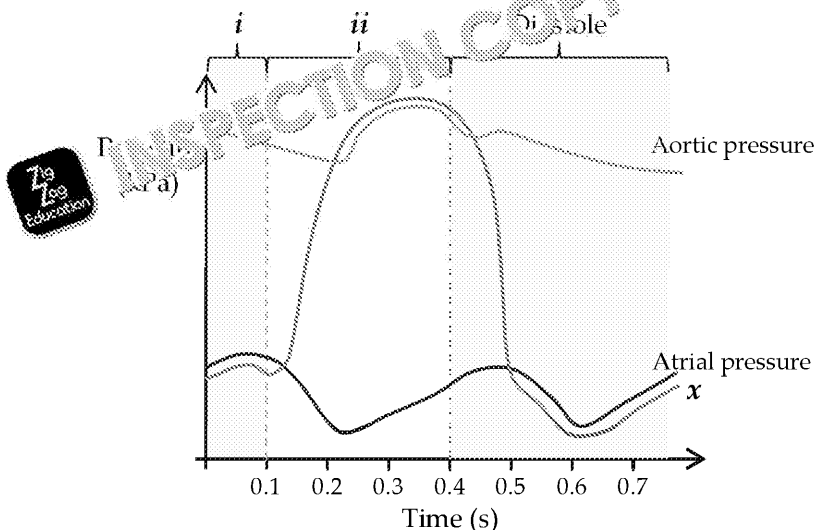


### 3. Adaptations for transport (1)

- Define each of the following types of system and give an example of each.
  - Single circulatory system
  - Double circulatory system
- What are the main differences between closed and open circulatory systems?
- Excluding substances for gaseous exchange, name one other substance transported in blood.
- The diagram below shows the human heart.
  - Fill in the missing labels for the heart valves and blood vessels.
  - Draw and label two arrows showing the movement of oxygenated blood.
  - How do the ventricles differ and why are they necessary?
  - Why is the left ventricular muscle of the ventricle thicker on the left side?



- The graph below shows pressure changes within the heart and aorta.
  - Name the two missing stages of contraction © and ii.
  - What does line x represent?
  - Circle and label where the semilunar valves and atrioventricular valves are closed.



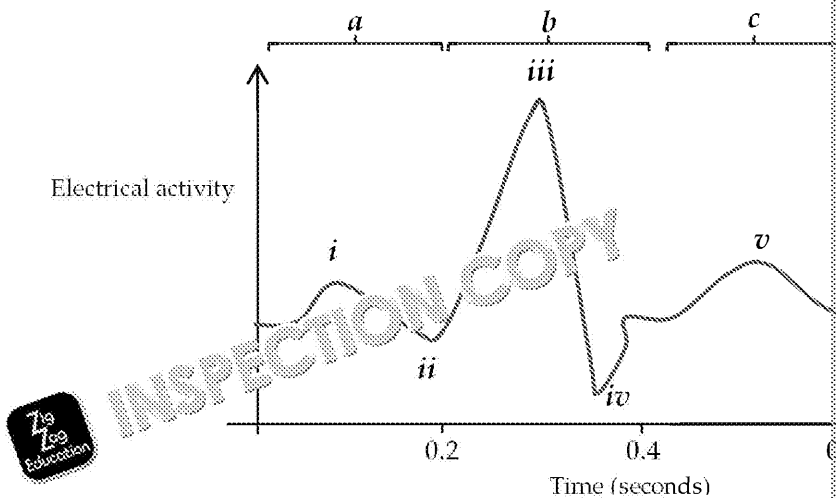
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6. a) Label parts *i* to *v* on the electrocardiogram shown below.  
 b) What does wave *a* represent?  
 c) What does wave *b* represent?  
 d) What does wave *c* represent?

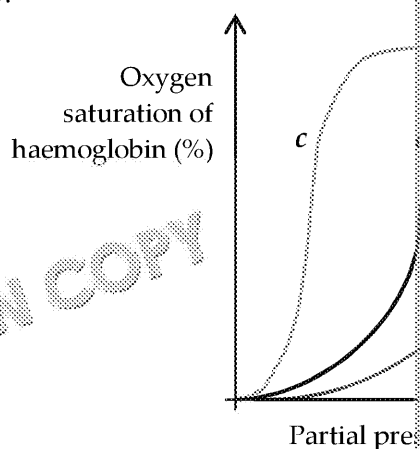


7. Haemoglobin carries oxygen around the body.  
 Copy and fill in the gaps in the process of oxygen dissociation below:

If carbon dioxide concentration increases in the blood, some diffuses \_\_\_\_\_ combines carbon dioxide with water to form \_\_\_\_\_. The \_\_\_\_\_ and hydrogen ions. The former diffuses out of the cell, while \_\_\_\_\_ balance the hydrogen ions – a process known as \_\_\_\_\_. The hydrolysis of haemoglobin, causing dissociation of oxygen and producing \_\_\_\_\_

8. The graph below shows three oxygen dissociation curves. Curve *a* is a normal curve. Curve *b* shows the dissociation curve in high  $\text{CO}_2$ . Curve *c* shows the dissociation curve of an animal adapted to high-altitude environments.

- a) Name the effect shown in curve *b*.  
 b) Why is this effect necessary in muscles during exercise?  
 c) What are the advantages of curve *c* in a high-altitude environment?



9. Cells are bathed in tissue fluid, which acts as an exchange medium between cells and capillaries.  
 a) Give an example of substances found in tissue fluid.  
 b) Describe how tissue fluid is formed and two ways it re-enters the capillaries.

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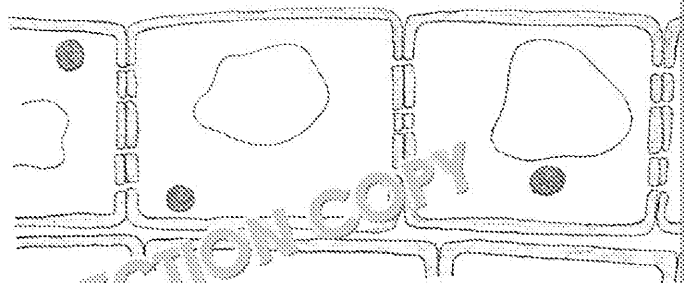
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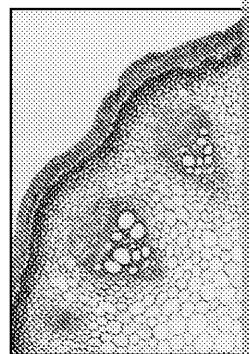
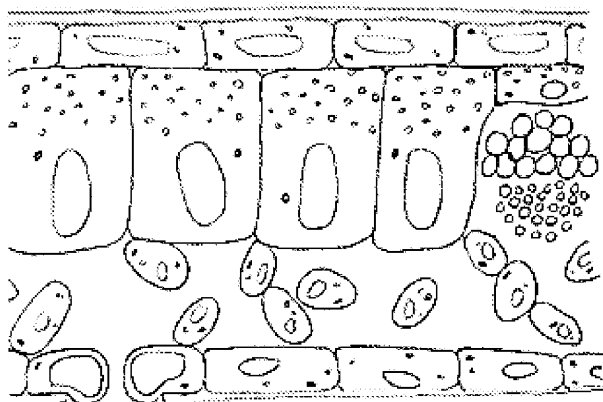
### 3. Adaptations for transport (in plants)

1. Describe the pathway of water from absorption by the roots to entering the leaves.
2. Draw and label the three pathways of water movement through plants.

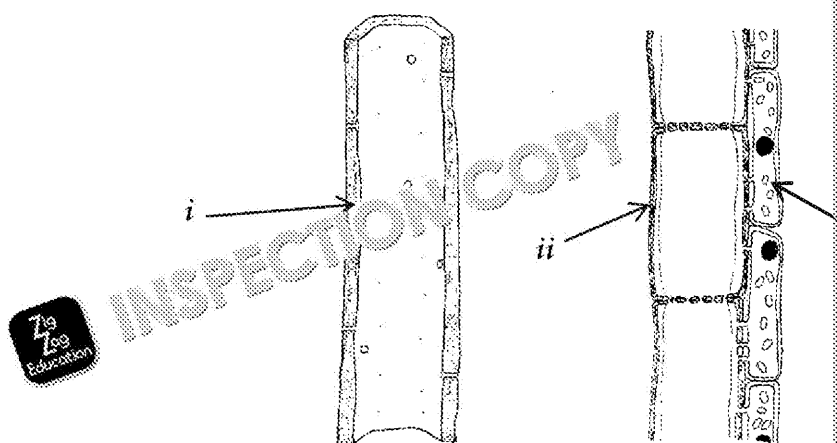


3. Label the xylem and phloem in the cross-section of the leaf in diagram A and photograph B.

Diagram A



4. a) Label cells *i* to *iii* in the diagram below.  
b) Give two differences between xylem and phloem.



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5. Copy and fill in the missing words in the process below:

The \_\_\_\_\_ theory of water movement in the xylem depends on \_\_\_\_\_ evaporates out of the \_\_\_\_\_ into an area of more \_\_\_\_\_ water molecules are held together by \_\_\_\_\_ bonds, a continual stream \_\_\_\_\_ upwards.

6. State two environmental factors that affect transpiration.

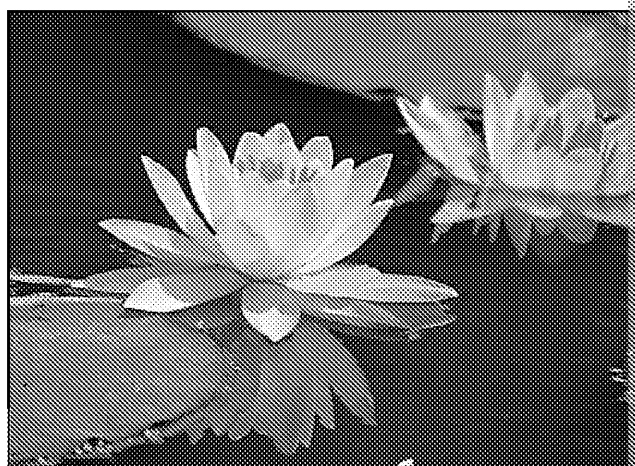
7. Two experiments were set up to compare the amount of water transpired under normal conditions and in bright light. The rate of bubble movement in a 5 mm capillary tube was measured every one minute and the results shown below

Calculate the ratio between the volume of water transpired in each condition.



Condition	Rate of bubble movement (cm/min)
Normal conditions	0.13
Bright light	0.32

8. a) What are xerophytes?  
b) List three adaptations of xerophytes, and describe how they reduce water loss by transpiration.
9. Describe three adaptations of hydrophytes for survival in an aquatic environment.



10. What is translocation?

11. Describe the process of mass flow in the phloem.

12. There are two experiments that provide evidence for mass flow.

- a) How are radioactive tracers used to show mass flow?  
b) Describe how a ringing experiment is performed and what would be the results.

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## 4. Adaptations for nutrition

1. Why do heterotrophic organisms rely on autotrophic organisms for the production of organic molecules?
2. Autotrophic organisms can be divided into \_\_\_\_\_ organisms, the energy source for \_\_\_\_\_, and \_\_\_\_\_ organisms, those that obtain energy from \_\_\_\_\_ reactions.
3. Compare saprotrophic and holozoic nutrition and state an example of each.
4. Describe how *Amoeba* obtain the nutrients they need.
5. Copy and complete the table below!

	Hydra	
Ingestion/Egestion		Distinct anterior and posterior digestive system.
Gut shape		
Differentiation of gut	None	

6. Below are microscope slides of cross sections of two distinct sections of the digestive system of a cow.
  - (a) Label A and B.
  - (b) Why might the intestine in slide B contain many mitochondria in its cells?
  - (c) Would you expect the gut of a cow to be longer, shorter or about the same length as the gut of a rabbit? Explain your reasoning.



7. List three different enzymes involved in digestion and name their substrates.
8. Ruminants make up a large proportion of all the animal protein eaten. Their diet consists mainly of plant material, yet they lack the enzyme to break down cellulose. Explain how a ruminant obtains its nutrients.

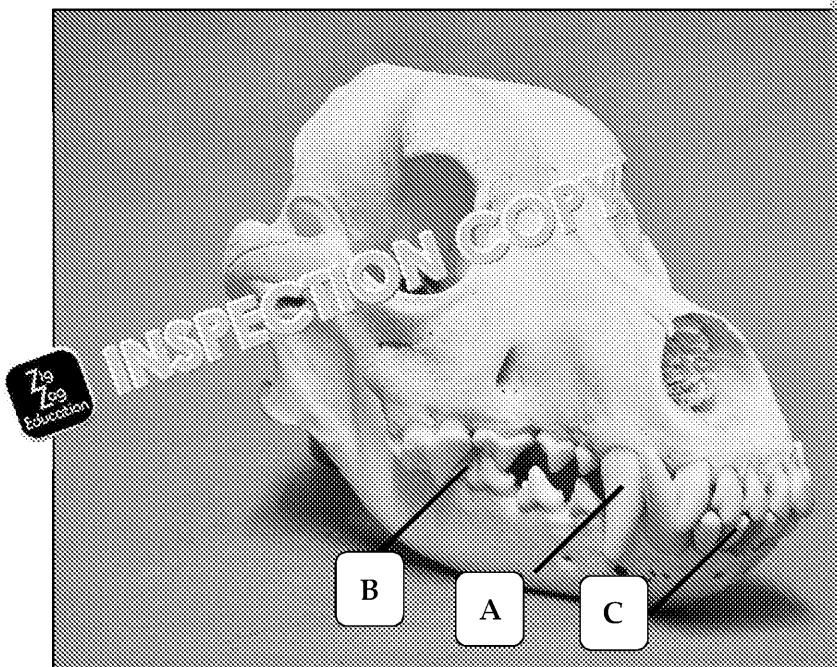
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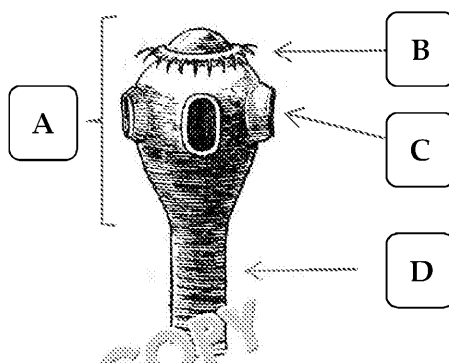




9. Below is a diagram of the skull of an animal.
- Does this skull belong to a carnivore, herbivore or omnivore?
  - What is tooth A used for?
  - Which label points to the carnassial tooth?
  - Describe two ways in which this dentition would differ to that of a



10. A tapeworm is an example of a parasite, an organism that lives on or inside another organism and obtains its own nourishment at the host's expense. Below is a diagram of a typical tapeworm.
- Label parts A–D.



- What is the binomial name for the pork tapeworm?
- What are the parasite's primary and secondary hosts?
- Describe two modifications of the tapeworm that enable it to be a parasite.

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## **Preview of Questions Ends Here**

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This is a limited inspection copy. Sample of questions ends here to avoid students previewing questions before they are set. See contents page for details of the rest of the resource.



# Mark Scheme

## AS Unit 1: Basic Biochemistry and Cell Organisation

### 1. Chemical compounds are joined together to form biological compounds

1. 'Inorganic' means does not contain carbon (1)  
'Ion' means they have gained or lost an electron (1)  
Found in the cytoplasm (1)

2. Their positive or negative charge (1) [Accept charge]

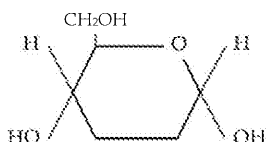
3.

Inorganic ion	Function
Iron ions	Structure of haemoglobin (1)
Phosphate ions (1)	Structure of DNA and ATP
Calcium ions (1)	Structural component of bones and teeth
Magnesium ions	Formation of <i>Chlorophyll</i> (1)

4. Hydrogen bonds form strong cohesion (1); allows transport of water through plants (1)  
Large latent heat of vaporisation (1) means effective cooling with little water loss for plants (1)

5. Water is a *solvent* (1); therefore, substances can dissolve and move freely for reactions (1)  
It is also a *metabolite* (1) [Accept reactant] in hydrolysis and *condensation* (1) reactions (1)  
forming and breaking up *polymers* (1).

6. a)  $\beta$ -glucose [1 mark for  $\beta$ , 1 mark for glucose]  
b) In the isomer, the OH and H molecules are swapped on the first carbon (1)



[1 mark for drawing, ensure correct orientation of OH groups]

7. A *condensation* (1) reaction causes two glucose monosaccharides to form a  $\alpha$ -1,4-glycosidic bond between them, creating a disaccharide called *maltose* (1). If more monosaccharides are joined together, a *polysaccharide* (1) is formed.

8.

Monosaccharides	Disaccharides
Glucose + Fructose	Sucrose
Glucose + Galactose	Lactose

9. [Any two from:]  
Glycogen is used for storage; cellulose is used structurally (1)  
Glycogen is formed from  $\alpha$ -glucose; cellulose is formed from  $\beta$ -glucose (1)  
Glycogen has a heavily branched structure; cellulose forms a mesh / long fibres (1)

10. Iodine [or] potassium iodide (1)

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11. a) Benedict's solution (1)
- b) Enzymes have broken down / hydrolysed starch into glucose (1); therefore detecting glucose monosaccharides (1)
- c) Reducing sugars (1)
- d) Using reagent strips (1) means can compare results to a colour-coded chart (1)

## 1. Chemical compounds are joined together to form biological compounds

1. Triglyceride (1)
2. Condensation reaction (1) which produces an ester bond (1)
3. One of the R-groups / fatty acids of a triglyceride is replaced with / substituted by a phosphate group (1)
4. Phospholipid A is a saturated phospholipid with no carbon-carbon double bonds (1)  
Polyunsaturated phospholipid with multiple carbon-carbon double bonds (1)
5. The head of the phospholipid is polar / negatively charged (1)  
The head of the phospholipid is hydrophilic (1)  
The tail of the phospholipid is non-polar / not charged (1)  
The tail of the phospholipid is hydrophobic (1)  
Therefore, they form a bilayer in water (1)
6. a) Emulsion test (1)  
b) Add ethanol and mix/shake (1) [ $\frac{1}{2}$  for only ethanol]  
In a test tube, mix the above with water and shake gently (1)  
Test tube containing lipids turns cloudy (1)
7. The ratio of saturated and unsaturated fats in our diet has implications for health. A diet high in saturated fats, *low-density lipoproteins* (1) build up causing fatty material deposited in the *coronary arteries* (1). It impedes blood flow and restricts oxygen, a disease known as *atherosclerosis* (1). If the diet contains more unsaturated fats, *high-density lipoproteins* (1), which carry harmful fats to the liver for processing (1).
8. a) Amino group (1)  
b) R group (1)  
c) Acid group / Carboxyl group (1)
9. A *dipeptide* (1) is formed from two amino acids. A *peptide* (1) bond is created through a *condensation* (1) reaction.
10. a) Fibrous proteins are long and insoluble (1) and have structural roles (1); Globular proteins are compact and soluble (1) and have metabolic roles (1)  
b) [Fibrous – any two from:]
  - Collagen
  - Keratin
  - Elastin
 [Globular – any two from:]
  - Haemoglobin
  - [Accept any named enzyme, e.g.] Amylase
  - [Accept any named hormone, e.g.] Insulin

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11. a) Hydrogen (1)  
b) Disulphide bonds (1), Ionic bonds (1), Hydrogen bonds (1) [Remove 1 mark]  
c) Includes other polypeptide chains (1) and prosthetic groups (1)  
d) Any appropriate answer, e.g. Haemoglobin/collagen (1)

12. Biuret test (1)

Biuret solution / copper (II) sulfate solution added one drop at a time (1)

Turns mauve/lilac in presence of proteins (1)

## 2. Cell Structure and Organisation

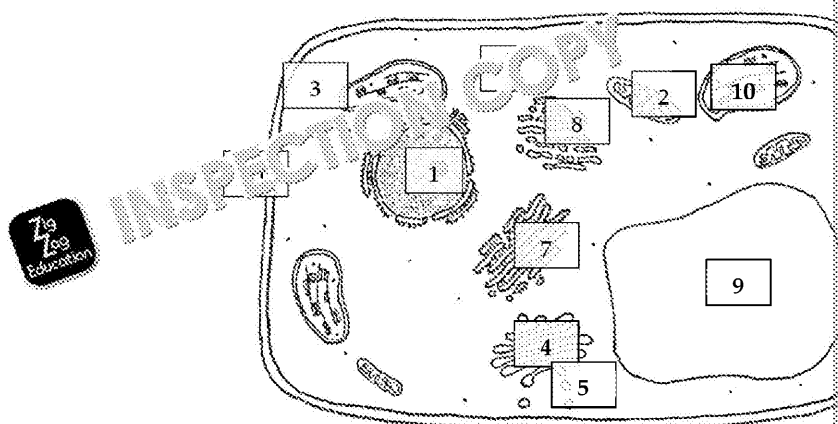
1. [Award half mark for each correct answer]

Organelles → Cells → Tissues → Organs →

2. a) [1 mark for each correct grouping of organelle, feature and function:]

1. Nucleus	→	b) Double membrane with pores	→	x)
2. Mitochondria	→	f) Folded inner cristae	→	ix)
3. Cell-surface membrane	→	k) Phospholipid bilayer	→	xi)
4. Golgi apparatus	→	a) Flattened sacs that bud into vesicles	→	iii)
5. Lysosomes	→	c) Small vesicles	→	vi)
6. Ribosomes	→	i) Small proteins	→	ii)
7. Rough ER	→	g) Thin tubes coated in ribosomes	→	i)
8. Smooth ER	→	e) Thin tubes with many vesicles	→	v)
9. Vacuole	→	h) Large sac of liquid	→	vii)
10. Chloroplasts	→	d) Double membrane, free DNA	→	iv)
11. Cell wall	→	j) Rigid cellulose/chitin polymers	→	viii)

- b) [1 mark for correct placement of following numbers:]



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3. [1 mark for correct measurement of image diameter, accept any reasonable answer]

Diameter of nucleus is 20 mm

[1 mark for rearrangement and substitution into equation:]

$$\text{Object size} = \frac{\text{Image size}}{\text{Magnification}}$$

$$\text{Object diameter} = \frac{20}{1000}$$

[1 mark for diameter:]

Object diameter = 0.02 mm

[1 mark for equation for circumference:]

$$\text{Circumference} = \pi \times d$$

[1 mark for calculation of circumference:]

$$\text{Circumference} = \pi \times 0.02$$

$$\text{Circumference} = 0.0628318 \text{ mm}$$

[1 mark for answer in standard form to 3 sf:]

$$\text{Circumference} = 6.28 \times 10^{-2} \text{ mm}$$

4. Many cells are colourless, so detail cannot be seen unless stained (1)

5. a) 1 micrometer unit = 2.7 graticule units (1)

$$\text{Therefore, 1 micrometer unit} = \frac{10}{2.7} = 3.7 \mu\text{m} (1)$$

- b) [1 mark for calculation:] Length =  $20830 \times 3.7$   
= 77 071  $\mu\text{m}$

[1 mark for correct conversion to 3 sf:]

$$= 77.1 \text{ mm}$$

6. a) *a* is the nucleus (1), *b* is vesicles (1), *c* is the Golgi apparatus (1)  
b) Rough endoplasmic reticulum (1) [Do not accept rough ER or r ER for full mark]

7. a) Plasmid (1)  
b) Single circular DNA [Accept DNA] (1)  
c) Capsule (1)  
d) Cell wall (1)  
e) Flagellum [Accept flagella even if plural] (1)

8. Any two from the following [1 mark each]

Prokaryotic cells have a single, circular DNA; eukaryotic cells have multiple  
Prokaryotic cells have no membrane-bound organelles; eukaryotic cells have  
[Accept eukaryotic can have nucleus, mitochondria and chloroplasts; prokaryotic  
Prokaryotic cell has a capsule; eukaryotic cells do not (1)  
Prokaryotes have 70s ribosomes while eukaryotes have 80s ribosomes (1)  
Prokaryotic cells are much smaller than eukaryotic cells (1)

9. a) RNA (1) [Accept genetic material or DNA as some viruses contain DNA]  
b) Capsid / protein coat (1)  
c) Lipid membrane / phospholipid bilayer (1)

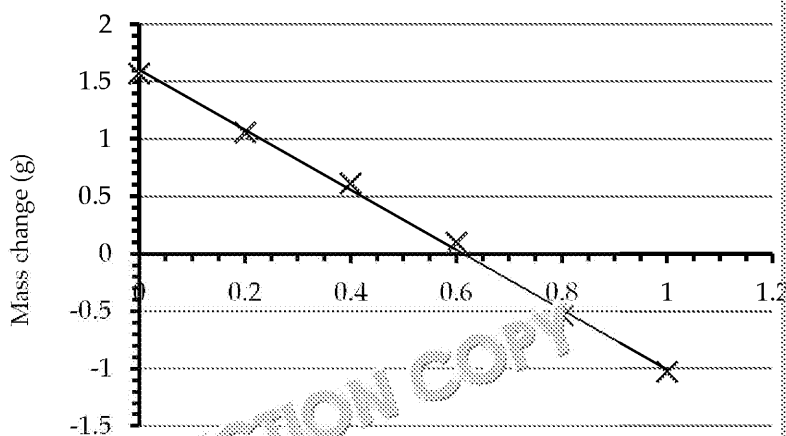
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### 3. Cell Membranes and Transport

1. a) i: Glycoprotein (1)  
ii: Cholesterol (1)  
iii: Glycolipid (1)  
b) Receptors for hormones (1) and cell signalling (1)  
c) Adds strength/rigidity to membrane by holding surrounding molecules
2. Fluid part means membrane is not fixed / is flexible (1), mosaic part means it is made of small molecules (1)
3. a) Increasing temperature increases kinetic energy (1); therefore, the components move more (1) and can cause the membrane to 'leak' / allow substances to enter (1)  
b) pH (1) or alcohol/solvents (1)
4. Diffusion is the net movement of molecules/ions (1) from a region of higher concentration (1) to a region of lower concentration (1)
5. Small (1), non-charged/non-polar (1) lipid soluble (1)
6. Osmosis is the movement of water molecules (1) across a partially permeable membrane (1) from high water potential to low water potential (1)
7. Unlike simple diffusion, facilitated diffusion requires carrier/channel proteins (1) as molecules are too big to pass through the membrane (1)
8. a) Line c shows active transport (1)  
b) Respiration produces ATP (1), ATP is required by the membrane proteins (1)  
c) Potassium cyanide (1)
9. a) To maintain a constant temperature and to increase rate of movement / diffusion (1)  
b) [ $\frac{1}{2}$  mark for each point, 1 mark for line of best fit:]



- c) Sucrose concentration is approximately  $0.62 \text{ mol dm}^{-3}$  (1). There should be no net movement of water (1); therefore, the water potential of the solution and potato tuber are equal (1).

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d) [1 mark for equation:]

$$\% \text{ error} = \frac{\text{maximum error}}{\text{measured value recorded}} \times 100$$

[½ mark for each correct calculation of % error for final mass:]

$$0.0 \text{ conc.: \% error mass} = \frac{0.01}{4} \times 100 = 0.25$$

$$0.2 \text{ conc.: \% error mass} = \frac{0.01}{3.43} \times 100 = 0.29155$$

$$0.4 \text{ conc.: \% error mass} = \frac{0.01}{3.18} \times 100 = 0.3$$

$$0.6 \text{ conc.: \% error mass} = \frac{0.01}{2.24} \times 100 = 0.4464$$

$$0.8 \text{ conc.: \% error mass} = \frac{0.01}{2.25} \times 100 = 0.4$$

$$1 \text{ conc.: \% error mass} = \frac{0.01}{1.26} \times 100 = 0.8$$

[1 mark for calculation of total error to 3 sf:]

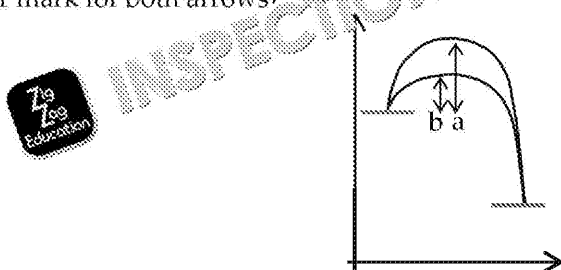
$$\text{Total \% error} = 0.25 + 0.29155 + 0.3 + 0.4464 + 0.4 + 0.8$$

$$\text{Total \% error} = 2.57\% \text{ [Accept between 2.54 and 2.57]}$$

10. a) Endocytosis and exocytosis can be used for bulk transport (1)  
b) ATP is used to make membranes 'pinch off' to form vesicles (1) and to move them (1)
11. a) As temperature increases, the permeability of the membrane increases (1)  
red pigment found extracellularly in the tube increased (1)  
b) Time or temperature (1)

#### 4. Biological reactions are regulated by enzymes

1. Tertiary structure provides a 3D active site (1)  
Spherical shape allows easy transport out of cells (1)
2. It would affect the functioning of an enzyme (1) because different amino acid sequence would change the structure of the enzyme to change (1)
3. Increase the chemical reaction (1) without being consumed / permanently changed (1)
4. a) The minimum energy required to trigger a chemical reaction (1)  
b) [1 mark for both arrows]



c) Line b (1)

5. Induced fit model (1); Induced fit model states the active site is flexible / can change shape (1)

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6. a) Extracellular (1)  
 b) Intracellular (1) breaks down hydrogen peroxide into water and oxygen product of many metabolic reactions (1)  
 c) [Extracellular:] Digestive enzymes, such as amylase/lipase/protease, in the example including:] lysozymes, DNA polymerase, DNA helicase [etc.] (1)
7. a: Not enough substrate to fill all available enzyme active sites [and] substrate (1)  
 b: Most active sites are becoming occupied, reducing excess substrate (1)  
 c: All active sites filled; therefore, excess substrate cannot increase rate of reaction (1)
8. a: Increased kinetic energy increases encounters between enzyme and substrate (1)  
 b: Enzymes start to denature (1); therefore, active site changes and substrate cannot bind (1)  
 c: Enzymes are fully denatured (1)
9. a) [1 mark for correct calculation of pH]  
 $pH = -\log_{10}[H^{+}]$   
 [1 mark for correct answer to 2 dp:]  
 $pH = 6.03$  (1)
- b)  $3.60 \text{ g min}^{-1}$  (1)
- c) [1 mark for equation:]  

$$\text{Rate} = \frac{\text{change in } y}{\text{change in } x}$$
  
 [1 mark for correct answer – example shown below – and 1 mark for correct units]  

$$\text{Rate} = \frac{0.6 - 0.2}{9 - 3}$$
  

$$\text{Rate} = 0.0\dot{6} \text{ g sec}^{-1}$$
  

$$\text{Rate} = 4.00 \text{ g min}^{-1} \text{ (accept } 4\text{--}6 \text{ g min}^{-1}\text{)}$$
10. Buffers help maintain a constant pH (1) by absorbing excess hydrogen or hydroxide ions (1)  
 [Accept]:  $H^{+}$   $OH^{-}$  ions
11. Competitive inhibitors bind to active site and block [and] non-competitive inhibitors bind to a different site and change the shape of active site (1)
12. (a) Enzymes are bound/fixed/trapped (1) within an inert matrix (1) such as microfibrils (1)  
 (b) [Any three from:]  
 • Increased stability and function over a wide range of temperature and pH (1)  
 • Products are not covalently bound with the enzyme (1)  
 • Enzymes are easily recovered for reuse (1)  
 • Enzymes can be easily added or removed, allowing more control over the reaction (1)  
 • Sequence of columns containing different enzymes with different substrates (1)  
 • Experiment (1)
- (c) [Any one from:]  
 • Used in biosensors (1)  
 • Manufacture of high-fructose corn syrup (1)  
 • Manufacture of lactose-free milk (1)

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## 5. Nucleic acids and their functions (i)

1.
  - a) A = nucleotide/mononucleotide (1)
  - b) B = pentose/deoxyribose (1)  
C = phosphate group (1)  
D = organic/nitrogenous base (1) [Do not accept base pair]
  - c) E = phosphodiester bond (1)
  - d) Condensation reaction (1)
2.
  - a) TTCGCAGA [1 mark for A-T pairing; 1 mark for C-G pairing]
  - b) Hydrogen bond (1)
  - c) [1 mark for two bonds between A-T pairs; 1 mark for three bonds between C-G pairs]
  - d) Double helix (1)
3. Provides genetic information / instructions for:  
Cell division (1)  
Protein synthesis (1)
4. RNA is a single strand; DNA is a long double strand (1)  
RNA contains ribose; DNA contains deoxyribose (1)  
RNA contains the base uracil; DNA contains the base thymine (1)
5. The enzyme *DNA helicase* unwinds the DNA strand and breaks the *hydrogen bonds*.  
*Free nucleotides* are attracted to and bond with their exposed complementary bases.  
*Phosphodiester bonds* are formed when the enzyme *DNA polymerase* initiates a new strand of deoxyribose phosphate backbone together.  
Half of the original DNA is now part of the new DNA fragment. This process is called semi-conservative replication.  
[1 mark for each]
6. DNA is made from new molecules (1) and the original DNA remains intact (1)
7.
  - a)  $N^{14}$  parent and offspring all have light DNA (1)  
 $N^{15}$  parent has heaviest DNA (1)  
1<sup>st</sup> generation has an intermediate band because DNA in the 1<sup>st</sup> generation has one heavy  $N^{15}$  bases (1) and one new strand of light  $N^{14}$  bases (1)  
2<sup>nd</sup> generation has lighter and intermediate band because the original strands have one  $N^{15}$  bases (1) and one new  $N^{14}$  bases (1); however, the new  $N^{14}$  strands from the first generation are entirely light DNA (1)
  - b) The conservative model would have retained the original  $N^{15}$  DNA with one new  $N^{14}$  strand.  
DNA would be all light  $N^{14}$  DNA; therefore, no intermediate band would be present.
8. Adenosine triphosphate (1)
9.
  - a) Adenosine (1)
  - b) Ribose/pentose (1)
  - c) Three inorganic phosphates (1)
10.
  - a)
  - b) Condensation reactions synthesize ADP and the inorganic phosphates in the process.  
ATP synthase removes inorganic phosphates from ATP to release energy (1)

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11. [Any four from:]

- Active transport to secrete substances (1)
- Active transport to extract substances against a concentration gradient (1)
- Muscle contraction (1)
- Provide activation energy in enzyme reactions (1)
- Nerve cells / neurones (1)
- Phosphorylating compounds / providing phosphorus for other reactions (1)

## 5. Nucleic acids and their functions (ii)

- The triplet code of DNA (1) codes for specific *amino acids* (1). The code is *united* overlapping and *degenerate* (1), which means that *different* amino acids have more than one code (1). [1 mark for each]
- Exons code for amino acids (1)
  - Introns are non-coding (1), consist of multiple base repeats (1), can act as structural genes within a chromosome (1) and exons within genes (1)
- mRNA forms a long helix; tRNA forms a short clover-leaf shape (1)  
mRNA just contains uracil, adenine, guanine, cytosine / the four bases; tRNA contains an amino acid (1)
- In *transcription* (1) the DNA strands are separated and free nucleotides are attached by *polymerase* (1) to form *messenger* (1) RNA.  
mRNA is formed directly in *prokaryotic* (1) cells, while the immature form *pre-mRNA* is formed in *eukaryotic* cells. This molecule is then *spliced* (1) to remove introns.  
The mRNA leaves the *nucleus* (1).
- Ribosomes hold the mRNA and tRNA in place (1)
  - tRNA bonds to the mRNA with its complementary anticodon (1) and holds the amino acid (1)
  - ATP provides energy to create a peptide bond between the amino acids (1)
  - Amino acids join together to form a polypeptide/protein (1)
- mRNA: GCUUAG (1)  
Amino acids: Alanine, (stop) (1)  
*i* = Transcription (1)  
*ii* = Translation (1)
- This hypothesis suggests that each gene is a sequence of DNA bases (1) that code for a specific protein (1).
- Golgi
  - [Any two from:]
    - Addition of a carbohydrate to form a glycoprotein (1)
    - Addition of a lipid to form a lipoprotein (1)
    - Addition of a phosphate (1) to form a phosphoprotein (1)
    - Any other acceptable example (1)
- The detergent destroys phospholipid cell membranes to release the DNA (1).  
The detergent also coagulates DNA to make it more visible (1).
  - Cellular enzymes are denatured at 60 °C (1). Any enzymes released during the process will be prevented from destroying the DNA (1).
  - Add drops of trypsin (1) to the boiling tube to break down the histone protein (1). Cold ethanol (1) is slowly added so it sits on top of the extract (1). A white precipitate will be seen as a white precipitate (1) between the filtrate and ethanol layer (1).

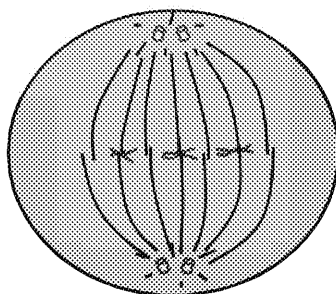
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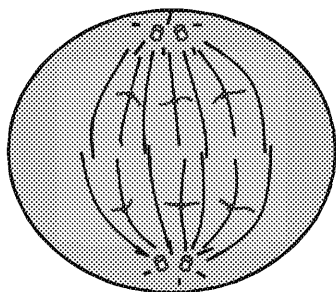


## 6. Genetic information is copied and passed onto daughter cells

1.
  - a) Mitosis (1)
  - b) [Any three of:]
    - Organism growth
    - Tissue repair
    - Cell turnover/replacement
    - Asexual reproduction
2. *Interphase* involves DNA replication and the organelles duplicate.  
 In *prophase* the chromosomes become visible and the *nuclear envelope* disintegrates.  
 During *metaphase* the chromosomes line up on the cell's equator.  
 In *anaphase* the *chromatids* are split and pulled to opposite poles of the cell.  
 During *telophase* the nuclear envelope reforms around the separated chromatids.  
 Cytokinesis completes division where *cytoplasm* divides. [Do not accept cell wall]  
 [1 mark for each]  
 [Drawing for metaphase – 1 mark for including spindle fibres and centrioles, 1 mark for aligned chromosomes]



[Drawing for anaphase – 1 mark for including spindle fibres and centrioles, 1 mark for chromatids pulled away from centre:]



3. Spindle fibres attach to the centromeres (1), separate the chromatids (1) and divide the cell (1)
4. Cancer is uncontrolled rapid division of cell (1) when the genes that regulate cell division are faulty (1)  
 Cancer drugs interfere with the cell cycle to prevent division (1)
5.
  - a) The diploid parent cell divides and homologous chromosomes split (1), the cells divide again (1) forming four haploid gametes (1)
  - b) Mitosis produces two identical diploid daughter cells (1) with the same genetic information as the parent cell (1)
  - c) Diagram A is Prophase I, diagram B is Anaphase II [1 mark for anaphase]
6. Crossing over / Recombination between homologous chromosomes (1) where they exchange DNA and recombine (1)  
 Independent segregation/assortment of homologous chromosomes (1) where they separate independently (1)  
 Random fertilisation (1)



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