



Course Companion

for Pearson Level 3 AAQ BTEC National
in Applied Science (Extended Certificate)

Unit 1 Principles and Applications of Biology



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Teacher's Introduction

Thank you for choosing this Course Companion which has been written specifically for the Level 3 BTEC National Extended Certificate in Applied Science (AAQ) qualification (first teaching from September 2025). The theory notes and recap questions cover the essential knowledge and understanding prescribed in the BTEC Unit 1 specification.

About Unit 1: Principles and Applications of Biology

Unit 1 ^(60 GLH) is assessed through a 1-hour (50-mark) written examination, which is set and marked by Pearson. There are two opportunities for assessment each year – in January and in May/June.

Unit 1 is a mandatory unit and will be assessed in a variety of styles, such as: multiple-choice questions, calculations, and short-answer and extended-response questions.

Each of the three learning aims (A–C) is given its own section in this resource.

The sections are as follows:

- A. Structure and function of cells and tissues
- B. Structure and function of biological molecules
- C. Cellular transport and enzyme activity

Remember!

Always check the exam board website for new information, including changes to the specification and sample assessment material.

Within each section there are student notes covering the specification content and structure. These aim to break down the content into manageable chunks and are further supported with diagrams, images and examples of how to structure answers.

Questions are interspersed throughout the guide to test and develop understanding. Suggested answers are included at the back of this resource.

NB the intention of the suggested answers is to save the teacher time, rather than to offer a comprehensive set of definitive answers. In some cases, there are equally valid alternative answers to those that have been given.

Best wishes and good luck to you and your students!

July 2025



A web page containing all the links listed in this resource is conveniently provided on ZigZag Education's website at **zzed.uk/12833**

You may find this helpful for accessing the websites rather than typing in each URL.

A: Structure and function of cells

A1 Structure and function of cells and tissues



Key points covered

- Ultrastructure and function of organelles in eukaryotic and prokaryotic cells
- Use of light microscope
- Similarities and differences between animal cells
- Calculating magnification

A cell is the basic building block of all living organisms. Some organisms are unicellular (made of one cell), while others are multicellular (made of many cells). All cells have some core features in common, such as a cell membrane and cytoplasm, but there is great variation between **prokaryotic** cells (without a true nucleus) and **eukaryotic** cells (with a true nucleus).

prokaryotic
nucleus

eukaryotic
nucleus
plant cell

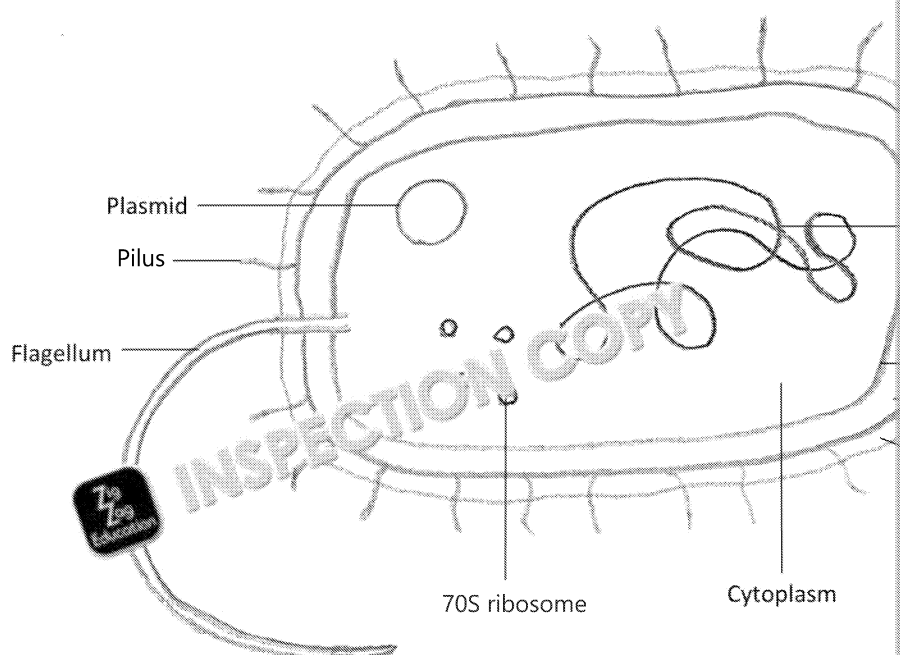
Test your knowledge

- Name the two categories of cells and give an example of each.
- State two features of a cell which are common to all cells.

Prokaryotic and eukaryotic cells

Prokaryotic cells

Prokaryotic cells include all Gram-positive and Gram-negative bacterial cells. These cells have several features that allow them to carry out a wide range of life processes that enable them to survive. These features are summarised in the table on the following page.

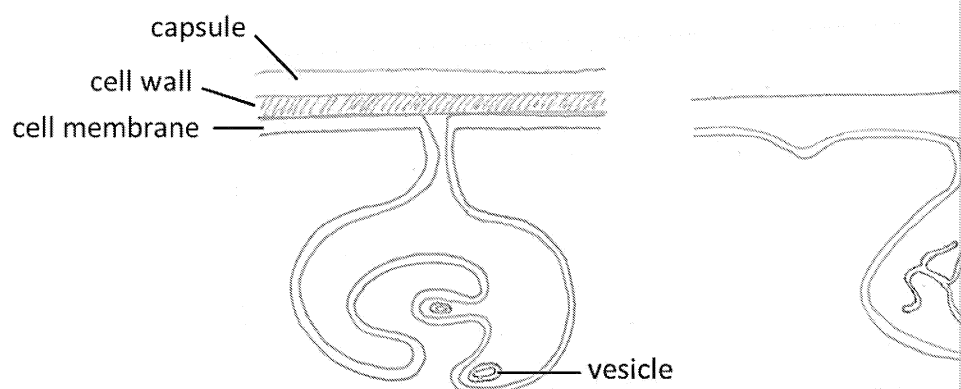


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Cell feature	Description
Nucleoid	A region of the cell which stores the genetic information, not distinct from the rest of the cytoplasm.
Plasmid	A small, circular piece of DNA which contains a few chromosomes in the nucleoid.
70S ribosomes	Smaller than the 80S ribosomes found in eukaryotic cells, used for protein synthesis.
Capsule	A jelly or slime-like layer that covers the whole of the bacteria and offers protection.
Cell wall	Made of peptidoglycan and provides structural support.
Plasma membrane	The cell membrane which forms the boundary of the cell, controlling the entry of substances.
Cytoplasm	The internal component of the cell, made largely of water and organic molecules (called cytosol).
Mesosomes	Invaginations (infolds) in the plasma membrane which are created by the chemical fixation process used when preparing cells for electron microscopy (see diagram) and some Gram-negative) for microscopy (see diagram).
Flagella (s. flagellum)	Whip-like structures used for motility. They do not occur in eukaryotic cells, and are powered by chemiosmosis.
Pili (s. pilus)	Short hairs involved in attaching the bacteria to the surface. Sex pili are involved in conjugation reproduction in some bacteria.

Mesosomes are chemical artefacts created when chemicals are applied to the cells under the electron microscope. They are convoluted membrane deformations that appear as vesicles of varying size and shape.



Test your knowledge

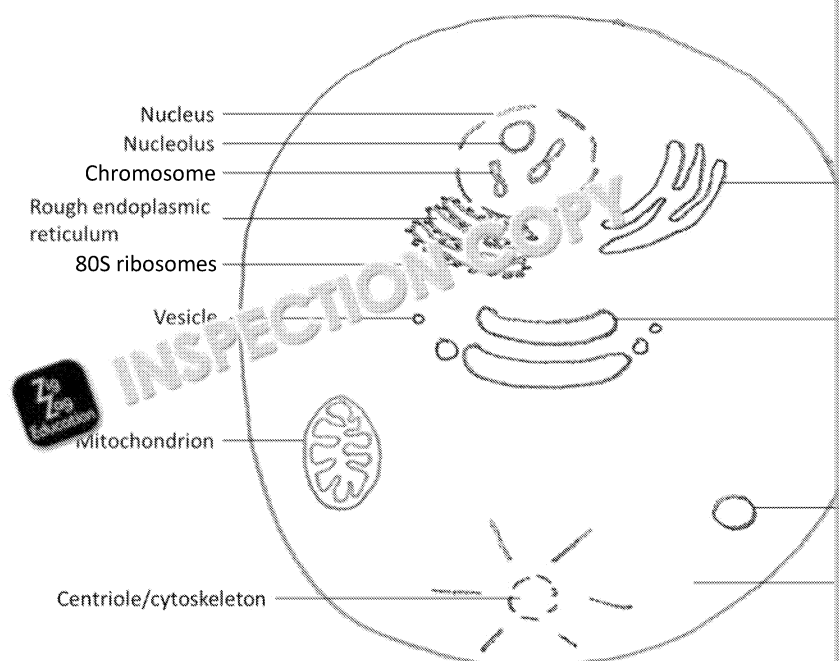
- State the function of a flagellum.
- State the difference between the prokaryotic and eukaryotic ribosomes.
- Describe how the eukaryotic nucleus and prokaryotic nucleoid differ.

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Eukaryotic cells

Eukaryotic cells have complex internal cell structures. The space is divided into a number of compartments which carry out specific functions. This allows specialised and **optimised** specific chemical reactions; it groups linked processes together in the same local space and it keeps molecules separate from other parts of the cell which might destroy, or be destroyed by, them. The key features of eukaryotic cells are listed in the table below.



Cell feature	Description
Plasma membrane	The cell membrane which forms the boundary of the cell and controls the entry of substances.
Cytoplasm	The main component of the cell, made largely of water and organic molecules (called cytosol).
Nucleus (p. nuclei)	Contains the genetic material for the cell.
Nucleolus	An area within the nucleus which produces ribosomes.
80S ribosomes	Larger than the 70S ribosomes found in prokaryotic cells, used for protein synthesis.
Rough endoplasmic reticulum	Flattened membranes with ribosomes attached to them, involved in protein synthesis.
Golgi apparatus	Modifies proteins after production and packages them for transport.
Smooth endoplasmic reticulum	Flattened membranes responsible for lipid and carbohydrate metabolism and storage.
Vesicles	Small, round sacs which store and transport substances within the cell.
Lysosomes	Small, round sacs which store and transport substances for degradation and secretion from the cell.
Mitochondrion (s. mitochondrion)	Site of the latter stages of aerobic cellular respiration.
Centrioles	Part of the cytoskeleton involved in cell division (microtubule organising centre).
Cilia	Short hair-like projections from the surface of the cell. They may be sensory, such as in the nasal cavity, or mobile and regularly waft, such as in the trachea to move mucus up to the throat.

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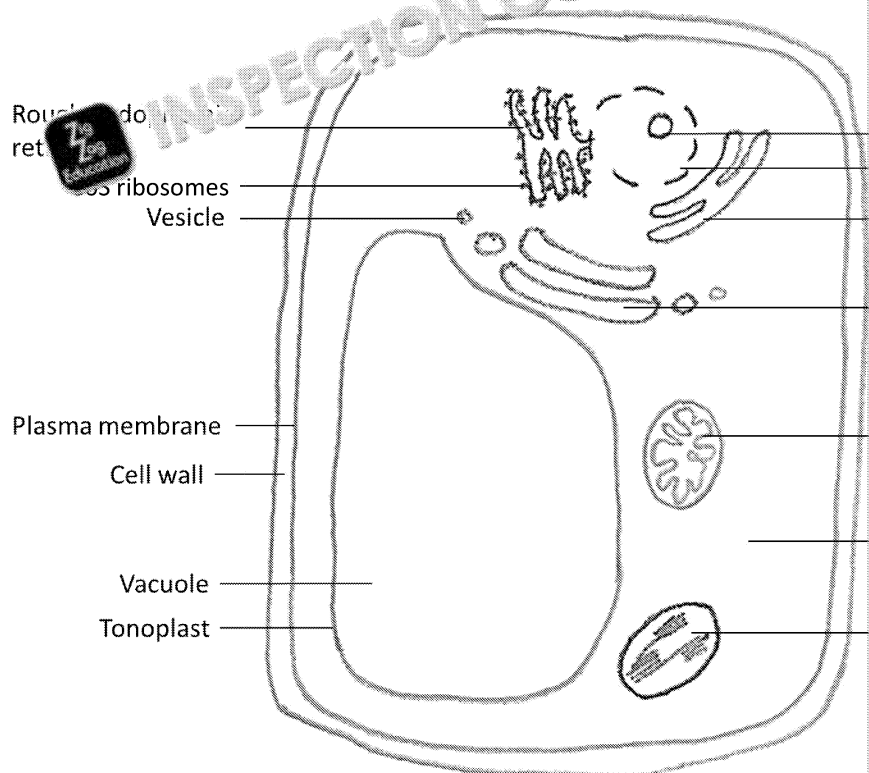


Test your knowledge

- State the function of centrioles.
- Describe how the nucleolus, the rough endoplasmic reticulum and the Golgi apparatus are involved in protein synthesis.
- Describe the differences between vesicles and lysosomes.

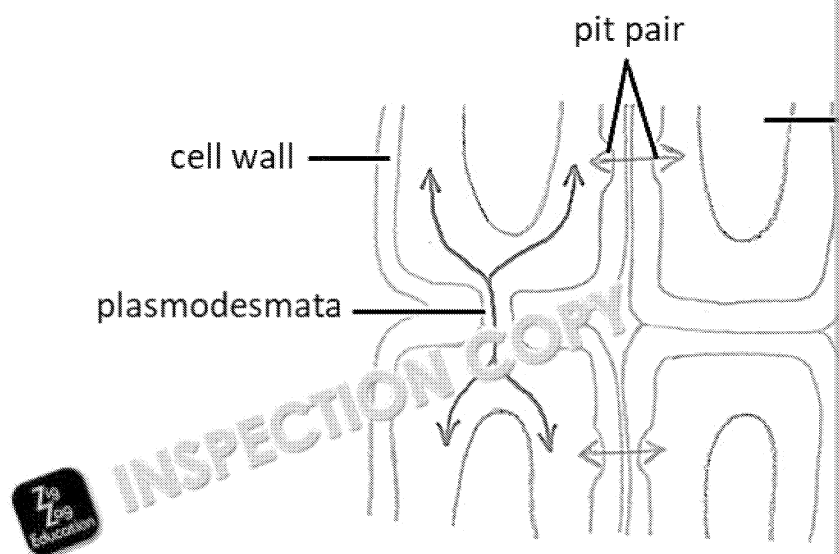
Plant-specific organelles

Plants are eukaryotic cells but because they are **producers** – they use photosynthesis to produce their own food in the form of glucose – they require some additional cell components to be able to carry out these functions. These plant-specific **organelles** are summarised in the table below.



Cell feature	Description
Cell wall	Made of cellulose and provides structural support and protection.
Chloroplasts	Contain chlorophyll and are responsible for photosynthesis.
Vacuole	These large, permanent features of plant cells store water and enable the plant to maintain turgor: strength created by water pressure.
Tonoplast	The membrane that surrounds the vacuole and controls the movement of water between the cytoplasm and the vacuole.
Amplified	Found in roots and storage tissues (e.g. potatoes), they are responsible for the synthesis and storage of starch.
Plasmodesmata	Channels that connect plant cells to each other to allow for communication between cells.
Pits	Thinner sections of the cell wall which form pits or channels for the exchange of water between cells. They are paired between adjacent cells.

Plasmodesmata and pits both facilitate the movement of water between cells. Plasmodesmata directly connect one cell to another, thus also facilitating the movement of other nutrients and chemicals, as shown in the arrows on the diagram below.



Test your knowledge

- Describe how plasmodesmata and pits allow communication between plant cells.
- Describe how the vacuole and the tonoplast are linked.

Similarities and differences between plant cells and animal cells

Plant cells and animal cells are both eukaryotic: they have a membrane-bound nucleus. However, plants are producers and animals are consumers, so they have different supporting processes that are reflected in the organelles that the cells contain. The following table lists the features that are found in plant cells and which are found in animal cells.

Organelle	Found in plant cells?	Found in animal cells?
Cell wall	✓	
Large, permanent vacuole	✓	
80S ribosomes	✓	
Mitochondria	✓	
Chloroplasts	✓	
Endoplasmic reticulum	✓	
Amyloplasts	✓	
Plasmodesmata	✓	
Golgi apparatus	✓	

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Test your knowledge

- Explain why plant cells require a cell wall but animal cells do not, and suggest a function for the cell wall.
- Explain why both animal and plant cells require mitochondria, but only some plant cells have large, permanent vacuoles.

Bacteria and antibiotics

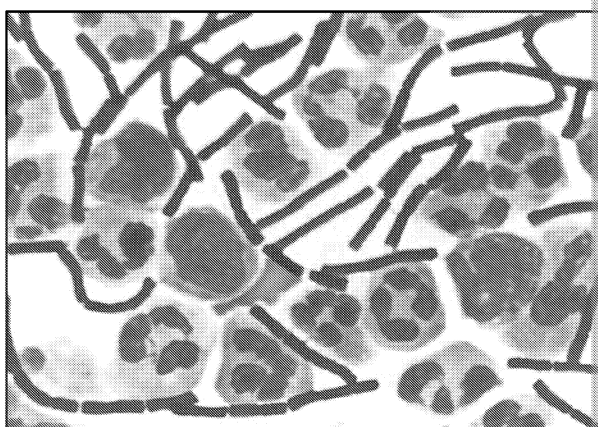
There are a great number of bacteria, many of which are pathogenic to other organisms. Bacteria are extremely small, of less than $10\text{ }\mu\text{m}$ in length, and therefore must be viewed using a microscope.

To aid in identification, bacterial samples are stained, and the chemical composition of their cell wall causes differences in the uptake of **stain** which leads to their classification as either Gram-positive or Gram-negative. The capital G is important here as this staining technique was developed by Christian Gram in the 1880s.

The stain used is crystal violet, which is fixed with iodine and then washed off with alcohol.

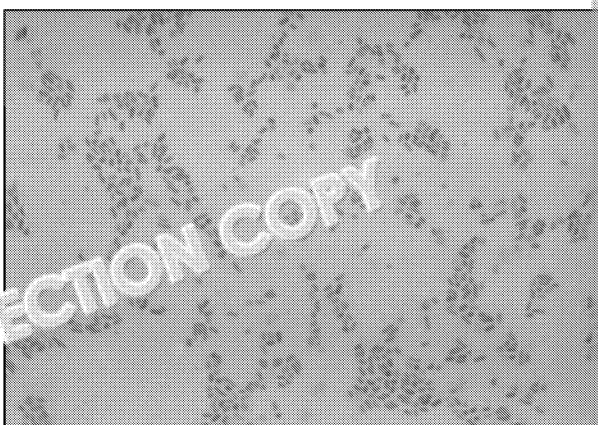
Gram-positive bacteria have almost no lipid within their cell wall, retain the crystal violet stain, and appear purple or blue under a light microscope. They are susceptible to the antibiotic penicillin. The formation of the cell wall, and includes the peptidoglycan layer. Examples include *Bacillus*, *Clostridium*, *Staphylococcus*.

In this photomicrograph, anthrax bacilli show as the dark purple rods.



Gram-negative bacteria have large amounts of lipid in their cell wall, which makes the crystal violet stain washes off. A different stain, called a counterstain, is applied: safranin. Gram-negative bacteria appear red. They are resistant to penicillin because of their outer membrane, susceptible to the enzyme lysozyme, and include *Salmonella*, *Escherichia* and *Azotobacter*.

In this photomicrograph, *Bacteroides fragilis* ss. *vulgatus* show as pink rods/dots.



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Test your knowledge

- Are *Salmonella* bacteria resistant or susceptible to penicillin?
- Will *Clostridium* bacteria stain with crystal violet or safranin dye?

Microscopy

Cells are usually too small to observe without a microscope, and some cell organelles require **electron microscopes** to be viewed at all.

The light microscope

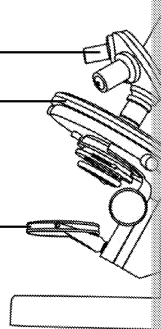
There are several key parts to a light microscope:

- Eyepiece lens – usually has a $\times 10$ magnification and is the part of the microscope that is looked through.
- Objective lenses – usually three lenses of $\times 4$, $\times 10$ and $\times 40$ magnification.
- Stage – where the specimen is placed.
- Focusing wheels – a coarse and fine wheel that allow the focus to be adjusted in larger and smaller increments, respectively.
- Light/mirror – which shines light through the specimen, objective lenses and eyepiece.

Objective lenses

Stage

Light/mirror

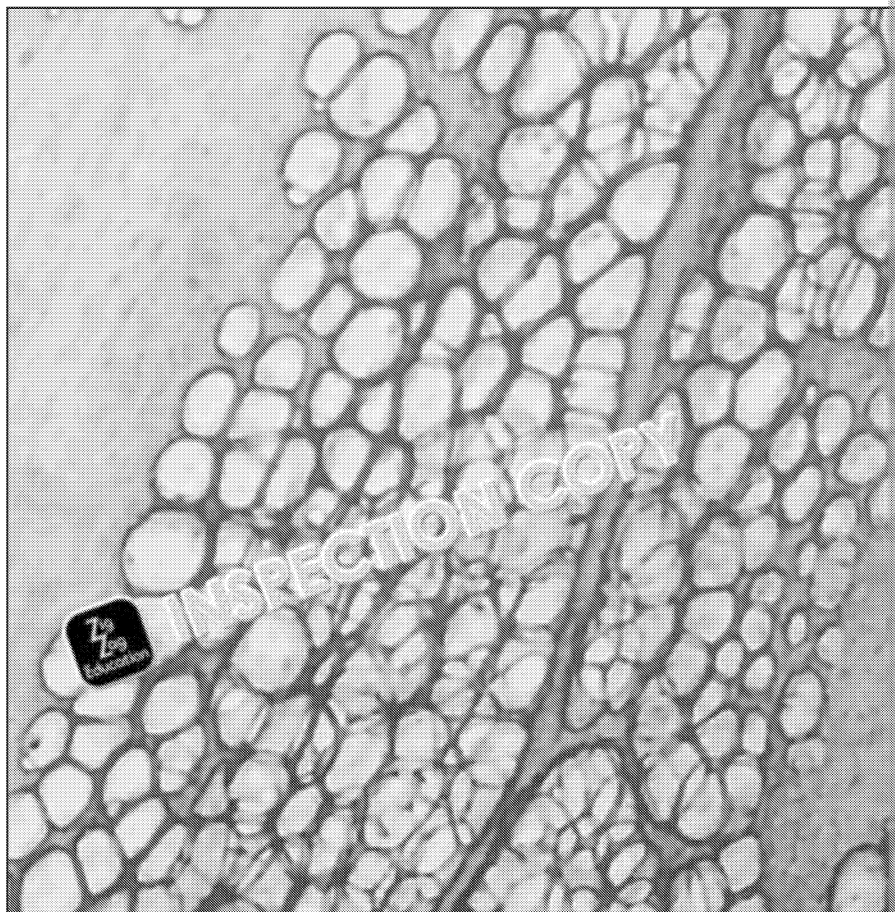


Using a light microscope

To observe a specimen using a light microscope:

1. Place slide on stage and secure with clips.
2. Set objective lens to lowest power.
3. Position stage at its highest point.
4. Turn on light / direct light onto mirror and check brightness through eyepiece.
5. Use coarse then fine focusing wheel to bring specimen into focus.
6. Adjust objective lens to medium then high power and use coarse then fine focusing wheel to bring specimen into focus.

Photomicrographs showing a range of eukaryotic cells:



Adipose cells which store fat are visible as a layer of filled cells of di
Fat is stored in these cells as an energy store, ready for when food su

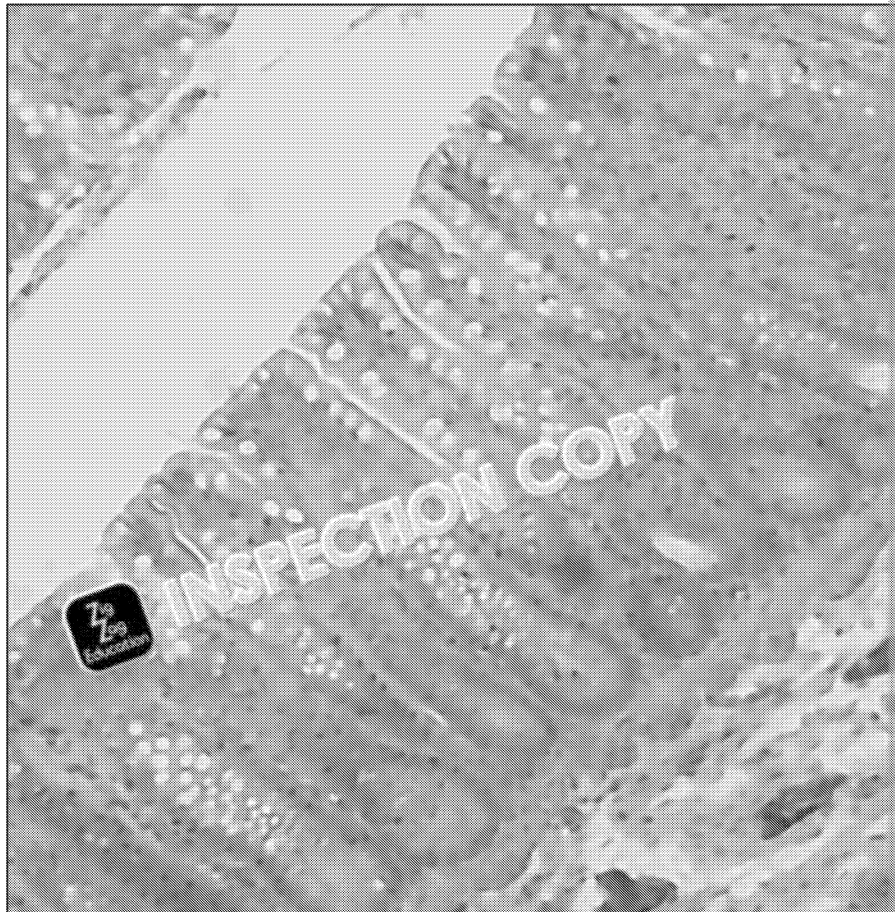
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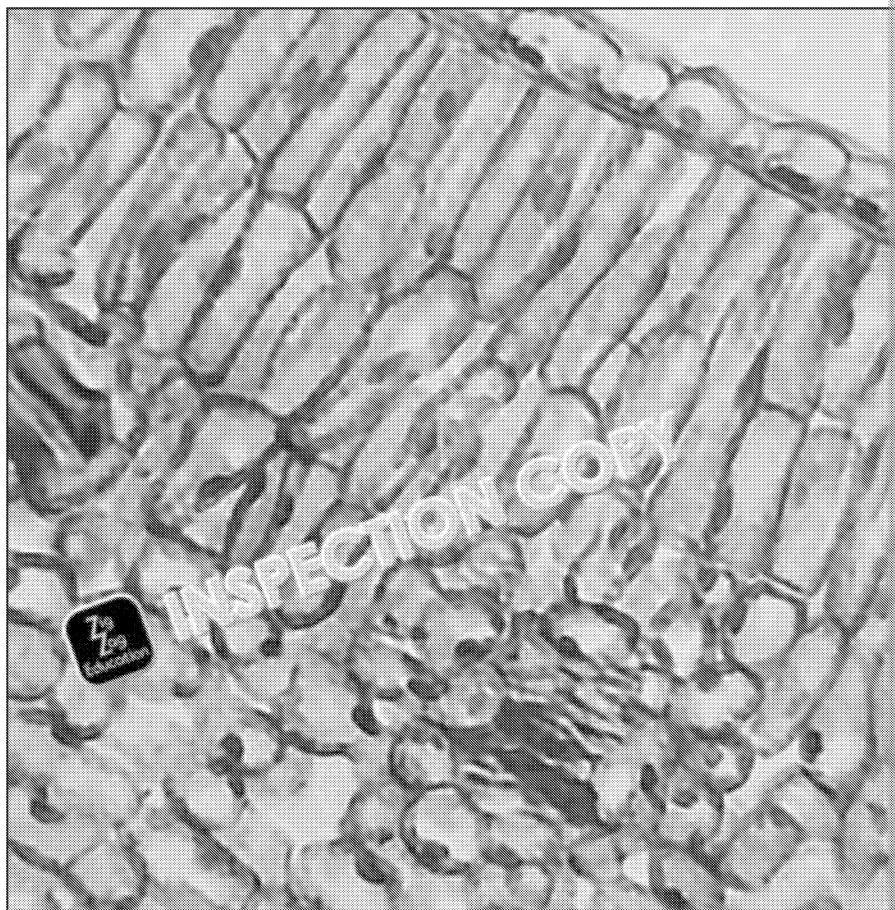
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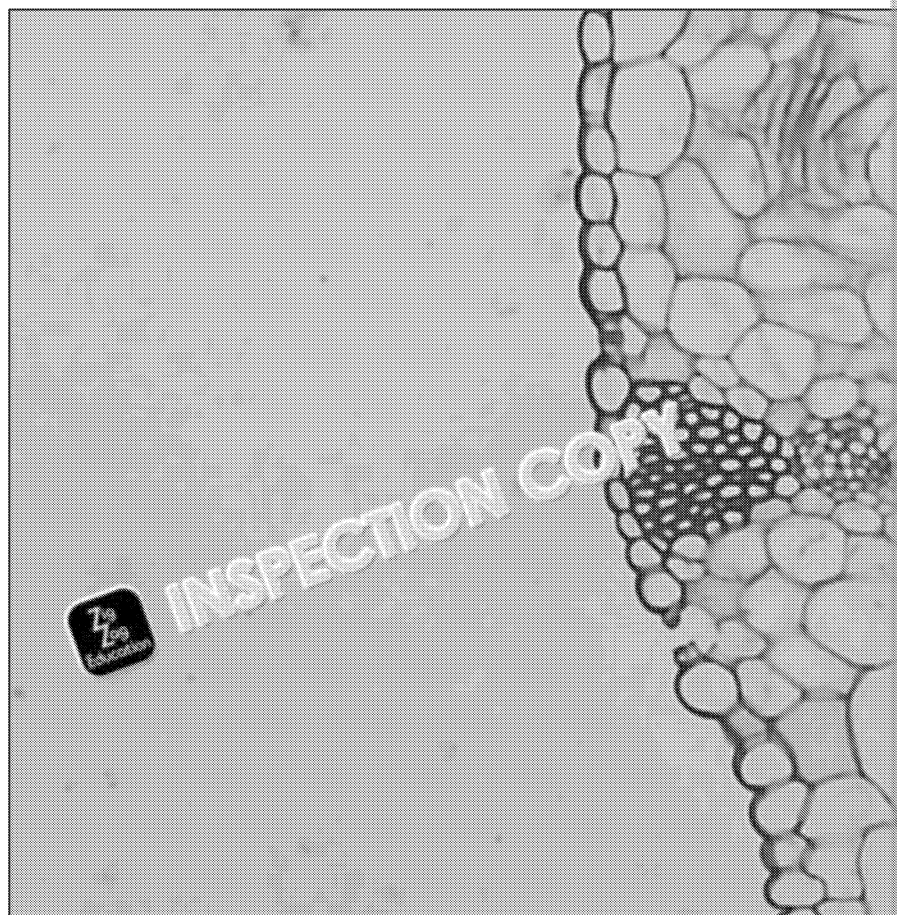
Columnar epithelial cells line the trachea and waft mucus up to the back of the throat. To see the cells on the left of the image adjacent to the gap, but to see them in detail, an electron micrograph is needed.



Palisade mesophyll cells in a leaf cross section, which contain many chloroplasts for photosynthesis. Spongy mesophyll cells, which also contain chloroplasts, are also visible across the top of the image, and the spongy mesophyll cells with air spaces are visible at the bottom.

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A stoma is visible in the lower epithelial tissue layer of a plant leaf, as a gap, surrounded by guard cells, which allow gas exchange. On the right are air spaces between the spongy mesophyll cells. Above the stoma indicates vascular tissue, which includes xylem and phloem.

Test your knowledge

- Which objective lens should be used first when viewing a specimen using a light microscope?
- Suggest why sunlight must never be used with a light microscope which has a high magnification. What would be better to use instead?

Electron microscopes

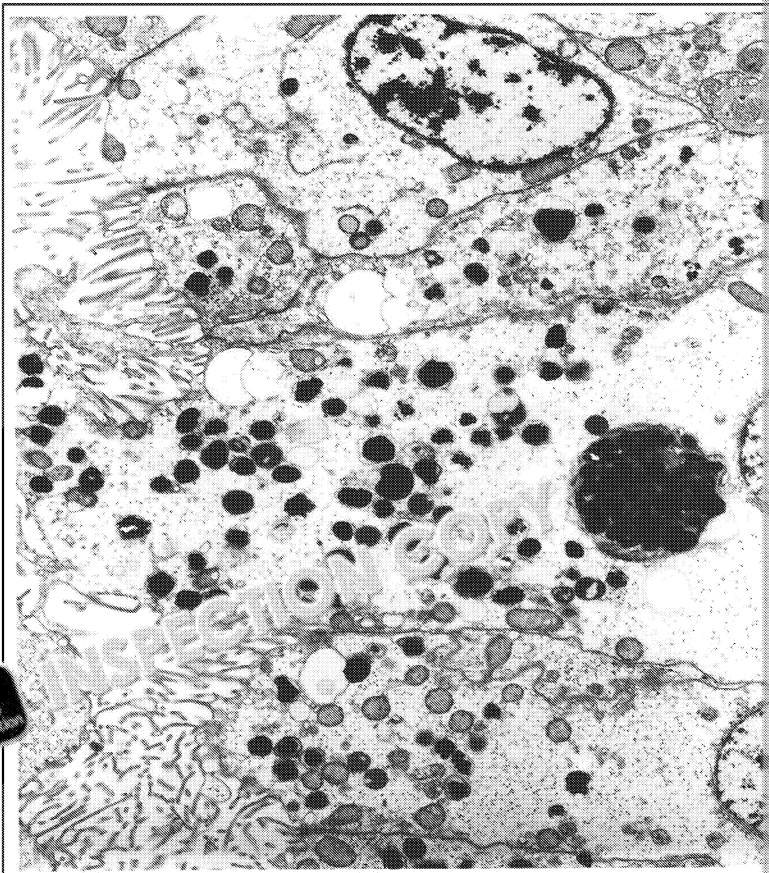
Electron microscopes use electrons, rather than visible light, to create an image. As a result, they have much higher resolution and magnification properties than a light microscope: up to $\times 2000$ magnification and 0.2 nm resolution, compared to up to $\times 500,000$ magnification and 0.2 nm resolution with a light microscope.

There are two types of electron microscope: transmission and scanning. Transmission electron microscopes use electron beams through the sample, so the sample must be extremely thin to allow electrons to pass through. Scanning electron microscopes bounce electrons off the surface of a specimen, which has used a conductive coating. This allows large specimens to be observed, and their surface detail examined.

An electron microscope is significantly more complex to use than a light microscope. It requires a vacuum inside the electron microscope, so that there are no air particles to interfere with the electron beam. This means that the specimen must be dead and carefully prepared to withstand the electron beam. Stains, often containing heavy metals, must be used to show up features, and the process is complex and technical.

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Scanning electron microscopy produces images of 3D specimens, showing the surface details seen with a transmission electron microscope. Colour in electron micrographs is artificially added.



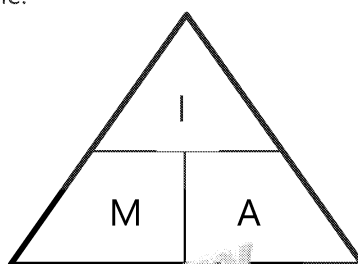
A tardigrade taken using a scanning electron microscope

Calculating magnification

To calculate the magnification of an object when viewed using a microscope, use the equation:

$$\text{magnification} = \frac{\text{image size}}{\text{actual size}}$$

You may prefer to use the equation triangle:



When calculating magnification, remember the following:

- Always show all your workings.
- Include the units in all your workings.
- State what value you know first, then rearrange the equation.
- Convert all values so that the units are the same: if using μm for one value, then convert the other value to μm and magnification will be correct.
- Plug in the numbers, with their units.
- Solve the equation then check your maths by doing the calculation the other way round.

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Worked example

A red blood cell (erythrocyte) measures $7\text{ }\mu\text{m}$ but appears to be $2800\text{ }\mu\text{m}$ when viewed in a light microscope. Calculate the magnification of the microscope.

- State the values that you know:
 - $M = ?$
 - $I = 2800\text{ }\mu\text{m}$
 - $A = 7\text{ }\mu\text{m}$
- State the equation:
 - $M = I \div A$
- Plug in the numbers with their units:
 - $M = 2800\text{ }\mu\text{m} \div 7\text{ }\mu\text{m}$
- Solve the equation:
 - $M = \times 400$
- Check your work:
 - $A \times M = I$
 - $7\text{ }\mu\text{m} \times 400 = 2800\text{ }\mu\text{m}$

Test your knowledge

You will need a calculator for these questions. Remember to show your workings.

- A macrophage (type of white blood cell) measures 8.4 mm down a light microscope with a magnification of $\times 400$. Calculate the actual size of the cell, in micrometres.
- An *amoeba* measures $2.4\text{ }\mu\text{m}$ and is being viewed at medium power with a magnification of $\times 100$. Calculate the apparent size of the amoeba in micrometres.

**Recap questions: Structure and function of cells and tissues**

- Salmonella* are an example of bacteria which have a plasma membrane, a cell wall and a nucleus. Describe the functions of these three cell parts, and suggest why all three are important for the survival of the bacteria. (4 marks)
- Describe how plant strength and structure are maintained with cell features.
- Clostridium botulinum* bacteria produce the botulinum nerve toxin. How might this toxin be identified in the lab? (4 marks)
- Before focusing a specimen using a light microscope, the stage with the specimen must be moved to the highest point without touching the objective lens. Explain why. (2 marks)
- A specimen of Ebola virus appeared to be 143.56 mm under an electron microscope with a magnification of $\times 148,000$. Calculate the actual size of this virus in micrometres. (4 marks)

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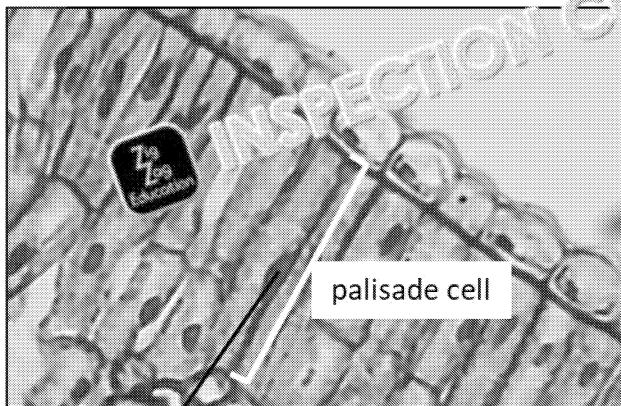
A2 Structure and function of specialised multicellular organisms



Key points covered

- Structure and function of plant cells and tissues, including palisade mesophyll and phloem tissue
- Structure and function of animal cells, including sperm and egg, erythrocytes, leucocytes

Palisade mesophyll cell

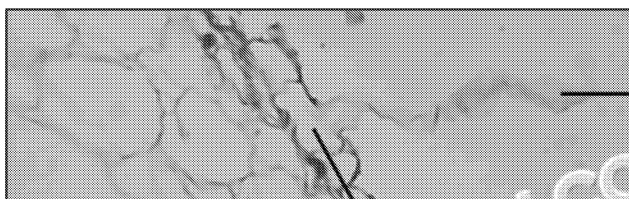


nucleus

Palisade mesophyll cells are responsible for most of the photosynthesis that takes place in leaves. Palisade mesophyll cells appear in the top layers of leaves. Their specialised features include:

- tall, columnar shape, packed tightly together to form a continuous layer for maximum light absorption;
- many chloroplasts for maximum photosynthesis;
- chloroplasts that can move through the cytoplasm to position themselves for maximum light absorption;
- thin cell walls for rapid gas exchange, especially carbon dioxide entering;
- large vacuole to maintain turgid pressure and ensure the optimal shape of the cell.

Root hair cell



hair-like protrusion

Root hair cells are found on the surface of plant roots. They are responsible for absorbing water and dissolved nutrients, such as mineral ions, into the plant, for transportation throughout the plant.

Root hair cells have two key specialised features:

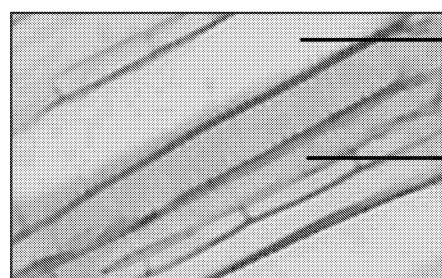
- the hair-like protrusion which significantly increases the surface area of the cell for water and nutrient absorption;
- a significant number of mitochondria, which provide energy for active transport of ions, including nitrates and magnesium.

nucleus

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Xylem tissue



vessel

lignin

parenchyma cells

lignified tissue

no end wall

hollow tube

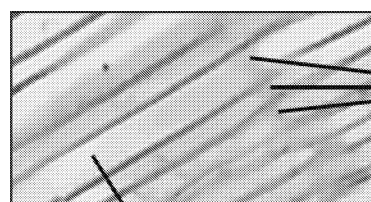
Xylem vessels transport water up the plant from roots, through shoots to the leaves, forming the **transpiration stream**. This is essential for moving water and dissolved minerals through the

transpiration stream – the movement of water up through the xylem

Xylem vessels are highly specialised. They are formed from dead cells joined end to end with no end plates or cytoplasm. Their walls are reinforced with lignin – a strong carbohydrate – which is impervious to water. These features mean that:

- water can only leave through structured pits in the vessel walls, so water reaches the leaves
- there are no obstructions through the vessel to slow down the movement of water
- the vessel is strong and stays open even when water pressure is low.

Phloem tissue



sieve plates

sieve tube

sieve plate

nucleus

sieve tube cell

Phloem tissue is responsible for moving sucrose and other dissolved nutrients, including amino acids, from where they are made to where they are needed. Phloem tissue is made of living sieve tube cells connected end to end. These cells have holes in them called sieve plates, and these allow the movement of substances from one cell to another. Companion cells are adjacent to the sieve tube cells and facilitate the movement of substances into or out of the vessel.

The sieve tube cells have a large, wide vacuole through which the substances move but do retain their nucleus, pushed up against the cell wall to maximise the space available.

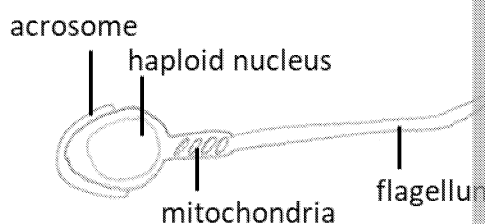
The companion cells have large quantities of mitochondria which provide energy for active transport. This is required for active transport by membrane-bound carrier proteins which are used to move the substances from the source cells into the companion cell, then into the sieve tube.

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Sperm cell

Sperm cells carry the genetic material from the male and fuse with the female egg during fertilisation to start the creation of a new organism. As a **gamete**, they have 50 % of the DNA that a normal body cell contains: they are **haploid**.



They also have a number of other specialised features:

- a flagellum which provides a whip-like tail to propel the sperm through its environment;
- many mitochondria which provide ATP to fuel its movement;
- an acrosome on the head of the cell which contains enzymes capable of digesting the zona pellucida, enabling fertilisation.

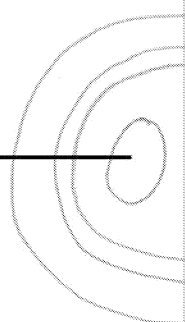
Egg cell

Egg cells carry the genetic material from the female and are fertilised by a sperm cell. They are also a gamete, so are haploid, containing half the normal body cell amount of DNA. Other specialised features include:

- a rich cytoplasm with lipids and lysosomes, providing sufficient nutrients for early stage **embryonic** development until the placenta is available for this purpose;
- a jelly-like layer outside the cell membrane, called the zona pellucida, which prevents other sperm from entering;
- a further layer on the outside, containing ovary follicle cells.

haploid nucleus

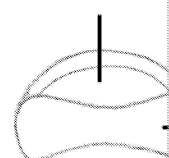
embryo – the first stage of development of a multicellular organism



Erythrocytes



biconcave shape



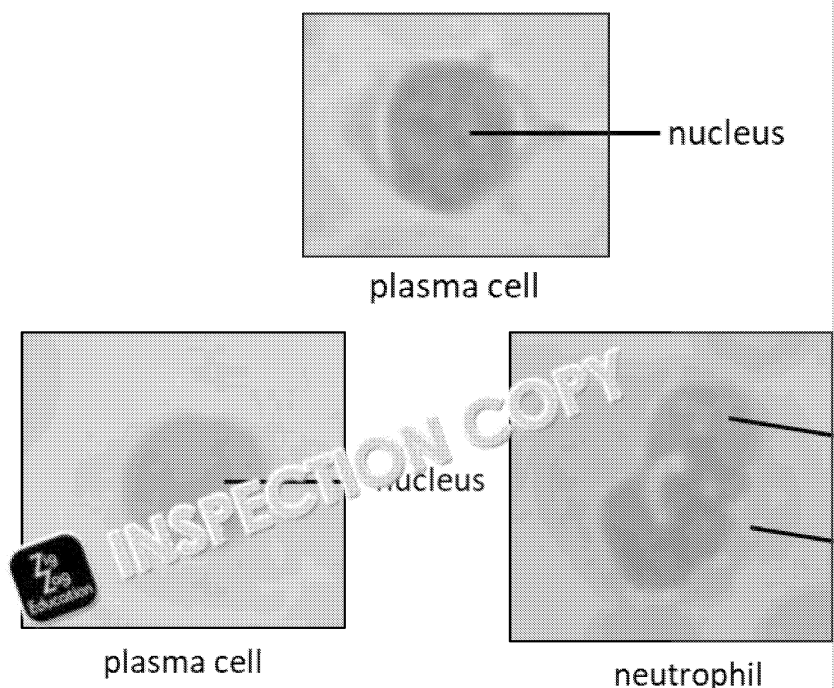
Erythrocytes are red blood cells. They transport oxygen around the body. Their specialised features include:

- no nucleus, maximising space for haemoglobin;
- haemoglobin, an iron-containing **conjugated** protein, which binds oxygen;
- a biconcave disc structure which increases the surface area so increases the rate of diffusion of oxygen into or out of the cell.

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Leucocytes

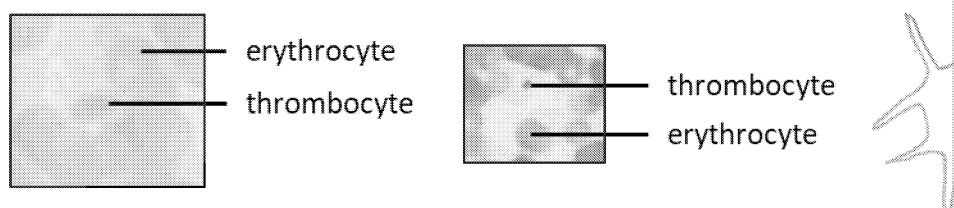


Leucocytes are white blood cells. There are many different types of leucocytes, all of which are part of the immune system.

Neutrophils are the most easily identifiable, because of their multi-lobed nucleus. They move through small gaps between cells to reach the source of infection. They have granules and many enzyme-containing lysosomes which are used to destroy pathogens.

Plasma cells have a large nucleus and a large quantity of rough endoplasmic reticulum. They manufacture large quantities of antibodies, which provide a specific response to an antigen.

Thrombocytes

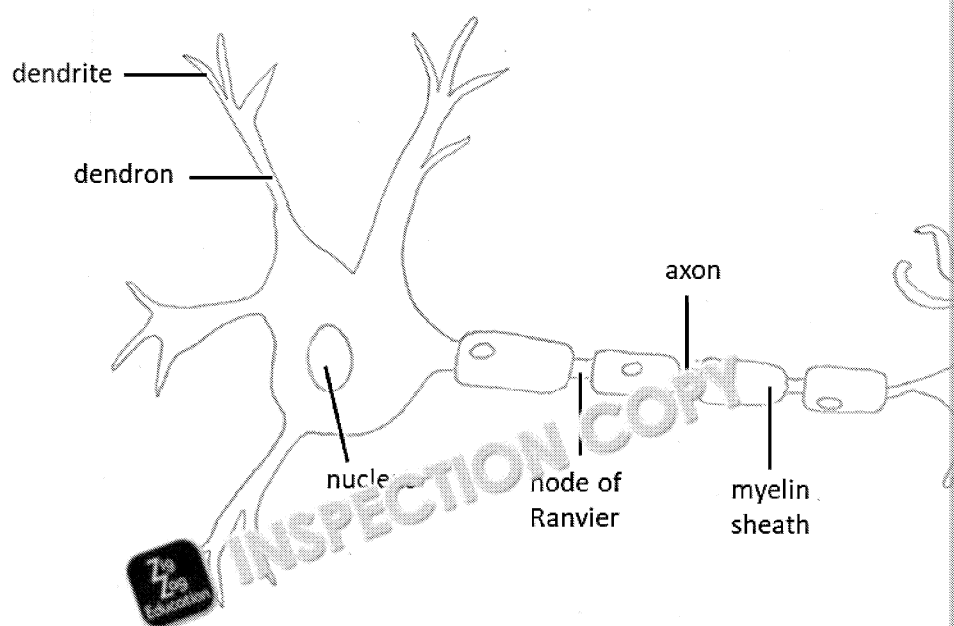


Thrombocytes, or platelets, are essential for blood clotting and scab formation. They are made from the remnants of other cells, contain no nucleus, and have many small surface **protrusions** which allow them to attach to blood vessel walls and each other to plug a hole. They cause blood clotting by releasing chemicals to trigger this process and are therefore essential in reducing blood loss.

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Neurones



Neurones are highly specialised cells responsible for conducting electrical signals from the brain and then to effectors, enabling our body to understand and respond rapidly to the external environment. They are so specialised and complex in structure that, once damaged, which is why nerve damage is usually permanent.

Neurones have the following specialised features:

- dendrons, which divide into dendrites, to connect to other neurones or receptors per neurone, allowing multiple pathways to be possible;
- a long axon which connects one neurone to another in different parts of the body;
- multiple terminals at the axon, so the neurone can connect to multiple other neurones;
- a myelin sheath, present on many but not all neurones, and created by Schwann cells around the axon and prevent it from being interrupted;
- nodes of Ranvier on the axon between each myelin sheath, which allow the electrical signal to jump the gap, thus greatly increasing its speed by up to 100 times.

Test your knowledge

- State which cell has a biconcave disc shape and no nucleus, and explain why this is important for its function.
- State which cell has sieve end plates and little cytoplasm, and explain why this is important for its function.
- State which cell has a flagellum and mitochondria, and explain why these are important for its function.
- State which cell has either a small nucleus or a large nucleus and lots of cytoplasm, and explain why these are essential for its function.

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Recall questions: Structure and function of specialised cells

- 1 Thrombocytopaenia is a disease where there are low numbers of thrombocytes. Describe the symptoms of this disease and why it could be dangerous if left untreated. (3 marks)
- 2 An over-enthusiastic gardener brushes the roots of their plants before replanting. Describe the damage this will cause to the plant roots, and what the impact of this damage will be. (3 marks)
- 3 Plants which have been grown to be lignin-deficient are often also dwarfed. Explain why this might be the case. (4 marks)



A3 Structure and function of biological tissues



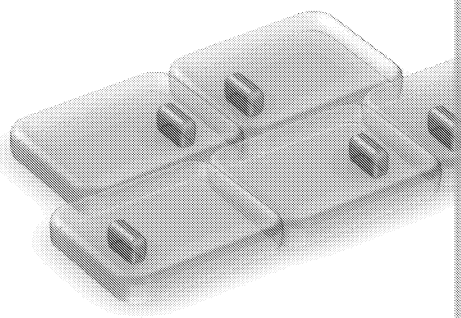
Key points covered

- The structure and function of specialised tissues, including epithelial, endothelial, muscular and nervous tissue
- Diseases that are caused by damage to some of these tissues
- How nerve impulses are transmitted, including saltatory conduction
- The function of the nervous system
- Chemical imbalance and the effects of drugs on the synapse

Epithelial tissue

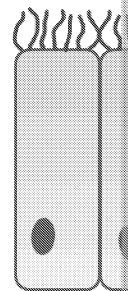
Squamous epithelium

Squamous epithelial tissue is made up of thin, flat cells, one layer thick, which form a barrier between the air in the alveoli and the blood capillaries. This allows gas exchange to occur efficiently across the squamous epithelium because of the specialised features of the cells.



Columnar epithelium

Columnar epithelium in the lungs comprises of two types of tall, column-shaped cells: ciliated cells and goblet cells. These cells work together to protect the lungs from pathogens: goblet cells release mucus which traps pathogens and dirt particles; cilia on the ciliated cells waft this mucus up the trachea to the throat so it can be coughed up and spat out or swallowed.



Chronic obstructive pulmonary disease (COPD)

Chronic obstructive **pulmonary** disease, or COPD, is a collection of diseases of the lungs, including emphysema and chronic bronchitis. These are long-lasting conditions often caused by smoking, but frequent and long-term inhalation of other particulates including flour, coal dust and welding fumes, can also cause the disease.

Emphysema is a disease of the alveoli in the lungs, where the inner lining of alveoli tears and many small alveoli merge into fewer, larger alveoli instead. This reduces the surface area for gas exchange, causing breathlessness, wheezing and an increased production of water.

Bronchitis results in inflammation of the bronchi, which are branches of pipework in the lungs. An inflammatory response is produced by these columnar cells, restricting the diameter of the bronchi, making it harder to inhale and exhale. An increase in mucus production also occurs because

Test your knowledge

- Describe the appearance of squamous epithelial cells and explain why this helps with gas exchange
- Describe the structure and function of goblet cells and their role in the symptoms of COPD

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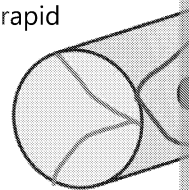
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Endothelial tissue

Structure and function

Endothelial cells make up the wall of blood vessels, including the capillaries (see diagram on previous page). Endothelial tissue is very similar to squamous epithelial tissue: the cells are thin and flat and facilitate rapid exchange of nutrients and waste products between the blood vessels and the body tissues. They also regulate the amount of fluid that exchanges between these two tissues and ensure that the blood remains fluid and does not clot unnecessarily.



Atherosclerosis

Atherosclerosis is a disease which is caused by damage to the endothelial cells of the blood vessels. The endothelial cells can be damaged by high blood pressure, high cholesterol, smoking, and inflammatory conditions such as lupus and arthritis. The damage causes particles to build up on the cells, gradually causing a build-up or plaque. This narrows the blood vessel, so that part of the body that it supplies is not getting enough blood. The plaque can also rupture or break off, which leads to a stroke or a heart attack. A blood clot can also cause a blood clot elsewhere, causing secondary symptoms.

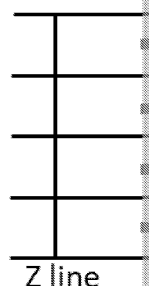


Test your knowledge

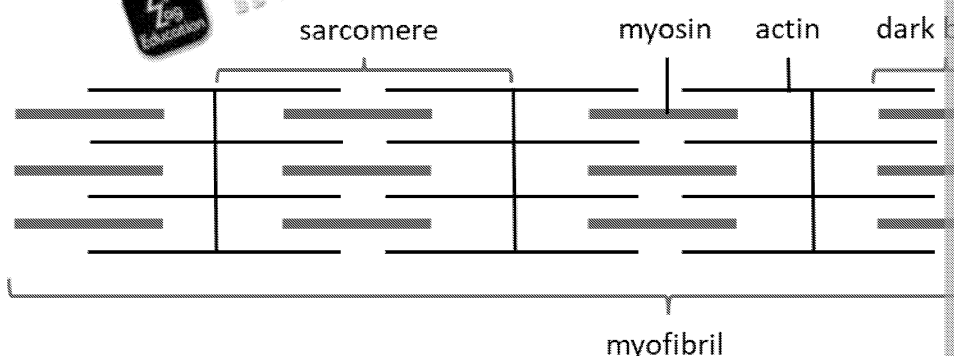
- Name two functions of endothelial cells: one similar to epithelial cells, and one unique to endothelial cells.
- Explain how atherosclerosis develops, by using one risk factor and one secondary symptom. Provide a flow chart of the development of the process.

Muscular tissue

Skeletal or striated muscle fibres form muscles within our body that can be consciously controlled. They are made up of millions of tiny units called sarcomeres, which contain actin and myosin proteins that work together to produce muscle contraction.



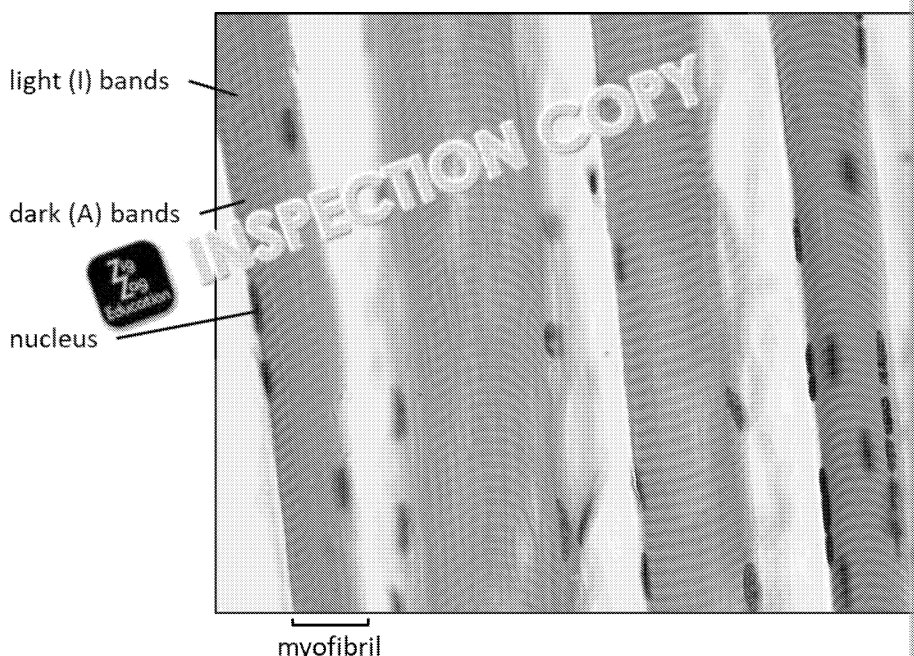
Many **sarcomeres** form together into a cylindrical fibre called a **myofibril**, and many myofibrils make up a muscle fibre, which are grouped into motor units. Motor units are then grouped again to form muscle tissue.



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In the photomicrograph of skeletal tissue, below, the striations of sarcomeres are clearly visible. The dark bands (A bands) are where myosin is present, and are particularly dark where actin and myosin overlap. The light bands (I bands) are where only actin is present. The fibres are aligned with each other so that they produce muscle contraction in one direction to be most effective. It is also possible to see the nuclei of the muscle cells – the dark oval dots. In striated muscle, one nucleus is shared between several muscle cells: they are multinucleated. This removes the nucleus from the muscle fibres, which would prevent them from contracting effectively.



There are two types of muscle fibres: fast-twitch and slow-twitch. As their names suggest, the speed of contraction varies considerably between these two groups of muscles, and therefore the type of muscle varies depending on the job of the muscle group they are found in.

Slow-twitch muscle fibres can contract for longer periods of time than fast-twitch muscle fibres. Slow-twitch muscle fibres are required for endurance-based activities, including standing upright for long periods of time, where the muscles in the back and legs must remain contracted in order to maintain posture. In comparison, fast-twitch muscle fibres contract quickly and powerfully, but can only contract for a short time. They are therefore useful for quick activities like fast movement from the arms (e.g. in tennis).

Test your knowledge

- Draw a diagram to show actin, myosin, a sarcomere, a myofibril, and the dark band.
- Describe the key differences between fast-twitch and slow-twitch muscle fibres.

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Nervous tissue

Neurone structure

In the previous section you saw how neurones are structured. Neurones transmit electrical signals down the axon, connecting one neurone to another at the synapse (see later). Sodium and potassium **ions** move into or out of the cell via gateways in the membrane. These charged particles cause the membrane to be **polarised** (positively or negatively charged), and polarisation moves like a wave down the axon.

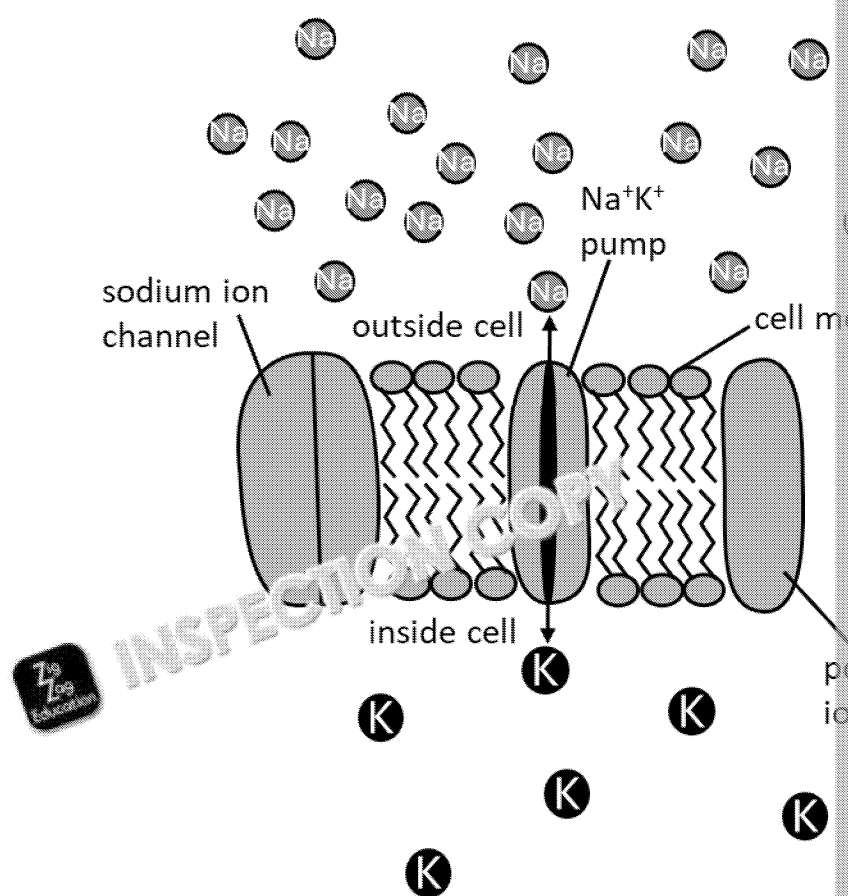
In myelinated neurones, the wave can jump from one node of Ranvier to another, out of the myelin sheath. This greatly speeds up transmission, from 1 m/s along unmyelinated axons to 100 m/s along myelinated axons. This is important especially in long neurones, such as motor neurones, which have a significant distance to be transmitted over and must do so rapidly and reliably. Some sensory neurones, do not need myelination because of the length of the neurone.

Nerve impulse conduction

Resting potential

A neurone transmits an electrical impulse along the axon using positively charged sodium ions (Na^+). When the neurone is at rest, the cell is polarised with a resting potential of about -70 mV. This is maintained by:

- the sodium-potassium pump, which pumps two K^+ ions into the axon for every three Na^+ ions it pumps out;
- the sodium ion channel being closed, so no sodium ions can re-enter the axon;
- the potassium ion channel being closed but leaky, so potassium ions diffuse down their **electrochemical gradient** and out of the axon.



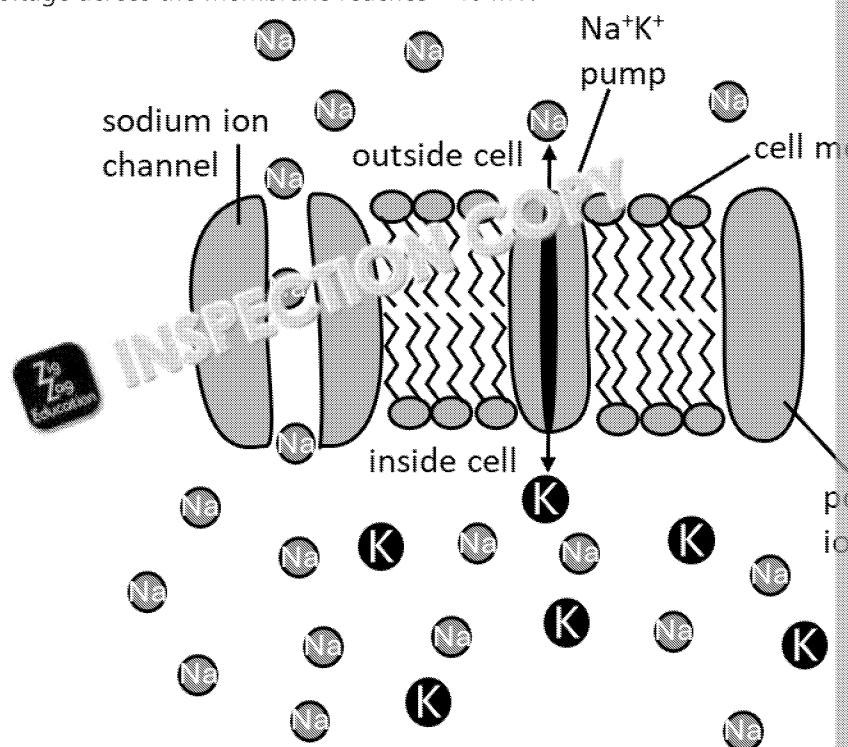
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Action potential

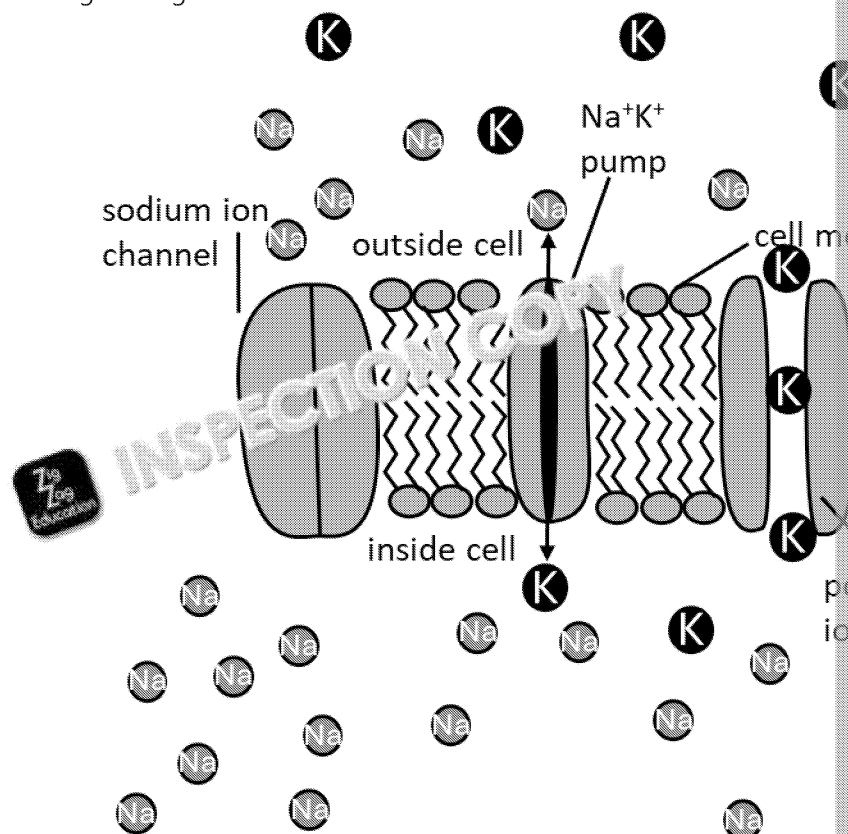
On action of a stimulus, such as a nerve receptor being triggered, the nerve **depolarises**. This occurs because the sodium ion channels open, allowing sodium ions to flood through into the axon down their electrochemical gradient.

The membrane is now more permeable to sodium ions. This causes depolarisation as the inside of the neurone is less negative than outside. Depolarisation continues until the voltage across the membrane reaches +40 mV.



Repolarisation

When the voltage reaches +40 mV, the sodium ion channels close, and the potassium ion channels open. The membrane is now no longer permeable to sodium ions but is now very permeable to potassium ions. Potassium ions move down their electrochemical gradient from inside the axon to the outside. The inside of the neurone becomes more negative again.



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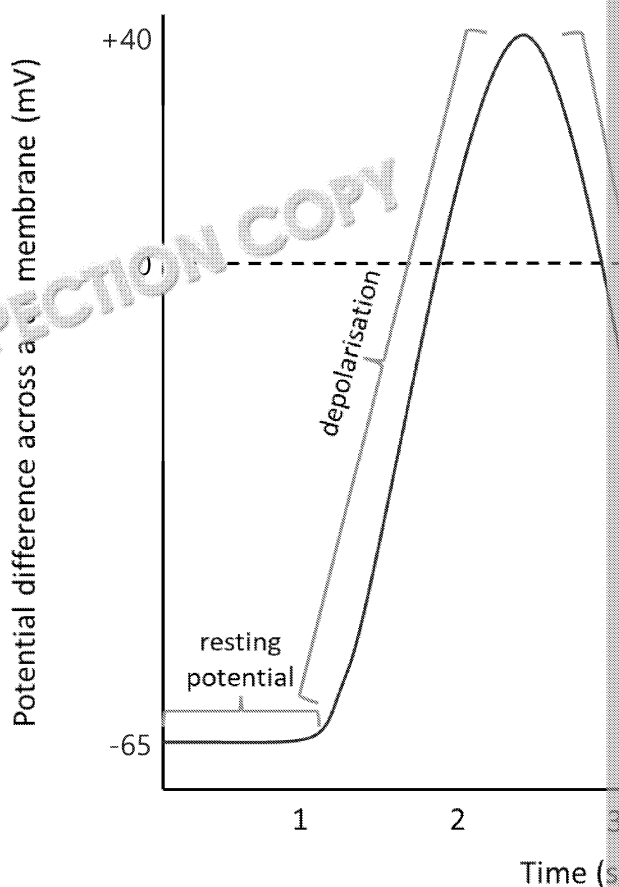


Hyperpolarisation

The process of potassium ions leaving the axon carries on beyond -70 mV, which w This is called hyperpolarisation. This causes the potassium ion channels to close, so pump is working: the membrane has low permeability to sodium and potassium ion resting potential, and it is ready for the next signal to activate it again.

Reading a nerve impulse from a potential difference graph

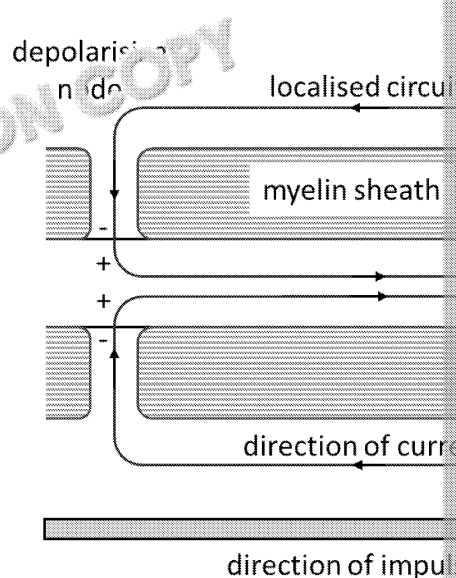
The potential difference across the axon membrane during the phases of nerve transmission can be represented on a graph. The phases of resting potential, depolarisation, repolarisation and hyperpolarisation are clearly visible.



Saltatory conduction

Saltatory conduction is the process of propagating a nerve impulse along the axon of a myelinated neurone. In unmyelinated neurones, the process of depolarisation and repolarisation to return to the resting potential has to happen along the entire length of the neurone. In myelinated neurones, most of the axon is **insulated** by the myelin sheath. This means that depolarisation and repolarisation can only happen at the nodes of Ranvier where no myelin sheath is present to prevent the movement of ions into or out of the axon.

The effect of the myelin sheath is that depolarisation occurs at one node of Ranvier, but the next node remains at the resting potential. This establishes a localised circuit between the two nodes: at the depolarising node, the movement of ions into the axon causes the axon to be positively charged and the outside to be negatively charged. Conversely at the next node along which is at its resting potential, the axon is negatively charged and the outside is positively charged. The impulse jumps along the axon from node to node, at far greater speeds than could be achieved without the myelin sheath.



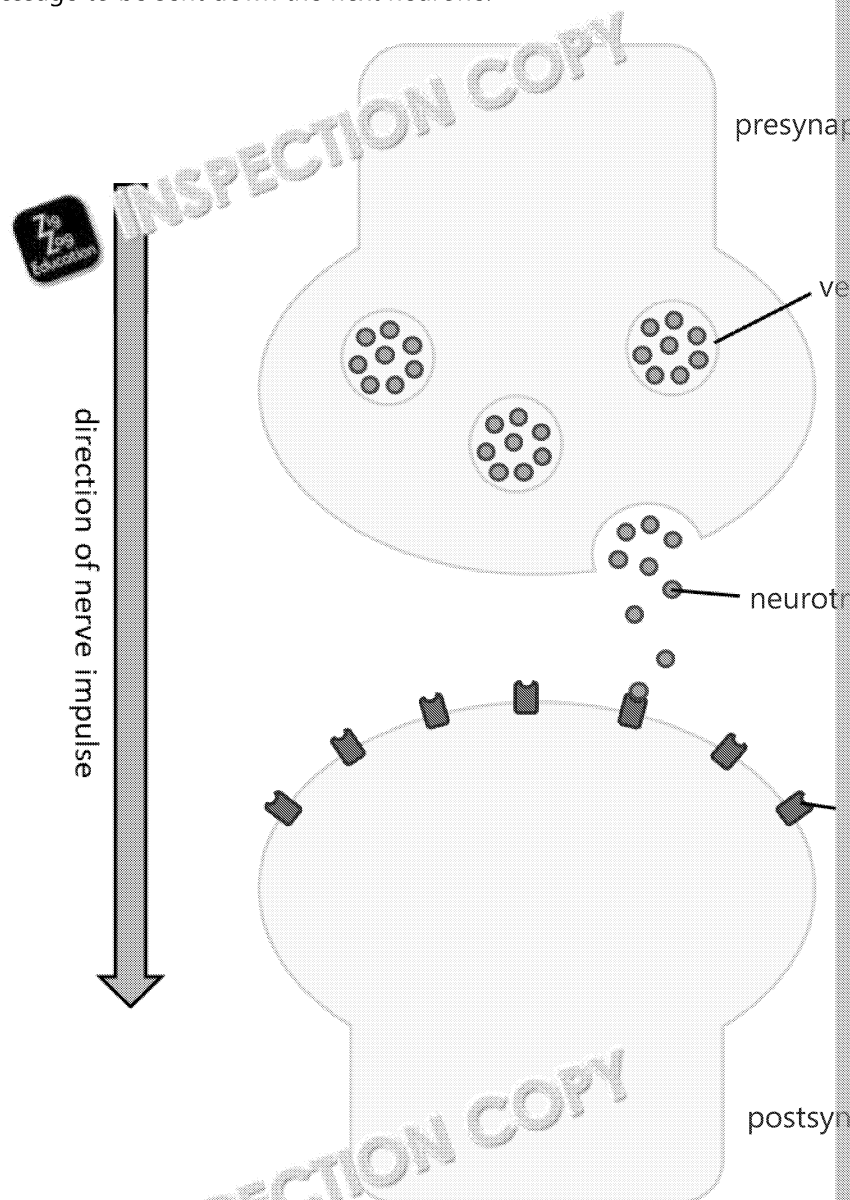
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The synapse

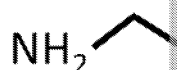
The synapse is the gap between two interconnecting neurones. Where these two neurones meet is about 20 nm wide: the synaptic gap. It is large enough to prevent the nerve impulse in its electrical form, so it is exchanged for a chemical message instead.

When the nerve impulse arrives at the presynaptic terminus, it triggers vesicles containing a **neurotransmitter** to move and fuse with the presynaptic membrane. This releases the neurotransmitter into the synaptic gap, and the neurotransmitter diffuses across the gap. The neurotransmitter binds to a receptor on the postsynaptic membrane, which triggers an onward neural impulse to be generated and the message to be sent down the next neurone.



Neurotransmitters are essential for this process, and also for ensuring that the message travels in one direction. They are stored in vesicles in the end of the presynaptic neurone only, so the message can only travel in one direction. Furthermore, receptors are only present on the postsynaptic neurone, so the neurotransmitters can only bind to them. Together, these two features ensure that the nerve impulse continues onwards and cannot go back down the axon it has just moved along.

There are many different neurotransmitters, including serotonin, dopamine, acetylcholine, endorphins and adrenaline. The diagram to the right shows dopamine.



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Brain damage due to chemical imbalance

The brain is a delicate organ and can be subject to disease. Chemical imbalances in the brain can cause a range of diseases by either slowing down or desensitising the nerve impulses, speeding it up or extending it. Any imbalance in this process will lead to impaired function.

Parkinson's disease is a chronic and progressive condition with a range of symptoms including tremors in the hands, arms, legs, face and the jaw; stiffness in arms, legs or torso; slow and poorly coordinated movement; loss of balance and impaired ability to speak. It is thought to be caused by the loss of function of dopamine as a neurotransmitter, which over time prevents these neurones from communicating effectively, leading to the range of symptoms.

There are many other diseases caused in a similar way. Many mental health illnesses are caused by an imbalance of neurotransmitters in the brain. Alzheimer's disease is caused by a build-up of plaques that prevent specific neurones from functioning. Motor neurone disease and multiple sclerosis are caused by damage to the myelin sheath surrounding neurones, which prevents them from functioning properly.

Effect of drugs on the synapse

Drugs can affect how synapses work, because synapses involve the transmission of information from one neurone to another. The effect that a drug has depends on the drug. Drugs can:

- increase or decrease the speed at which neurotransmitters are manufactured by increasing the precursors available or inhibiting enzymes that synthesise the neurotransmitter;
- increase the quantity of neurotransmitter available by inhibiting the enzymes that break it down;
- decrease the quantity of neurotransmitter available by causing the vesicles to release neurotransmitters prematurely or by causing neurotransmitters to be degraded by enzymes in the cytoplasm;
- increase or decrease the quantity of neurotransmitter released into the synaptic cleft;
- prevent or accelerate the reuptake of neurotransmitter by the presynaptic terminal;
- increase the time that the neurotransmitter is attached to the receptor or in the synaptic cleft;
- simulate the effect of the neurotransmitter by binding to the receptor instead of the neurotransmitter binding.

Agonistic drugs replicate the function of the neurotransmitter by binding to the synaptic receptors, producing the effect of the neurotransmitter and enhancing transmission in the onward neurone. Some drugs produce these effects.

Antagonistic drugs bind to the synaptic receptors and prevent the neurotransmitter from binding, stopping synaptic transmission in the onward neurone. Cannabinoids and alcohol are examples of antagonistic drugs.

Test your knowledge

- Describe what is meant by resting potential.
- Describe how saltatory conduction enables a very fast nervous impulse.
- Describe the features of the synapse.
- Describe three ways in which the synapse could be affected by drugs.

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Recap question: Structure and function of biological tissue

- 1 Describe the similarities and differences between epithelial and endothelial cells.
- 2 Explain how muscle fibres are arranged and why this is necessary for their function.
- 3 Describe how changing membrane permeability causes a nerve impulse to be generated.



B: Structure and function of biology

B1 Structure and function of water

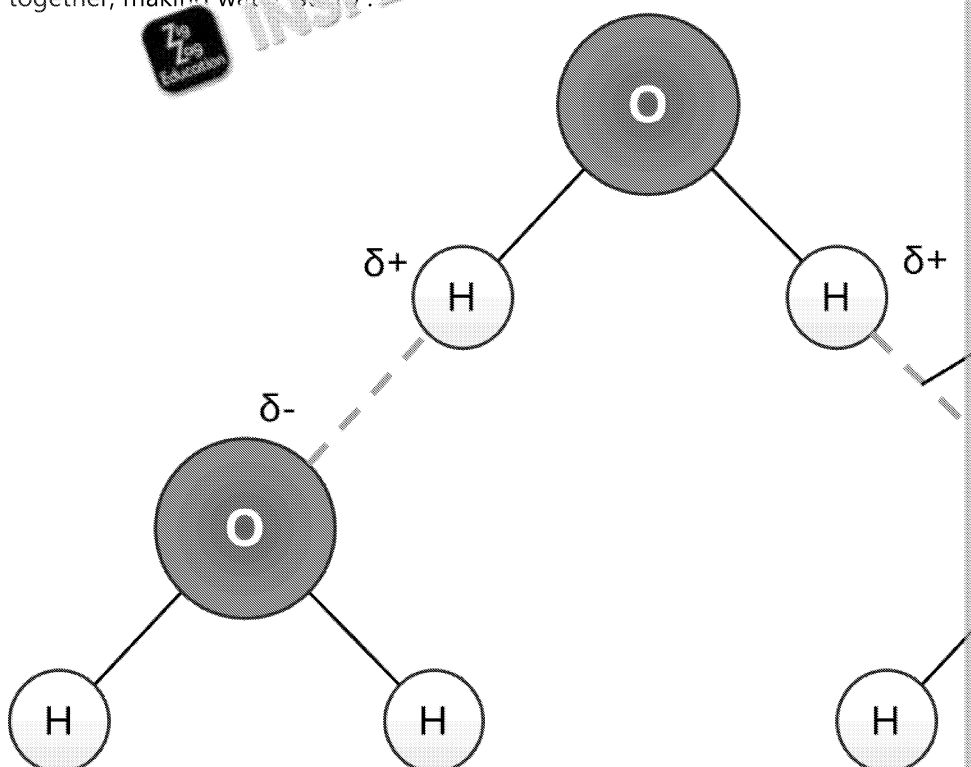


Key points covered

- Structure and function of water molecules, including their polar nature
- Function of water, including as a solvent, a transporter and a balance

Structure of water

Water, or H_2O , is a simple covalently bonded molecule consisting of two positively charged hydrogen atoms and one negatively charged oxygen atom. This makes a polar molecule, with positive and negative charges on the other end, which allows hydrogen bonds to form, binding water molecules together, making water 'sticky'.



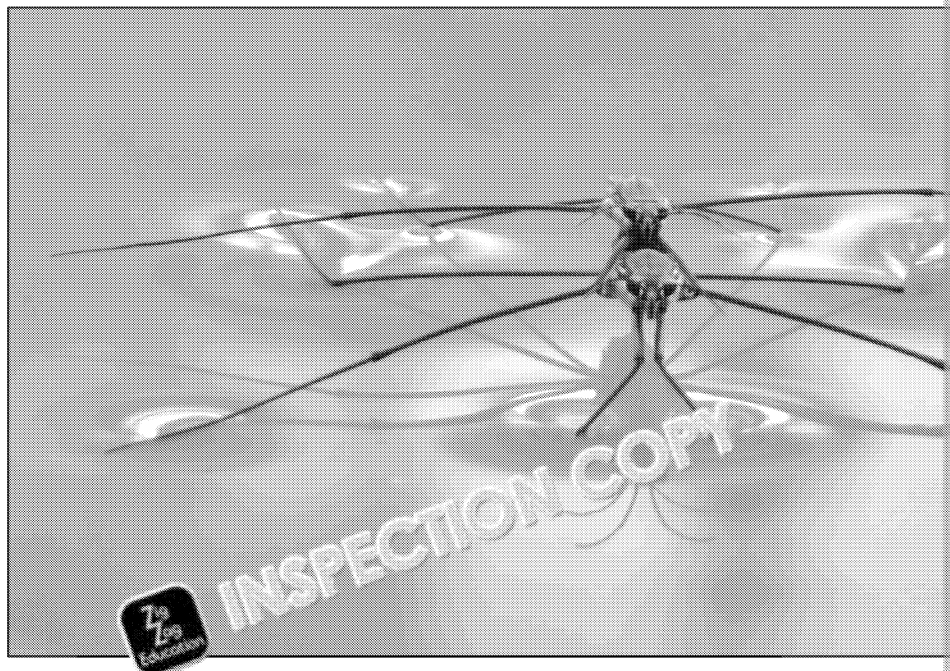
Functions of water

Water is a solvent: many other chemicals can dissolve in water due to its polar nature. For example, sodium chloride (NaCl), the Na^+ ions are attracted to the negatively charged oxygen atom, and the Cl^- ions are attracted to the positively charged hydrogen atoms. Because water is able to act as a solvent, it is an excellent medium for chemical reactions to take place within, and for substances to be transported. This is why water makes up more than 70 % of the content of the cytoplasm in cells.

The polar nature of water makes it 'sticky': water molecules adhere to other water molecules through cohesion and adhesion. Cohesion allows water to be drawn through cells and vessels, such as xylem, through transpiration pull. The hydrogen bonds between water molecules means that one molecule of water is attracted to the next molecule, and so on. Like a long train, when the first molecule moves, the rest of the pathway of water molecules, they will all move through the system.

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The cohesive nature of water is also visible on its surface: water forms a 'skin'. This is due to water molecules being attracted to each other more than other particles in the environment, through hydrogen bonding. Insects like pond skaters, with a low density and a high surface area can be supported by the surface tension that exists between the water molecules.

The polar nature of water also enables it to regulate pH and electrolytes, by forming hydrogen bonds with other polar molecules including ions, amino acids and nucleic acids. This manages the concentration of electrolytes and the overall pH of the solution.

Water has a high specific heat capacity. This means that it requires a substantial amount of energy to heat water by 1 °C. Therefore, water is relatively thermostable, doesn't heat up or cool down quickly, which can offset temperature changes caused by exothermic or endothermic chemical reactions. This helps buffer against abiotic climate changes such as winter or summer. This enables life to thrive and biodiversity to increase.

Water unusually becomes less dense as a solid than it is as a liquid, meaning that solid water (ice) floats. This provides an insulating layer on the top of bodies of water such as lakes and seas, preventing them from freezing below it slightly warmer than it would otherwise be, and being more hospitable to life.

Test your knowledge

- Explain why water's high specific heat capacity is important for survival of life.
- Describe what is meant by 'polar', and which features of water it is responsible for.

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Recap questions: Structure and function of water

- 1 Explain how hydrogen bonds form between water molecules. (3 marks)
- 2 Use the polar nature of water to explain how cohesion tension pulls water up the xylem of a plant. (5 marks)
- 3 Explain, using the properties of water, why lakes hidden deep within permanent ice sheets remain for many thousands of years. (3 marks)



B2 Structure and function of carbohydrates



Key points covered

- Structure of carbohydrates, including monosaccharides, disaccharides and polysaccharides
- Function of carbohydrates, including energy storage and transport

Structure of carbohydrates

Carbohydrates include a diverse range of molecules from simple sugars to complex molecules. They have a variety of functions, including as a store of energy and to create structures within cells. Despite a wide range of structures and functions, they all contain carbon, hydrogen and oxygen.

Monosaccharides (single sugar carbohydrates) are able to form straight chains or rings. At the end of the chain, and in disaccharides these are either an aldehyde group or a ketone group.



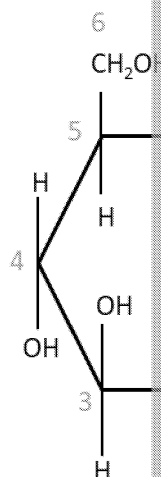
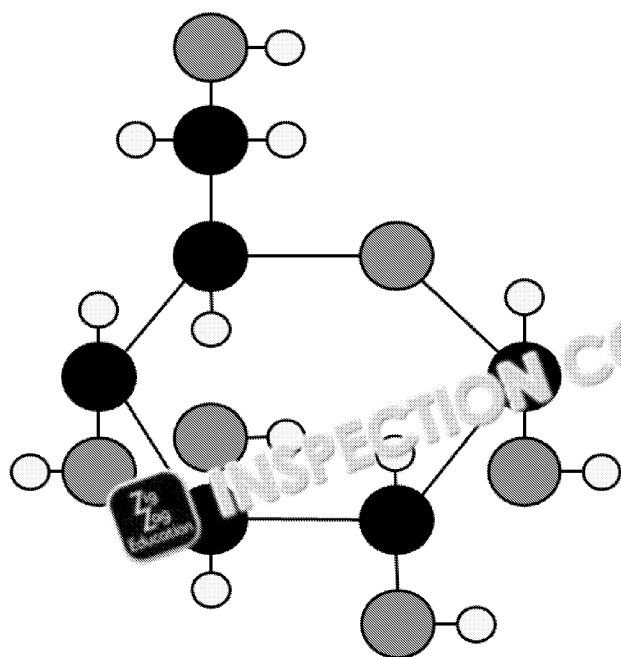
Monosaccharides are able to join together to form disaccharides and polysaccharides. These are made up of many carbohydrate molecules, each with different structures and properties.

Monosaccharides

Monosaccharides are monomers: a single unit of a carbohydrate that cannot be made from smaller units. They can combine together to make more complex molecules.

Monosaccharides include α glucose and β glucose, galactose, fructose, ribose and deoxyribose.

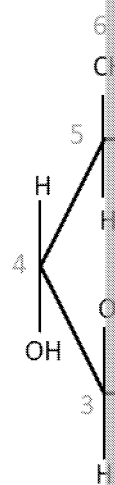
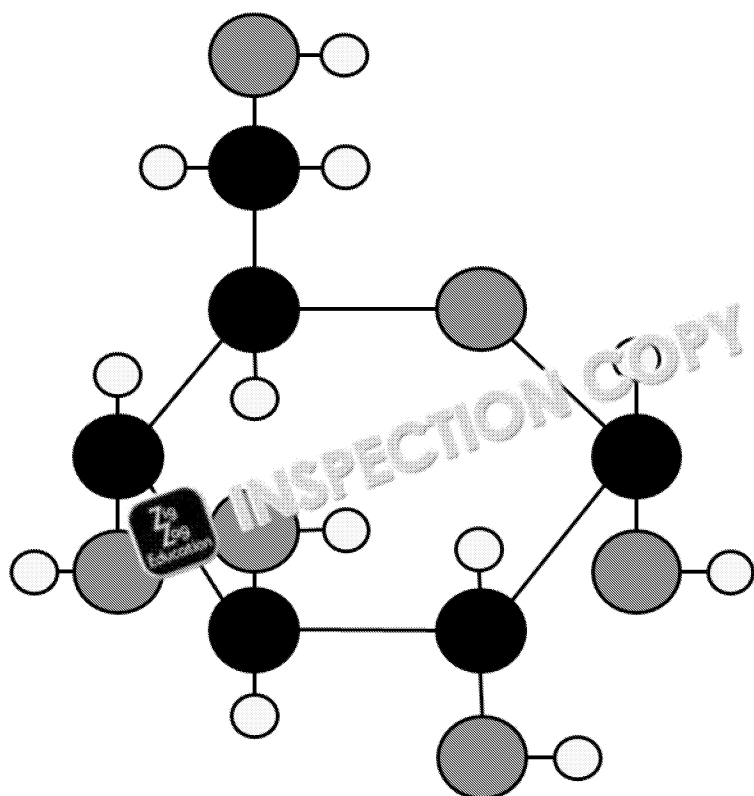
α glucose, $C_6H_{12}O_6$, contains six carbon atoms, and therefore is a hexose sugar. The structure of α glucose in its cyclic form is shown below. The carbon-1 is positioned down, as highlighted in the 2D structural diagram, below. C



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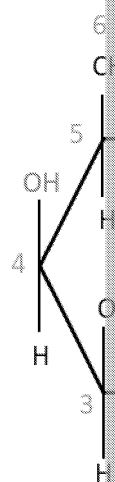
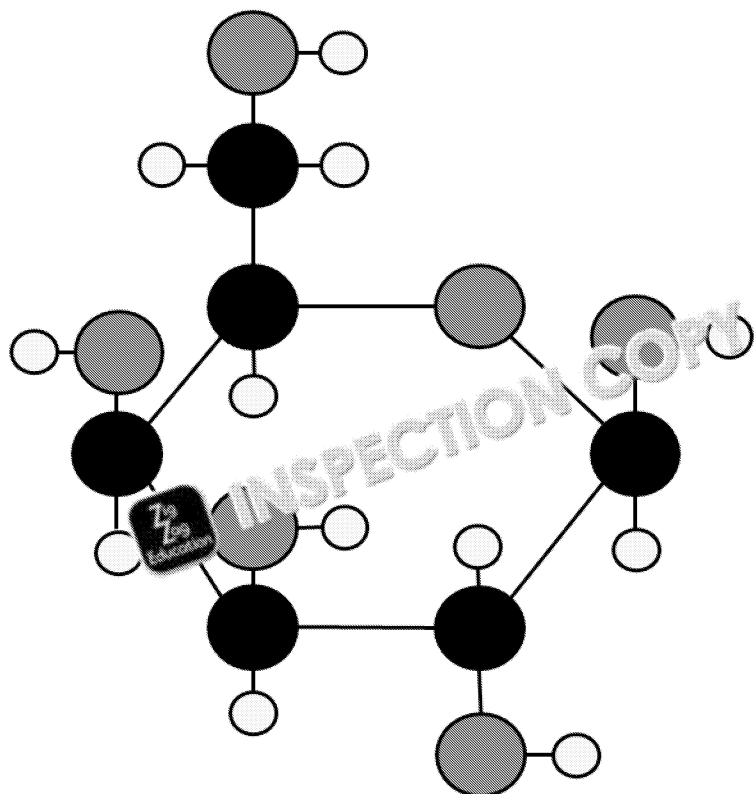


β glucose is otherwise identical to α glucose, except for the orientation of the OH group on the diagram below.



Both these sugars are polar, so are soluble in water.

Galactose, another hexose sugar, adds on one further structural change from β glucose. The OH group is reversed, as shown in the diagram below. Both differences between α and β are highlighted. When glucose and galactose bind, lactose is made.

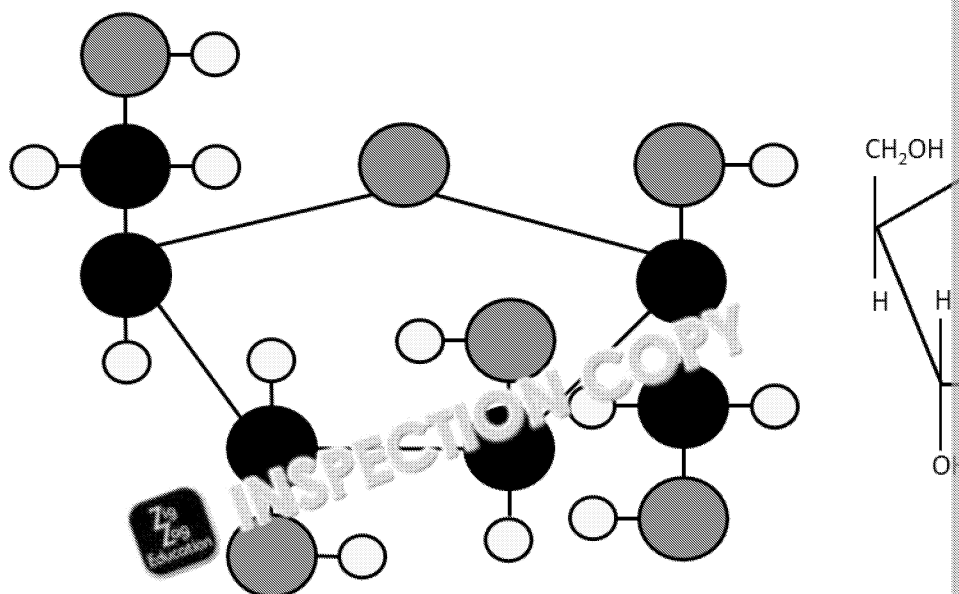


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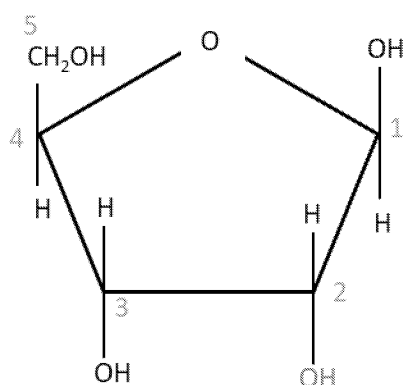


Fructose, a sugar found in many fruits, is also a hexose sugar; however, when found in its cyclic form, it is a five-membered pentagon ring with two CH_2OH groups as appendages, as shown in the diagram below.

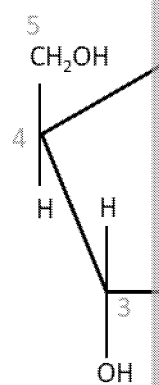


Ribose and deoxyribose are pentose sugars, containing five carbon atoms. Ribose and deoxyribose form the sugar part of DNA.

Ribose differs from fructose by missing the first CH_2OH group:



Deoxyribose differs from ribose by having a hydrogen atom instead of a hydroxyl group on carbon 2:



Disaccharides

Disaccharides comprise two monosaccharides, joined together in a condensation reaction.

- lactose – a fusion of glucose and galactose – found in milk and related products
- maltose – a fusion of two α glucose molecules – used in brewing and forms part of starch
- sucrose – one glucose and one fructose molecule – a sweet sugar which, when added to water, is used in soft drinks, and coffee drinks, on breakfast cereals, and used in baking.

The **condensation reaction** involves the fusion of two OH groups with the expulsion of H_2O and the formation of a chemical bond with oxygen bridging the gap. In the formation of maltose, for instance, the condensation reaction takes place between hydroxyl groups on carbons 1 and 4. This forms a 1,4 glycosidic bond, as shown in the diagram on the right.

condensation reaction – a chemical reaction which joins two molecules together with the expulsion of water

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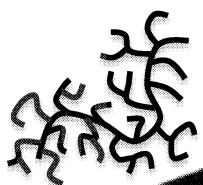


Polysaccharides

Polysaccharides are long chain molecules comprising of many monomers joined together. The monomers involved, and which carbons are linked together, will determine the properties of the polysaccharide.

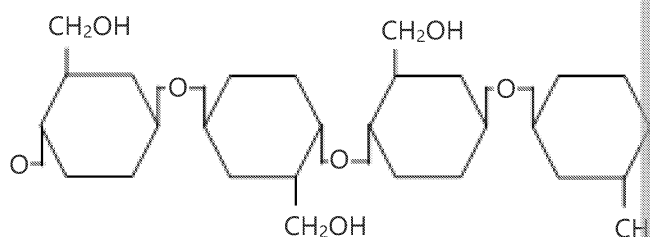
For instance, joining α glucose molecules together with only 1,4 glycosidic bonds forms amylose. This chain twists into a helix because of the angle of the bonds, and because it is insoluble in water, it can stabilise it.

If the same bonding pattern occurs but is occasionally interrupted by 1,6 glycosidic bonds, a polysaccharide called amylopectin is made. Amylopectin and amylose are combined to form starch, which is used as a store of energy.



In animals, the same bonding structures are used, but 1,6 glycosidic bonds are frequently introduced, introducing more branches into the chains, to form glycogen. This is also used for energy storage, but because of the heavily branched structure, it is rapidly digested and the glucose used in respiration to supply energy for animals is more important than for plants.

Cellulose is a linear polysaccharide, found in plant cell walls. Cellulose is formed by β glucose monomers joined one after another. However, for this to work, every other monomer must be a β glucose, so the two OH groups close enough together to react. This creates a long, straight chain of adjacent chains of cellulose, thus creating tough, fibrous tissue that can reinforce the cell wall.



Test your knowledge

- State which monomers form cellulose and describe how they are joined together to form fibrous chains.
- Describe a condensation reaction.
- State the differences between α glucose and β glucose, galactose and fructose.
- Describe how a hexose and a pentose sugar differ and how they are similar.

Function of carbohydrates

Glucose is used by cells in cellular respiration to produce ATP: the currency of energy. For any cellular process that requires energy, such as active transport or muscle contraction, if there is no energy, these processes cannot occur, so a supply of glucose is essential for every cell.

Glucose is stored as starch (in plants) or glycogen (in animals), so carbohydrates can store energy in an organism's body. Storing water-soluble glucose in an insoluble form allows for the movement of water in and out of the cell.

Carbohydrates such as cellulose, may have an important structural role in an organism. Cellulose forms strong, fibrous structures which reinforce plant cell walls and provide a source of fibre, which is important for gut health. Chitin is found in the cell walls of fungi, and the exoskeleton of insects is made using a modified form of carbohydrate which includes nitrogen as well.

Test your knowledge

- State three functions of carbohydrates and give an example of a carbohydrate for each.
- Explain why water insolubility is important in polysaccharides.

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Testing for carbohydrates

Testing for reducing sugars using Benedict's solution

All monosaccharides are **reducing sugars**. It is possible to identify the presence of reducing sugars in a sample by testing it with Benedict's solution. The presence of reducing sugar causes the Benedict's solution to change colour from light blue to green, yellow, orange, or red, or a mixture of the two colours. This is due to the sugars having a free ketone or aldehyde group, which reduces copper ions in the solution from Cu^{2+} ($\text{Cu}_{(\text{II})}$) to Cu^{+} ($\text{Cu}_{(\text{I})}$), changing the colour of the solution relative to the proportion of reducing sugars present.

To test for reducing sugars:

1. Ensure the sample to be tested is in liquid form: grind to a paste with distilled water.
2. Add an equal volume of Benedict's solution to the sample.
3. Heat in a water bath at approximately 90°C for a few minutes.

If the solution changes colour, it indicates an approximate quantity of reducing sugar present. Because this is a qualitative test, it cannot give you an exact figure.

Colour of solution	Approximate concentration of reducing sugar
Blue	None
Green	Very low
Yellow	Low
Orange	Medium
Red	High

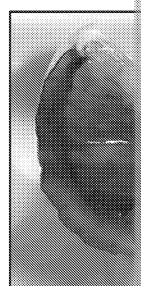
Testing for starch using iodine

Starch can be tested for using iodine. Iodine is an orange-brown solution but turns blue-black in the presence of starch. A few drops on the substance are enough to indicate whether starch is present or not. This photograph of starch-stained courgette shows starch collected in specific areas of the fruit.

Potatoes are stores of carbohydrates and are full of starch:



Even parts of a...



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Test your knowledge

- Name two reducing sugars and describe the test for them.
- State the colour changes and what they indicate for the reducing sugars and starch.



Recap questions: Structure and function of carbohydrates

- 1 Describe how the polymerisation of α glucose forms starch, but β glucose forms cellulose.
- 2 Plant sap transports sugars in the form of sucrose, a disaccharide. What colour change occurs in the presence of this sugar, and why? (3 marks)
- 3 Sucrose from the maple tree sap is collected and boiled in dilute hydrochloric acid to break sucrose into glucose and fructose. What colour will Benedict's solution turn?

B3 Structure and function of proteins

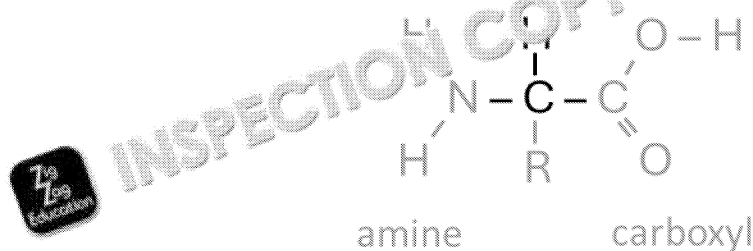


Key points covered

- Structure of proteins, including primary, secondary, tertiary and quaternary structures
- Function of proteins and their functional roles

Structure of proteins

Proteins are made up of amino acids. Amino acids contain a central carbon atom, a hydrogen atom, an amino group, a carboxyl group and an R group which differs for each amino acid.

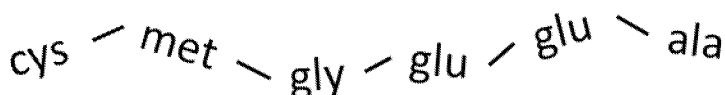


Amino acids are linked together to create long polypeptide chains, which undergo folding before being functional. Each amino acid has particular properties relating to its side chain. Amino groups can accept H⁺ ions (protons) so are basic, whereas carboxyl groups can donate H⁺ ions so are acidic. Non-polar amino acids are hydrophobic, and polar amino acids are hydrophilic, just like water.

Proteins have many functions within the body, broadly grouped in two categories: structural and enzymatic. Many are comprised of several subunits or prosthetic groups which ensure the correct function of the protein is possible.

Primary structure

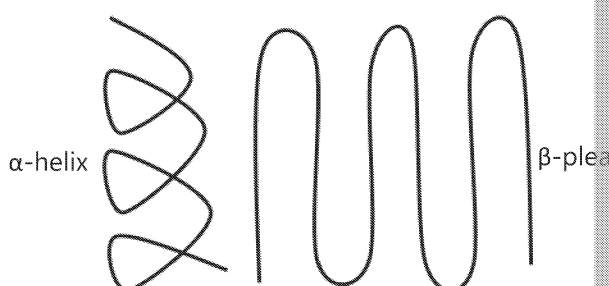
To build a protein, firstly the correct amino acids must be assembled in the right order. This sequence is determined by the gene encoding the protein and involves the translation of the mRNA triplet code into a **polypeptide** chain by ribosomes. Amino acids are bonded together with **peptide bonds** which are formed through condensation reactions, much like carbohydrates.



Secondary structure

The secondary structure is formed with the interaction between the partially H⁺ amino group and the partially O⁻ carboxyl group on a non-adjacent amino acid. Hydrogen bonds are created, creating either α -helices or β -pleated sheets, areas of folding within the protein. As the chain folds, it creates initial structure in the protein which is further reinforced during the tertiary structure.

The folding is dependent on the amino acids contained within the chain. Globular proteins, such as enzymes, are made of α -helices and β -pleated sheets, whereas actin and myosin, found in muscle, are made of long, thin, fibrous proteins.



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Tertiary structure

The tertiary structure involves further bonding between non-adjacent amino acids, interaction between the amino acids and the external environment. There are a number of forms at this level:

- Ionic bonding occurs with the loss or gain of electrons between differently charged amino acids, forming bonds that are very stable.
- Disulfide bridges/bonds form strong covalent bonds between two nearby cysteine (or methionine) amino acids, because they are the only amino acids to contain sulphur, which offers great stability to the protein.
- Hydrogen bonds are weakly interactive and occur between different R groups, stabilising the secondary structure.
- Van der Waals forces are weak forces between atoms due to differences in the electronegativity of atoms, which help to provide further stability to the structure of the protein.
- Hydrophobic interactions bring hydrophobic amino acids into the centre away from the water.

Quaternary structure

Some proteins are made of more than one polypeptide strand. The formation of the quaternary structure is the combination of multiple strands into one protein and may also involve the addition of non-protein components, such as iron.

Proteins which are made only of two or more polypeptide strands are called **non-conjugated** proteins. These consist only of protein, and include insulin, which is a globular protein, and collagen, which is a fibrous protein.

Other proteins contain multiple polypeptide strands and other non-protein components, such as ions, and are called **conjugated** proteins. Haemoglobin is a complex globular conjugated protein made of four separate polypeptide strands, where each strand also contains a haem group – an Fe^{2+} ion – which binds oxygen.

Proteins can be functional or structural in nature, and globular (spherical) or fibrous. Common proteins include:

- Functional:
 - haemoglobin – carries oxygen in red blood cells
 - insulin – hormone that regulates blood sugar levels
 - lipase – an enzyme that digests lipids
- Structural:
 - keratin – builds hair, nails, claws
 - collagen – builds strong connective tissue such as ligaments and tendons

Test your knowledge

- Describe the key features of primary, secondary, tertiary and quaternary protein structure.
- Name an example of a fibrous non-conjugated protein, and a globular conjugated protein.

Function of proteins

Given that DNA encodes for the proteins used in our body, it should be unsurprising that proteins have a wide variety of shapes and functions.

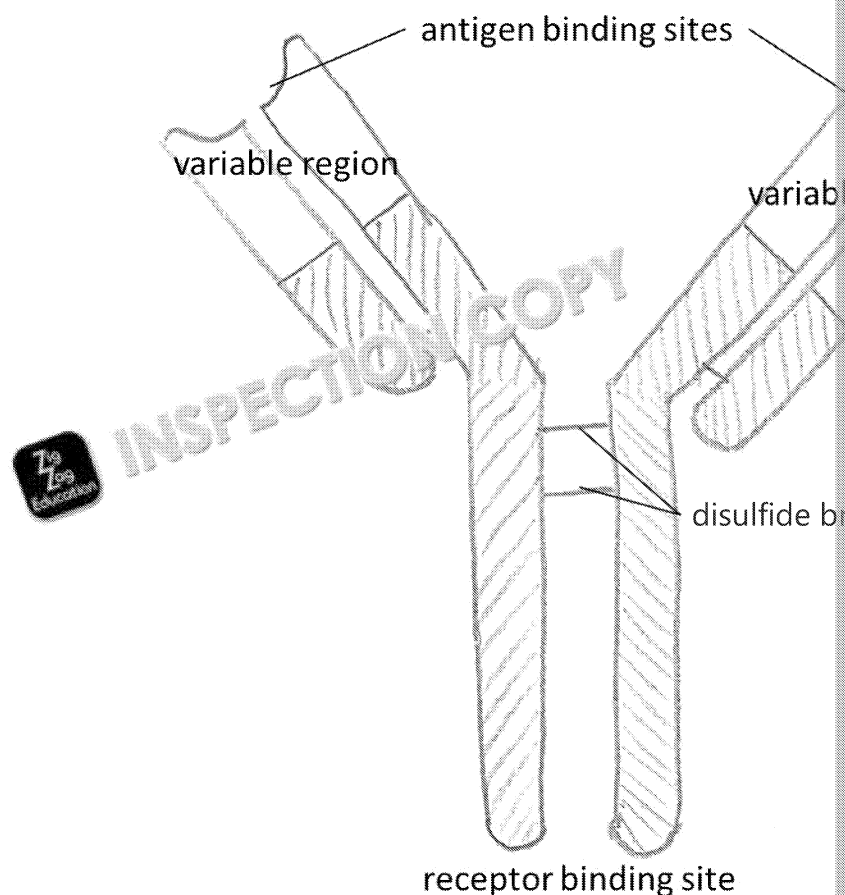
Muscles are made of two significant proteins called actin and myosin. They interact to produce movement by the actin filament sliding over the myosin filament. This shortens the total length, contracting the muscle. The space between the two actin fibres narrows during contraction.

All biological enzymes are proteins. Enzymes catalyse chemical reactions, significantly speeding up the rate of reaction. They are fundamental to the function and survival of organisms and are responsible for the metabolism (breaking down) of molecules. Common examples include digestive enzymes (digests starch to maltose) and trypsin (digests some proteins into amino acids); and cytochrome c (converts hydrogen peroxide into water and oxygen at the end of the electron transport chain reactions).

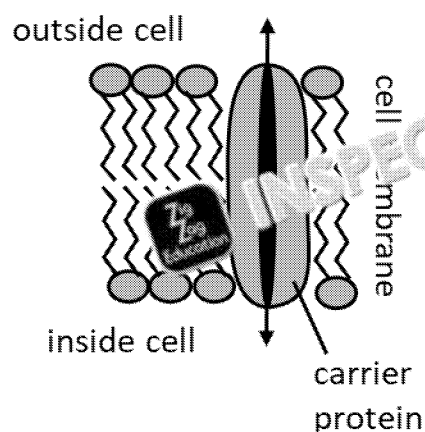
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Antibodies, produced by special leucocytes called plasma cells, are proteins. They recognise a pathogen as an invader which needs to be destroyed, and for grouping pathogens together so they are able to move around the body. They are Y-shaped proteins with various sections to the molecule. The region which differs depending on the pathogen.



Antibodies bind antigens. Antigens are molecules that protrude (stick out) from the surface of the cell, identifying the cell and its function. These are specific to the organism and can differ between individuals in the species: the antigens on your cells are specific to your cells only, for instance. This is useful to the immune system because it helps to identify cells that are non-self (not from your body). These cells should not be in your body and therefore need to be destroyed. These antigens are used by leucocytes to identify pathogenic cells for destruction.



Other proteins go through the cell membrane with the inside of the cell. Carrier proteins move molecules against the concentration gradient. They bind a molecule on the outside (cellular form of stored energy) to change its shape and move the molecule through the cell membrane into the cell. This is how glucose is brought into a plant's root using carrier proteins across the membrane in a neurone.

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Proteins can also be hormones, such as insulin. Hormones are chemical messengers that stream from the gland that they are released from (with insulin, this is the pancreas, not the liver) where an effect is triggered. This allows a coordinated response to happen in a change in environment.

Some proteins, such as haemoglobin, are responsible for transporting other molecules. Haemoglobin binds oxygen in the lungs where it is at high concentration, and releases it where oxygen is at a lower concentration. The presence of iron ions enables this function. The function is demonstrated because the part of the gene coding for the haemoglobin protein has been highly conserved for millions of years.

Proteins are essential in the process of tissue growth and repair. There are many proteins involved in the process of cell mitosis (division of body cells), and the general maintenance of cells.

Proteins are also essential in blood clotting and provide a number of blood clotting factors. Fibrinogen forms a cascade or chain reaction that rapidly turns dissolved fibrinogen into insoluble fibrin for plugging the holes in the blood vessel wall. Fibrin also traps platelets and blood cells to form a clot, which can create a scab, under which new skin cells can grow and repair the wound.

Test your knowledge

- State three functions of proteins and give three examples of proteins with their functions.
- State why proteins are needed to transport molecules into a cell and what factors affect this.
- Describe how antibodies and antigens interact.

Testing for proteins

It is possible to test for the presence of proteins within a solution. A light blue reagent called Biuret's solution is used, which turns purple in the presence of protein. Biuret solution includes sodium hydroxide solution. In this test, the copper ions in the alkaline solution (sodium hydroxide) form a complex with the peptide bonds in the protein, causing the change in colour.

To test for proteins:

1. Add an equal volume of sodium hydroxide solution to the sample solution.
2. Add copper sulfate solution a few drops at a time to the solution until it turns blue.
3. Mix the solution and leave for a few minutes.
4. If protein is present, the solution will turn purple.

Test your knowledge

- Name the two solutions which make up Biuret's solution.
- State the colour change expected when testing with Biuret's solution if protein is present.

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Recap question: Structure and function of proteins

1. Describe the different bonding interactions occurring to hold the tertiary structure of a protein. (5 marks)
2. Catalase contains four iron(II) ions. Describe its structure and function. (4 marks)



B4 Structure and function of nucleic acids



Key points covered

- Structure of nucleic acids, including DNA and RNA
- Function of nucleic acids

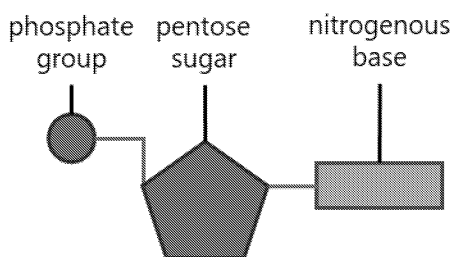
Structure of nucleic acids

Nucleic acids are polymers made from repeated nucleotide monomers. The nucleic acids differ in the minor differences leading to the four different DNA bases and four RNA bases. They are important because they encode information: instructions which allow the processing of DNA into proteins, which are uniquely designed for their specific cellular roles. Changes to the order of the bases can lead to a different protein being produced, which is usually, although not always, harmful.

Nucleotide structure

The nucleotide is the monomer which, when combined with many others, forms the nucleic acid. There are three main parts to the nucleotide:

- the phosphate group (PO_4^{2-})
- the pentose sugar group (deoxyribose in DNA, and ribose in RNA) with five carbon atoms
- the nitrogenous base (guanine, cytosine, adenine, thymine or uracil)
 - in DNA, the bases C, G, A and T are present, and pair G with C and A with T
 - in RNA, the base T is replaced by U

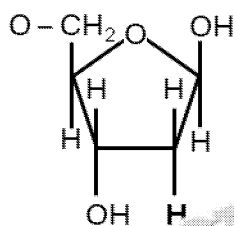


The phosphate group is connected to the carbon atom at position 5 of the pentose sugar.

The carbon atoms are labelled 1 to 5 anticlockwise around the pentose sugar, starting from the carbon atom at the top right (see diagrams on next page).

The sugar in DNA is deoxyribose: it is without an oxygen (deoxy) off the second carbon, highlighted in bold underlined.

The sugar in RNA is ribose: it has an oxygen (ribo) off the second carbon, highlighted in bold underlined.



The five bases – G, C, A, T and U – are grouped into purine and pyrimidine forms.

- Pyrimidines (the longer of the two words) are the single-ring (shorter) nitrogenous bases: cytosine, thymine and uracil
- Purines (the shorter of the two words) are the double-ring (longer) nitrogenous bases: guanine and adenine

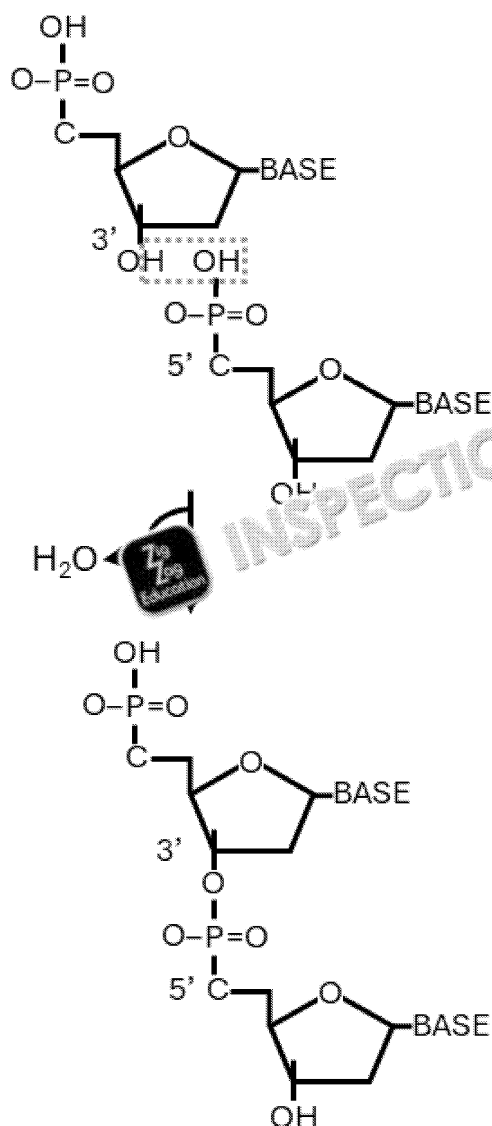
When base pairing occurs in the DNA double helix and the ribosome (during polypeptide synthesis or to form the structure of tRNA for instance), a purine base pairs with a pyrimidine base to keep DNA of equal width throughout: if two purines were to pair together, the DNA would be wider than normal; if two pyrimidines paired together, the DNA strand would be narrower than normal. The width is governed by the number of hydrogen bonds formed between the pair of bases. G and C form three hydrogen bonds when paired together; A and T (or U in RNA) form two hydrogen bonds. Hydrogen bonds are weakly interacting but provide sufficient stability to the structure.

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Polynucleotide structure



Nucleotides can be joined together to form DNA or RNA. Phosphodiester bonds join nucleotides in a condensation reaction, which can be undone with a hydrolysis reaction. The bond is formed between the phosphate group attached to the 5' carbon of one pentose ring and the hydroxyl group attached to the 3' carbon of another pentose ring (3').

The staggered nature of this bond is seen in DNA.

DNA

DNA is double-stranded. The sense strand is orientated 5'–3' and contains the code for protein. The antisense strand is orientated 3'–5'. They are **antiparallel**.

Complementary base pairing binds the two strands of the double helix together. Guanine pairs with cytosine with three hydrogen bonds; adenine pairs with thymine with two hydrogen bonds.

RNA

There are three forms of RNA: messenger (mRNA), ribosomal (rRNA) and transfer (tRNA). These are made of nucleotides containing the bases G, A, U, and C. Complementary base pairing occurs between the bases of the two strands.

Test your knowledge

- State which bases complementary pair with G and T and describe how hydrogen bonding influences these pairings.
- Name the two forms of bases and describe how they differ.
- State the name of the bond that joins one nucleotide to another, and describe how its formation occurs.

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Function of nucleotides

DNA

Sections of DNA called genes encode for proteins. The order of the DNA bases determines the order of the amino acids in a protein. Any changes to the order of the bases can lead to a loss of information called a mutation.

Different versions of genes are called alleles, and contain slight differences in the base sequence. This leads to the expression of slightly different proteins, resulting in changes to the phenotype.

RNA

RNA has different functions depending on the type of RNA.

- mRNA is responsible for transferring the DNA code of a gene out of the nucleus to the site of protein synthesis.
- rRNA forms part of the ribosome which translates the mRNA code into a polypeptide chain.
- tRNA is responsible for binding the correct amino acid to the growing polypeptide chain.

In more detail, the process of protein synthesis is:

- **Transcription:**
 1. The section of DNA containing the gene to be transcribed is 'unzipped': the two strands are broken by DNA helicase and the strand untwists.
 2. The antisense strand is copied into mRNA using complementary base pairing. The mRNA strand contains the same sequence of bases as the sense strand of DNA. Adenine, guanine and cytosine bases of ribose nucleic acids are used, and thymine is swapped for uracil.
 3. The enzyme RNA polymerase joins the sequence of mRNA bases together using condensation reactions to form phosphodiester bonds.
 4. The mRNA strand detaches from the DNA strand and leaves the nucleus.
- **Translation:**
 5. The mRNA strand joins to a ribosome (containing rRNA and other enzymes) in the rough endoplasmic reticulum (RER) membrane.
 6. The mRNA strand is read by the ribosome three codons (bases) at a time. A complementary tRNA which has the corresponding amino acid bound to it joins to the mRNA.
 7. The enzyme peptidyl transferase binds each amino acid to the next with a peptide bond, forming a growing polypeptide chain.
 8. Once the stop codon is reached, no further amino acids are added to the chain. The polypeptide chain, tRNA and mRNA are released from the ribosome.
 9. The polypeptide chain is further processed into the final protein. This includes folding, glycosylation and conjugation, before it is ready for use within or outside the cell.

Test your knowledge

- State the function of DNA and m, r and tRNA.
- Outline the process of transcription and translation of DNA into protein.

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Recap questions: Structure and function of nucleic acids

1. Explain how complementary base pairing works and why this is necessary for the structure of DNA. (5 marks)
2. Calculate what percentage of the other bases will be recorded if a sample has 20% adenine.
3. Convert the following DNA antisense strand into its corresponding mRNA sequence.
G T A A G C C T A G A T T C T G C A



B5 Structure and function of lipids



Key points covered

- Structure of fats, oils and waxes
- Esterification of glycerides
- Testing for lipids
- Function of lipids in cell membranes and homeostasis

Structure of lipids

Lipids include fats, oils, waxes and steroids. Fats are solid and oils are liquid at room temperature. They are made of carbon, hydrogen and oxygen but unlike carbohydrates, they are non-polar molecules. The atoms are evenly distributed, so there is no area which is negatively charged or makes them not water-soluble (because water is a polar molecule: O^- and H^+). The elements: phospholipids contain a phosphate group, for instance.

Lipids are highly complex large molecules which are not made from repeating monomers. They are made from a few similar features, with three main families:

- Triglycerides contain one glycerol and three fatty acids.
- Phospholipids contain one glycerol, two fatty acids and a phosphate group.
- Fatty acids can be saturated or unsaturated (see next).

Saturated and unsaturated fats

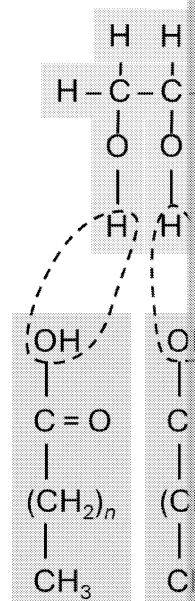
Fatty acids are long carbon chains with hydrogens bonded. Saturation describes whether all the carbons are fully loaded with hydrogen atoms, or whether some hydrogens are missing, allowing carbon to form a double bond.

Saturated fats have no carbon-carbon double bonds and so are fully saturated with hydrogen. This creates straight chains with no kinks, which means the molecules can pack close to each other, making them solid at room temperature: fats.

Unsaturated fatty acid chains have some hydrogen atoms 'missing', meaning that there are carbon-carbon double bonds in the chains. This forms kinks in the chains, so the molecules cannot pack so closely together. This means they are liquid at room temperature: oils. Monounsaturated fats have only one carbon-carbon double bond. Polyunsaturated fats have two or more carbon-carbon double bonds.

Esterification

Diglycerides and triglycerides are formed when a glycerol bonds to two (di-) or three (tri-) fatty acid chains, in a condensation reaction (release of water) called esterification. One water molecule is released for each fatty acid chain bound to glycerol. The lengths of the fatty acid chains can vary significantly.



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Test your knowledge

- Define the term 'saturated' and describe the difference between saturated and unsaturated fatty acids.
- Explain why fats are solids and oils are liquids at room temperature, with reference to their fatty acid chains.
- Describe the process of esterification.
- State the components of a phospholipid and a triglyceride.

Function of lipids

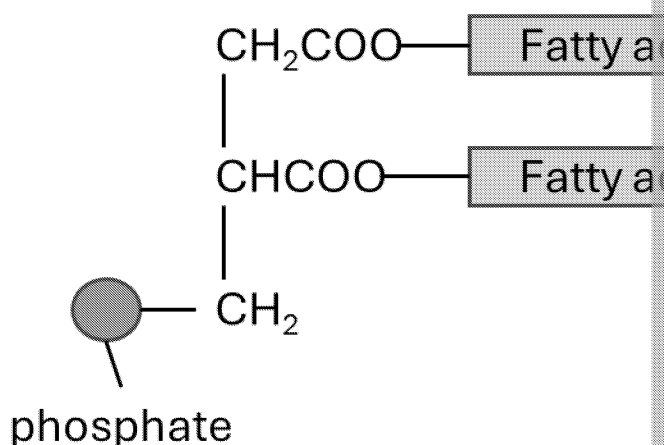
Lipids have many functions:

- they can be used as a source of energy or for storing energy;
- they can provide insulation and protection of organs;
- they are used to create cellular membranes;
- they are used as hormones.

Lipids are good sources of energy because the density of chemical bonds means more energy is stored per gram compared to carbohydrates. Therefore, consuming fats provides a long-term source of energy. As fats can be stored in adipose tissue, they provide a long-term source of energy, too. When a long-term source of energy is needed for work coming up, like completing long migration routes or surviving in poor food availability.

Fats are dense because of their saturated chemical bonds. Lipids are insulators because they can't transfer energy, which means they can be used to provide insulation and protection. For example, animals such as whales and seals are made of thick layers of fat to provide insulation. Around all vital organs there is a layer of fat which provides some protection against damage.

Phospholipids comprise a glycerol molecule with two fatty acid chains and a phosphate group.



This structure results in a hydrophilic (water-loving) head (the glycerol and phosphate section) and a hydrophobic (water-hating) tail (the fatty acid chains). They are amphipathic (hydrophobic and hydrophilic), and the polar nature of a phospholipid means it orientates itself with the hydrophobic tails facing away from water, such as the cytoplasm. When multiple phospholipids are aligned alongside each other, and, when one line of phospholipids is lined up opposite another line facing in the opposite direction, so the hydrophobic tails are facing inwards, they form a bilayer: a membrane.

All cell membranes are formed in this manner, such as around mitochondria, chloroplasts and the tonoplast of vacuoles. In addition, a single layer of phospholipids can form a capsule or a sphere in cytoplasm, trapping hydrophobic contents inside. This is useful for cellular transport. Or, the tails can stick out with the hydrophilic heads in the water, such as in the lungs, thus forming a surfactant, which helps to keep the alveoli open and not stuck together.

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Lipids can also act as hormones when in the form of steroids. All steroids have the same basic shape:

- four hydrophobic carbon rings;
- a variable end;
- a hydrophilic hydroxyl group at one end.

They include testosterone, oestradiol and cholesterol, and can pass through the phospholipid cellular membrane because they are non-polar.

HO

Test your knowledge

- Explain why lipids are a good source of energy, and in which cells lipids are stored.
- Describe the shape of a phospholipid and explain why this is useful for building cell membranes.
- Name two steroid hormones.

Testing for lipids

Lipids can be tested for using the emulsion test. As a liquid, mix the sample with ethanol. This is best done in a test tube with a bung! If a lipid is present, it will form a white emulsion. The solution will be cloudy and opaque. If left undisturbed for a few minutes, the mixture will separate into a layer on the top of the ethanol as they are less dense.

Test your knowledge

- Name and describe the test for lipids.



Recap questions: Structure and function of lipids

- 1 A student tests three substances for the presence of lipids. Sample A turns ethanol cloudy after about 7 minutes; sample B turns ethanol cloudy but no separation occurs; sample C remains ethanol colourless and unchanged. Which sample contains a lipid? Explain your answer. (3 marks)
- 2 Compare and contrast a triglyceride and a phospholipid. (7 marks)

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C: Cellular transport and enzy

C1 Cell transport mechanisms

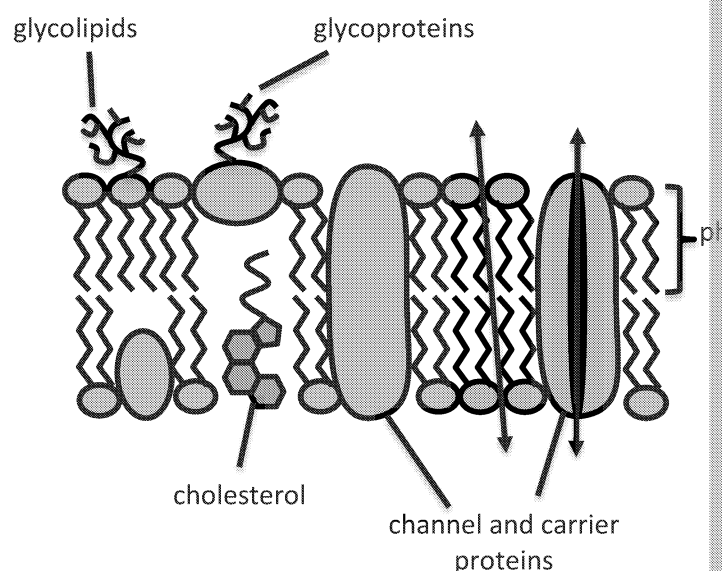


Key points covered

- The structure of the cell surface membrane, including the fluid mosaic model
- Endocytosis and exocytosis
- How molecules are transported, including by diffusion, facilitated diffusion, osmosis and active transport
- Surface area, volume and the effects of size on living organisms

Structure of the cell surface membrane

The cell surface membrane, like most membranes within living organisms (especially eukaryotic ones), is largely made from phospholipids. The amphipathic nature of phospholipids means that they form as a double layer, hydrophobic tails pointing inwards away from the water-based cytoplasm, and hydrophilic heads forming the membrane's inner and outer surface.



Embedded within the phospholipid membrane are many other components:

- channel proteins, which do not change shape, and carrier proteins, which do change shape to move substances across the membrane; these would otherwise be unable to pass through the hydrophobic middle of the membrane;
- aquaporins, a specialist form of carrier protein, transport water;
- cholesterol, a steroid, adjusts the fluidity of the membrane and allows adjustment to temperature changes;
- glycoproteins (proteins with carbohydrate chains attached) and glycolipids (phospholipids with carbohydrate chains attached) allow cell-cell communication, cell recognition and adhesion.

This is known as the fluid mosaic model and describes how cell membranes are assembled. This formation provides a flexible barrier which can be adapted to suit the specific functions of the cell. It can separate parts of a cell, such as the mitochondrial space from the cytoplasm, to allow different functions to take place within uniquely controlled microenvironments. The membrane also forms vesicles (see bulk transport, later) which enables bulk transportation of substances. The phospholipid nature of the membrane means that many substances are naturally prevented from moving through it, either because they are too large to pass through the tightly packed membrane, or because they are too polar to pass through the hydrophobic middle. This enables the cell to control the entry and exit of substances.

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The elements embedded within the cell membrane may have one of many functions, including:

- enabling cell-to-cell recognition and communication, including identifying the cell as self or foreign, and the binding of hormones;
- adhesion between cells to form tissues, including by forming a **glycocalyx**, gap or tight junctions and **desmosomes**;
- channels between cells, such as **plasmodesmata** in plant cells, to enable long-range transportation of substances such as water, ions and sugars;
- providing channels to facilitate the movement of substances by diffusion, osmosis and active transport, including ions, carbohydrates and proteins.

Test your knowledge

- Describe the fluid mosaic model
- State three benefits of this model to cells.
- List four functions that may be carried out by components of the cell membrane

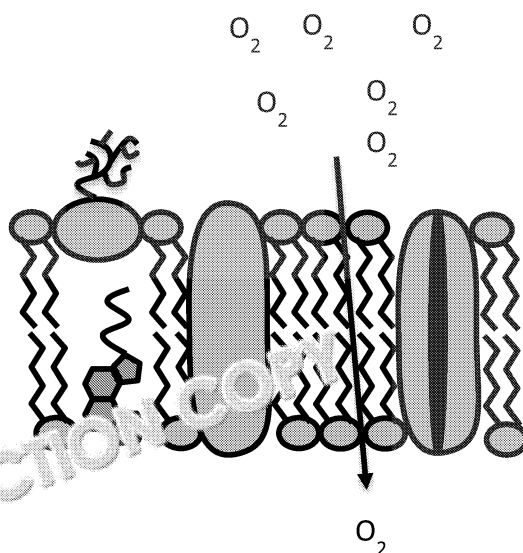
Transporting molecules

Some molecules can freely pass through the phospholipid membrane; others cannot and are required in higher concentrations inside the cell than they are present outside the cell. These molecules require specialist transport.

Diffusion and osmosis

Diffusion is the net movement of particles from an area of higher concentration to an area of lower concentration, therefore down the concentration gradient. It does not require energy, and the molecules can move directly through a membrane, but if they do, they will do so freely. Particle movement is random, but the overall direction of movement is down the concentration gradient.

The diagram below shows diffusion of oxygen through the cell membrane.



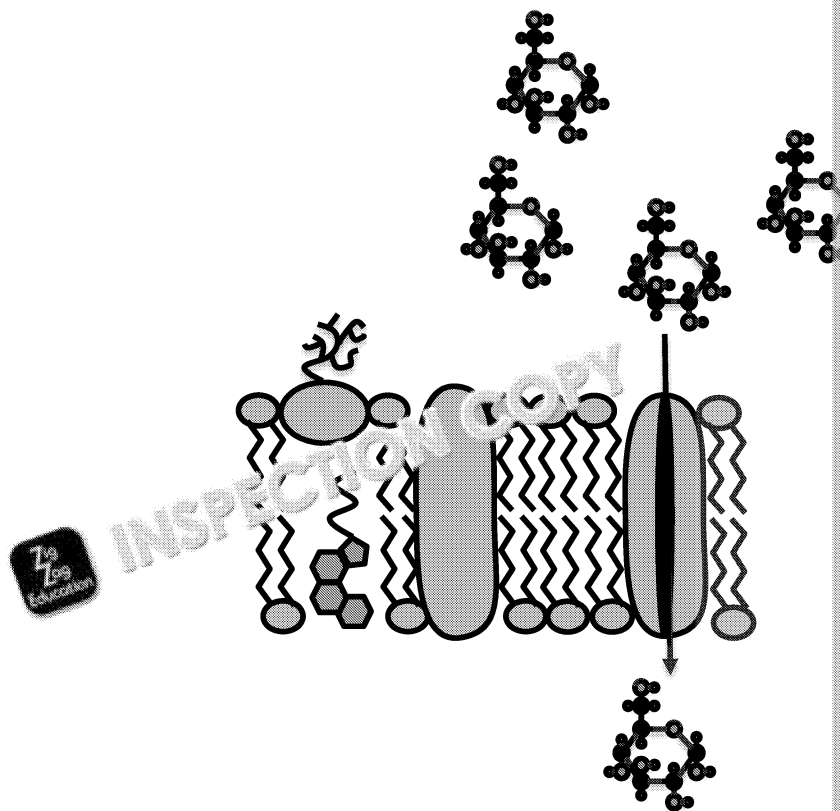
Some molecules can diffuse through the cell membrane: oxygen and carbon dioxide are small, uncharged molecules that can pass unhindered through the membrane. This is the simplest form of transport.

Facilitated diffusion is the next step up; as with diffusion, particles diffuse down the concentration gradient, but they move through a protein channel instead of the phospholipid bilayer. Some ions and large molecules cannot pass through the hydrophobic middle of the membrane. Channel proteins and carrier proteins do, and both provide selective permeability.

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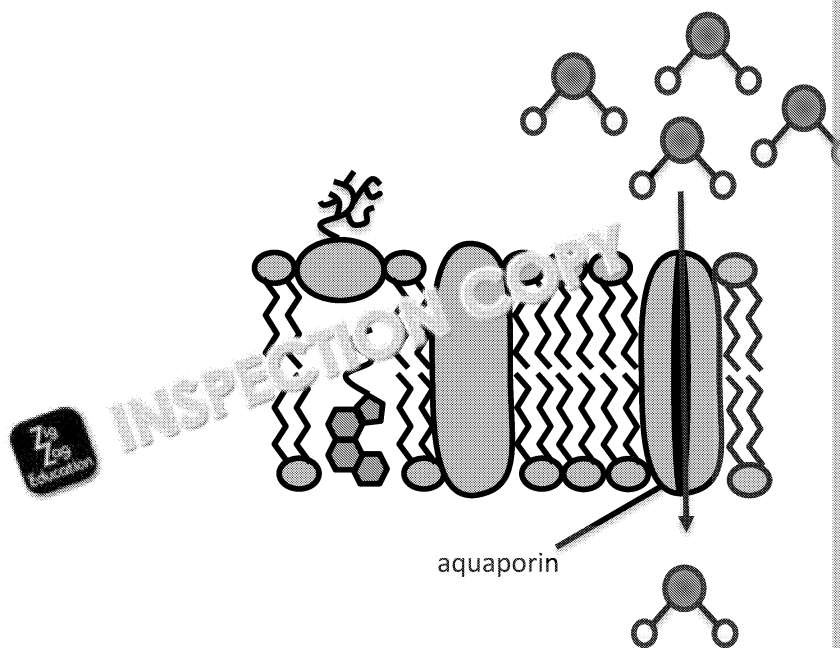


The simplified diagram below shows glucose being moved through the cell membrane



Osmosis is a specialised form of diffusion: it is the net movement of water particles from an area of higher water potential to an area of lower water potential, therefore down the water potential gradient, through a partially permeable membrane. In order to facilitate the movement of the polarised water molecules through the hydrophobic bilayer, the water molecules pass through aquaporins: integral membrane proteins (spanning the whole membrane).

The diagram below shows water passing through an aquaporin, an integral membrane protein.



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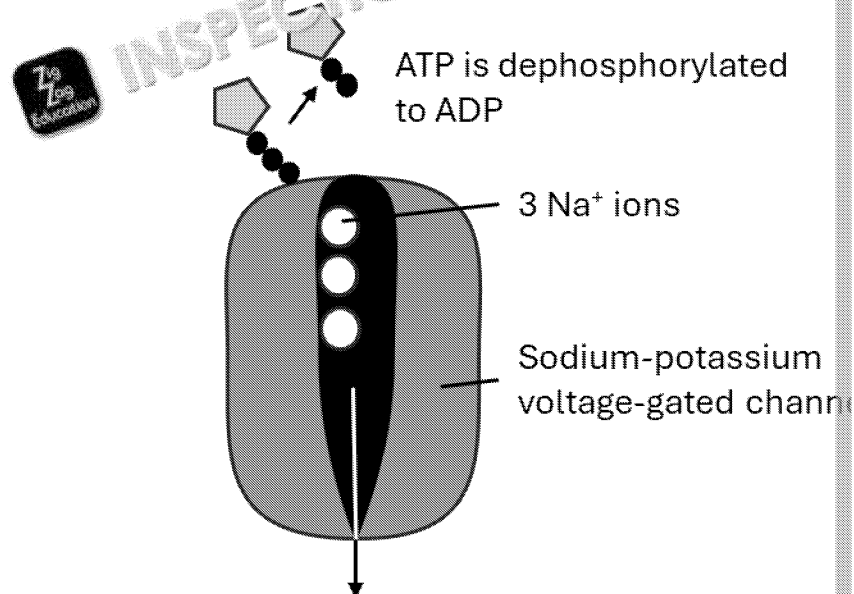


Active transport

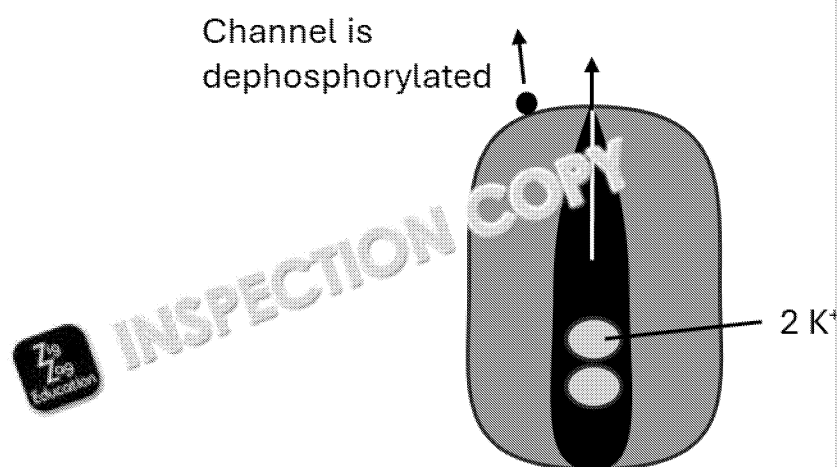
Active transport is the net movement of particles against the concentration gradient and requires ATP.

Indirect active transport requires the passive movement of one substance to enable the movement of another. Examples include the movement of glucose from the small intestine into the blood and the kidney nephron, using a sodium-dependent glucose cotransporter. Sodium ions move down their concentration gradient, and ATP is used to transport glucose against the concentration gradient. Both substances are transported in the same direction, from outside to inside the cell, at the same time by a carrier protein.

Gated channel proteins are another example of active transport. Sodium-potassium pumps are found in neurone axons and are responsible for generating membrane potential. The pump uses ATP. The molecule binds and, on dephosphorylation of ATP, the Na^+ ions are moved through the channel.



Next, two K^+ ions bind to the inside of the channel, causing the dephosphorylation of ATP. This causes it to change shape and transport the K^+ ions through the membrane and out of the cell.



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Bulk transport

When many molecules need to be transported at once, or the substances to be moved through membrane proteins, bulk transport is required. Transport into the cell is called endocytosis. Transport out of the cell is called exocytosis. Bulk transport occurs when a part of the phospholipid bilayer breaks off into a vesicle.

Endocytosis involves the **invagination** of the main cell membrane around a substance external to the cell, thus capturing it and bringing it into the cell. Phagocytosis (used to capture and destroy invading pathogens) and pinocytosis (used to transport liquids into the cell) are examples of this process.

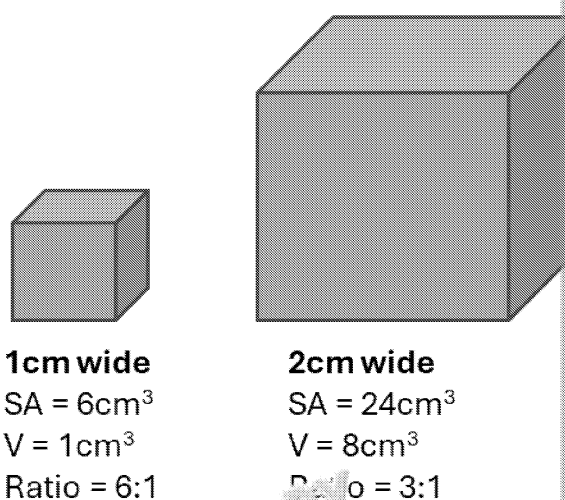
Exocytosis involves converting part of an organelle, such as the Golgi body or the rough endoplasmic reticulum, into a vesicle. This is used when transporting carbohydrates, lipids and proteins for use in digestion or with antibodies.

Test your knowledge

- State the difference between phagocytosis and pinocytosis.
- Describe the differences between diffusion, osmosis and facilitated diffusion.
- Explain why bulk transport may be used instead of active transport.

Surface-area-to-volume ratio

The surface-area-to-volume ratio describes how the surface area of an organism relates to its volume. Generally, as an organism's size increases, the surface-area-to-volume ratio decreases. This means that the area over which substances can diffuse to supply the growing volume is reduced.



Surface area and its relationship to the volume of matter contained within the surface area is important for diffusion: the surface area limits how much diffusion can take place per second – the larger the surface area, the more diffusion can occur. However, the distance that the substances have to travel once they have got over the initial surface area is also important. The longer this will take, so the larger the cell is, the less effective the diffusion of nutrients in and waste products out at a rate throughout the whole cell that allows for growth.

To get around this, organisms have instead evolved to split one large cell into lots of smaller cells. Each cell has specialised functions within the organism's body, and create systems such as the circulatory system that enable substances (such as oxygen and carbon dioxide) to be exchanged at one end (e.g. the lungs), and then transported around the body (by the circulatory system) to where they are needed (e.g. the other cells). Each cell in a multicellular organism's body has a highly specialised function, and each cell is specialised to work as effectively as possible. Some of these specialised cells were explained in the previous section.

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Understanding the impact of surface-area-to-volume ratio on aspects such as cell size is fundamental in biology. It underpins many of the unique adaptations seen in many organisms' bodies.

Calculating surface-area-to-volume ratio is usually done on the assumption that the organism is cuboid in shape. However, this is obviously not true of any organism, and most organisms have adaptations which maximise surface-area-to-volume ratio. In the example of the gills of fish, the surface area of the lungs of humans is vast: if the lungs were spread out to their maximum, they would cover approximately half a tennis court! However, this is bundled up within the body into a complex network of branching tubes, thereby creating a vast surface-area-to-volume ratio. Plants have similar adaptations. The surface area of a leaf is maximised to ensure light collection and gas exchange for photosynthesis. The roots of plants have a substantial surface area created by the protrusions of the root hairs, increasing the surface area by approximately 23 times!

Let's look at that in more detail:

If a root hair cell is a 15 μm cube, then:

$$\begin{aligned} \text{surface area} &= \text{width} \times \text{height} \times \text{number of faces} \\ &= 15 \mu\text{m} \times 15 \mu\text{m} \times 6 \\ &= 1350 \mu\text{m}^2 \end{aligned}$$

$$\begin{aligned} \text{volume} &= \text{width} \times \text{height} \times \text{depth} \\ &= 15 \mu\text{m} \times 15 \mu\text{m} \times 15 \mu\text{m} \\ &= 3375 \mu\text{m}^3 \end{aligned}$$

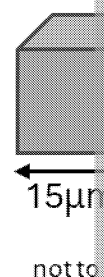
So, its surface-area-to-volume ratio is 1350 : 3375 or 1 : 2.5 (0.4 : 1)

However, the root hair cell has a hair-like protrusion. Let's say that measures 1500 μm long and 5 μm wide. So, the surface area and volume of the protrusion are:

$$\begin{aligned} \text{surface area} &= \text{width} \times \text{height} \times \text{number of faces} \\ &= 1500 \mu\text{m} \times 5 \mu\text{m} \times 4 \text{ faces (we will ignore the small surfaces at each end, as they are accounted for in the main cube)} \\ &= 30\,000 \mu\text{m}^2 + 1350 \mu\text{m}^2 = 31\,350 \mu\text{m}^2 \end{aligned}$$

$$\begin{aligned} \text{volume} &= \text{width} \times \text{height} \times \text{depth} \\ &= 1500 \mu\text{m} \times 5 \mu\text{m} \times 5 \mu\text{m} \\ &= 37\,500 \mu\text{m}^3 + 3375 \mu\text{m}^3 = 40\,875 \mu\text{m}^3 \end{aligned}$$

So, the surface-area-to-volume ratio of the root hair cell with protrusion is 31 350 : 40 875 or 0.77 : 1. This is nearly twice that without the protrusion.



Test your knowledge

- State the equations for calculating surface area and volume.
- Describe why surface-area-to-volume ratio is such an important factor when considering an organism's body.

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Recall questions: Cell transport mechanisms

- Calculate the total surface-area-to-volume ratio of a cube-shaped organism. Suggest adaptations that may be present to increase this ratio further. (5 marks)
- Suggest why root hair cells have a large number of mitochondria and many in their cell membrane. (3 marks)
- Mitochondria, chloroplasts, vacuoles, the nucleus, and endoplasmic reticulum are membrane-bound organelles within eukaryotic cells. Explain why organelles need a membrane. (3 marks)

C2 Enzymes as biological catalysts



Key points covered

- Enzyme structure and function
- Models describing enzyme function, including collision theory and lock and key theory
- Calculating rate of reaction
- Factors that affect reaction rate, including pH, temperature and concentration

Enzyme structure

Enzymes are biological **catalysts**: they speed up the rate of a reaction, perhaps by several million times. If it weren't for enzymes, it is unlikely that life on Earth would exist, as the rates of reaction would be so slow we would not be able to survive and evolve.

Enzymes are globular proteins, folded into a specific tertiary (3D) structure which contains an active site where the substrate binds, and the reaction is catalysed. Some enzymes have conjugates, known as cofactors, which activate the enzyme when bound, and leave the enzyme in its inactive, apoenzyme, state when not bound. This is useful for controlling enzyme function, especially if the chemical reaction it is catalysing must only be conducted on molecules outside of the cell, such as in digestion.

Test your knowledge

- Describe the structure of an enzyme.
- Describe how cofactors control enzyme function.

Function of enzymes

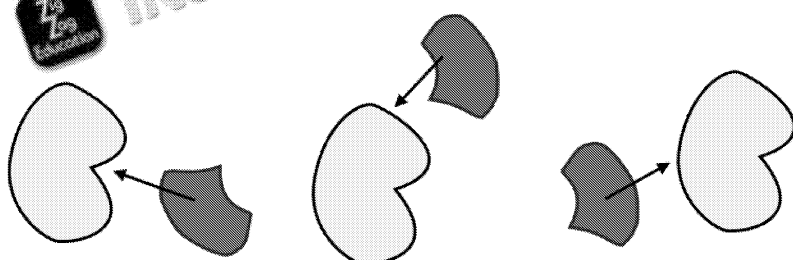
Enzymes catalyse chemical reactions by bringing together substrates to be **anabolism** or enabling the **catabolism** of one substrate into two or more smaller products. They are not used up in the process and remain available to catalyse further reactions.

Enzymes are highly specific for their substrate(s). The active site is shaped such that only one substance can be rightfully catalysed by the enzyme, although similarly shaped molecules can act as inhibitors by temporarily or permanently blocking the active site and preventing the actual reaction from taking place.

Several theories exist to explain how enzymes function, including collision theory and the lock and key model.

Collision theory

Collision theory explains that in order for a reaction to take place, the enzyme and substrate must collide successfully. This means that the substrate(s) must enter the active site of the enzyme. The higher the number of successful collisions, the higher the rate of reaction.



None of these collisions will result in a successfully catalysed reaction.

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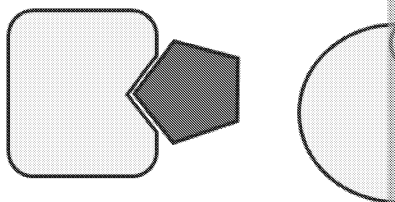


Lock and key theory

The lock and key theory states that the shape of the substrate and the shape of the specific and complementary to each other. Just like a lock and its key, only one substrate fits into the active site of an enzyme to form the enzyme–substrate complex. This requires both the correct combination, plus a successful collision.

Be clear though: it is a complementary fit; the enzyme and substrate are **not** the same shape.

All of these enzymes and their substrates are specific and complementary to each other.



Measuring rates of reaction

The rate of reaction is a measure of how quickly the enzyme is converting substrate. The initial rate of reaction is the fastest, as the greatest number of substrates exist in solution, and the greatest chance that the enzyme and substrate will collide successfully, rather than the enzyme colliding with a product. This is why, provided that all other conditions remain the same throughout the experiment, the initial rate of reaction is the fastest and gives the best indication of the overall rate of reaction.

Test your knowledge

- Describe the structure of an enzyme.
- Write a sentence with fewer than 10 words which summarises the collision theory.
- Convert the stages of the lock and key theory of enzyme function into a flowchart.
- Explain why the initial rate of enzyme reaction will be fastest.

Factors affecting enzyme activity

The rate of an enzyme-catalysed reaction is at its fastest when all conditions are optimal: temperature, pH, enzyme concentration and substrate concentration.

Effect of temperature

Temperature represents how much kinetic energy is available in a system.

At low temperatures, kinetic energy is low, so particles move slowly.

Their low speed and low energy collisions mean that reaction rate is low.

As temperature increases, kinetic energy increases. This increases both the speed that the particles move at, and the energy with which they collide. Therefore, as temperature increases, rate increases.

An increase in rate continues up until the optimum temperature, when rate reaches its maximum. After this, an increase in temperature causes a decrease in rate as the enzyme denatures. Essentially, kinetic energy continues to increase but, while this makes the particles move faster and collide with more energy, the particles in the enzyme itself vibrate faster which, at temperatures beyond the optimum, destabilises the shape of the enzyme, leading to denaturing.

At temperatures just above the optimum, denaturing is not permanent and can be reversed. However, when the temperature moves significantly beyond the optimum, the enzyme is permanently denatured and this cannot be reversed.

The optimum temperature and the degree to which an enzyme can withstand above and below this optimum varies between different species and variants of enzymes. For instance, enzymes from organisms that live in very cold conditions have low optimum temperatures and cannot withstand high temperatures above the optimum. However, organisms which live in very hot conditions have enzymes with high optimum temperatures and are usually much more thermostable. Amylase, responsible for the breakdown of starch, is very thermostable and can function well at temperatures at least 20 °C beyond its optimum.

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Effect of pH

The pH of a substrate measures the number of H^+ ions present within the solution. The more H^+ ions, the lower the pH and the more acidic the solution is.

Enzymes work optimally in a narrow pH range, dependent on the conditions. Enzymes in the stomach are optimised for pH 0–2 (enzyme 1 on the graph), whereas enzymes that work in the small intestine are optimally in pH 8–9 (enzyme 3 on the graph). Most cellular enzymes work best at a pH of 7 (enzyme 2 on the graph), but this varies throughout the cell depending on the type of reactions being carried out and the unique conditions within the organelle.

At the optimum pH, the enzyme is surrounded by the appropriate balance of positive and negative charges to allow it to adopt its correct shape and therefore its specific active site. This means that reaction rate will be fastest because the enzyme is specific and complementary for its substrate.

At pHs outside of the optimum pH, in either direction, the number of positive and negative charges in solution changes and this affects the shape of the enzyme. Ions in the environment which were preferentially hidden away from the enzyme's active site in the optimum solution may well move to be nearer it, while ions on the outside of the enzyme may be repelled towards the centre of the enzyme by the surroundings. This causes disruption to the shape of the enzyme, denaturing it.

As with temperature, minor movements away from the optimum pH can be tolerated. If the reaction rate falls, they will not cause it to stop entirely. A return to the optimum pH will allow the enzyme to renature and rate to return to its optimum. However, the further away from the optimum pH the reaction rate falls, the lower the reaction rate will be until reaction rate reaches zero. The enzyme will not be able to be reversed.

Effect of substrate and enzyme concentration

Substrate and enzyme concentration affects rate because of the availability of molecules to either be catalysed or do the catalysis. In either case, as the concentration increases, the rate increases until another factor limits the rate of reaction. For instance, as substrate concentration increases, the rate increases until the concentration of enzymes limits the rate of reaction because there are simply not enough enzymes to catalyse the reaction any faster, and adding more substrate does not improve this situation, but rather creates a longer queue! Similarly, if enzyme concentration is increased, reaction rate increases until the substrate concentration reduces because all the substrates have been converted into products. At this point, enzymes are empty and unable to catalyse the reaction because of the lack of substrate. In both scenarios, adding more of the other part (more enzyme if substrate concentration is increasing, or more substrate if enzyme concentration is increasing) until the limiting factor is reached. This assumes that temperature and pH are both optimal.

Test your knowledge

- Describe the effect of a 10°C temperature increase up to the optimum on rate of enzyme reaction.
- Draw a graph to show how pH affects the rate of enzyme reaction.
- Describe how the rate of reaction changes as enzyme concentration changes.

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Recap questions: Enzymes as biological catalysts

- 1 Amylase, present in the mouth to digest starch into maltose, is denatured in boiling water. Explain why. (5 marks)
- 2 Hyperthermia (extreme heat) can be fatal. Explain why this is the case, with reference to enzymes. (5 marks)
- 3 If supermarket trolleys are substrates, and checkout tills are enzymes, use a graph to show how increasing enzyme concentration will increase rate. (2 marks)

C3 Homeostasis



Key points covered

- The purpose and general function of homeostasis
- Examples of negative feedback loops and how they control body function
- Positive feedback loops and how they control body function
- The roles of the autonomic nervous systems
- The role of the endocrine system and adrenaline response
- What happens if homeostasis fails

What is homeostasis?

Homeostasis is the maintenance of a constant internal environment. As the conditions move away from the optimum, receptors detect the change and communicate this to the control centre. The control centre coordinates the response and sends a message to the effector which leads to changes that bring conditions back to the optimum once more.

Test your knowledge

- Give a definition for the term homeostasis.
- Draw a flow chart to show the process of homeostasis.



Response leads to restoration of optimal conditions

Response coordinated and communicated to effector

Condition

Negative feedback loops

Negative feedback loops operate an internal braking system. The production of one response inhibits the driver of that response, and this provides a self-regulating system.

As the second response decreases, the inhibition of the first response decreases too. This enables the first response to trigger more of the second response, which increases inhibition of the first response. This keeps the level at a relatively constant rate because acceleration by the first response causes the second response to slow down the first response.

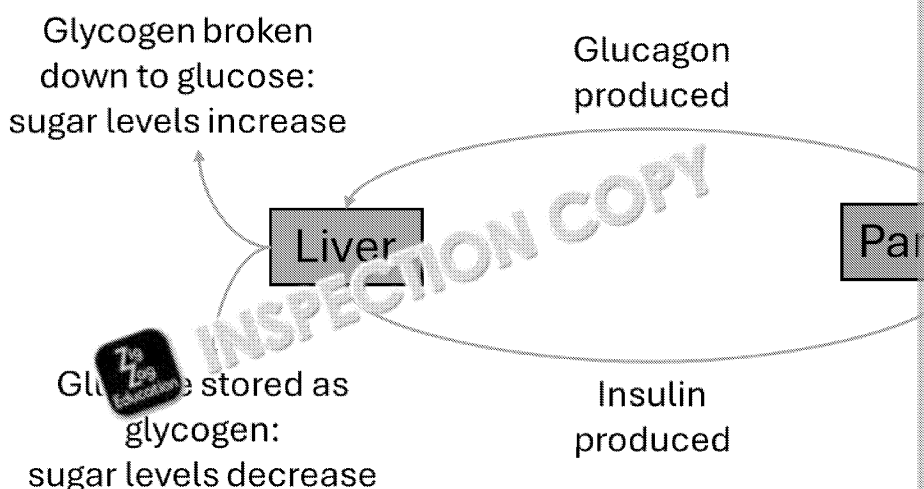
inhibits

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Blood sugar levels

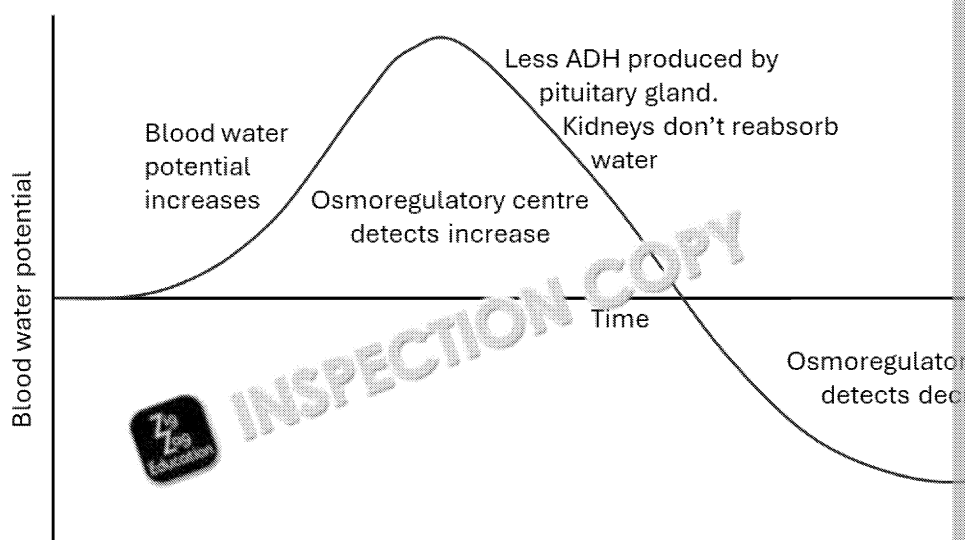
Blood glucose level homeostasis is maintained through negative feedback. As glucose is triggered to release insulin, which drives the liver to convert glucose into glycogen, the blood sugar level, causing the reduction in insulin production and, if levels decrease, glucagon instead. Glucagon encourages the liver to break down glycogen into glucose in the blood. This continuous adjustment maintains blood glucose homeostasis.



Body fluids (osmoregulation) and blood pressure

Osmoregulation (blood water potential) is regulated by the osmoregulatory centre and controlled by the kidneys. As blood water potential increases, the osmoregulatory centre detects the increase and causes the pituitary gland to secrete less ADH (antidiuretic hormone). Less ADH is inserted into the kidney tubules, so less water is reabsorbed, and more is sent to the bladder. More dilute urine is produced, and the water potential of the blood returns to optimum.

If water potential decreases, this too is detected by the osmoregulatory centre and causes the pituitary gland to secrete more ADH. More aquaporins are inserted into the kidney tubule walls, meaning more water is reabsorbed and less water is sent to the bladder for excretion. More concentrated urine is produced, and the water potential of the blood returns to optimum levels again.



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Blood pressure changes due to many different factors, including the restriction or increase in blood volume and concentration of solutes such as sugar, and changes triggered by changes in blood pressure. Managing blood pressure is by adjusting blood volume. Baroreceptors present in the walls of blood vessels can also trigger the pituitary gland to either increase or decrease its release of ADH, similar to the same as above.

Gas concentration

Gas concentration in the blood is controlled by gas exchange at the lungs; however depending on the concentration of carbon dioxide and oxygen in the blood. As chemical sensory receptors detect the change and send nerve signals to the diaphragm and breathing (inhalation and exhalation) and thus the release of carbon dioxide from the blood and oxygen into the blood via the lungs. If carbon dioxide levels increase quickly, these levels change frequently, increasing breathing rate, leading to deeper and more rapid breathing.

Test your knowledge

- Draw a summary process diagram to show how homeostasis is maintained with respect to gas concentration in the blood.
- Draw and label a graph to show how blood glucose levels change over the course of a day. This is monitored and responded to.

Positive feedback loops

Fewer systems are controlled by positive feedback loops. These operate by triggering a response which increases the number of triggers, thus increasing the strength of response further. This drives the situation increasingly quickly away from the initial starting point, as the response escalates.

Blood clotting

Blood clotting is one example of a positive feedback loop. This is desirable because blood clotting reduces blood loss and therefore increases survival chances. Blood clotting occurs when a blood vessel is damaged. Substances released by the damaged blood vessel wall cells encourage platelets to accumulate at the site. These platelets in turn release more of this hormone, which attracts more platelets to the site, which release more hormone, attracting more platelets. This causes the very rapid accumulation of platelets at the site of injury, leading to a swift and growing blood clotting response.

Labour contractions

Labour contractions are another example of a positive feedback loop. In preparation for birth, a mother will experience uterine contractions which increase in frequency and intensity as the baby being born. This is caused by the pituitary gland releasing a hormone called oxytocin, which triggers uterine contractions, which push the baby down towards the birth canal. As the baby moves through the cervix, this makes the cervix trigger the pituitary gland to release more oxytocin. Uterine contractions increase in frequency and intensity as this positive feedback loop escalates and only ceases when the baby has been delivered.

Test your knowledge

- Use the flow chart to convert either the blood clotting or the labour contractions into a positive feedback loop diagram.

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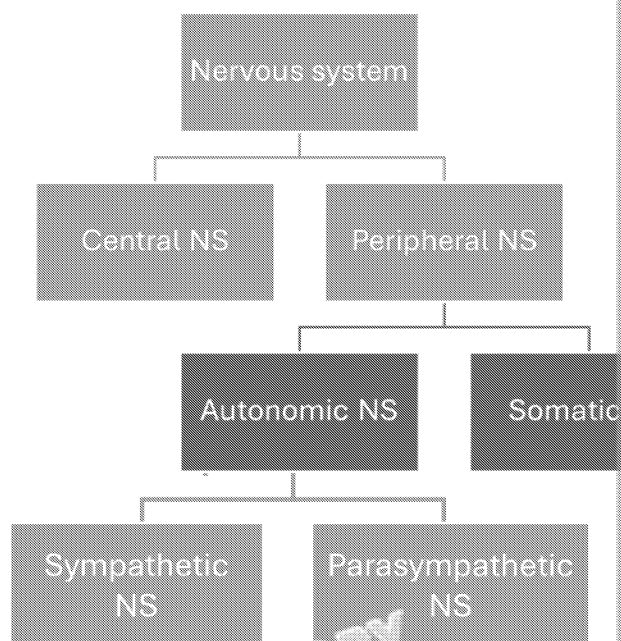
The nervous and endocrine systems

The nervous and endocrine (hormone) systems are interconnected, and each can trigger their own system. For instance, in the positive feedback example of labour contraction, the cervix sends nervous signals to the pituitary gland, which triggers the release of oxytocin. Likewise, the detection of changes in blood water potential by the hypothalamus causes the release of ADH to the pituitary gland which causes it to change the amount of ADH released, which is then reabsorbed by the kidneys.

The nervous system produces short-lived, rapid responses, using neurones to convey messages throughout the body. The endocrine system, conversely, produces responses that last longer (and are often reversed, such as with growth or puberty) but are delivered more slowly, using the bloodstream to carry messages with hormones which attach to receptors on the target cells.

Nervous system	Endocrine system
Rapid response	Slower response
Short-lived response	Longer-lived response
Uses neurones	Uses bloodstream
Electrical signals	Hormonal signals
Triggers muscles or glands	Triggers target cells via receptors

The peripheral nervous system



The nervous system is divided into two main parts: the central nervous system and the peripheral nervous system. The central nervous system consists of the brain and spinal cord, and the peripheral nervous system consists of all the other parts of the nervous system. As a result, the central nervous system contains only neurones, whereas the peripheral nervous system contains sensory and motor neurones.

The peripheral nervous system is further subdivided into the somatic nervous system and the autonomic nervous system. The somatic nervous system is responsible for all consciously controlled responses: any response that is consciously controlled happens through the somatic nervous system. Sensors (such as for taste, smell, touch, temperature, and pressure) send signals through sensory neurones, and responses are coordinated and sent to effectors (skeletal muscles or glands). The autonomic nervous system controls unconscious responses.

All nervous responses start with a stimulus, which is detected by a receptor. The signal is then passed through sensory neurones to the control centre (usually the brain but could be the spinal cord or an endocrine gland). The response is then coordinated via relay neurones. The effect is transmitted via motor neurones to an effector.

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The autonomic nervous system

The autonomic nervous system is responsible for the automatic control of body systems such as heart rate and digestion. These systems are adjusted automatically and without conscious thought. They are involuntary. They operate using sensors which detect changes to the status. This data is sent to a control centre (such as the brain or the pancreas), which coordinates the response. It then sends signals to muscles (smooth or cardiac) or glands, to deliver an appropriate response. Any part of the nervous system can be part of the autonomic nervous system. Any part of the nervous system can be part of the autonomic nervous system. Any part of the nervous system can be part of the autonomic nervous system.

The autonomic nervous system is divided into two further sections: the sympathetic and the parasympathetic. The sympathetic system controls the excitation responses, including the adrenaline response. The parasympathetic system controls the relaxing responses, including sleep.

The role of the hypothalamus

The hypothalamus is a small section of the brain which is responsible for controlling the autonomic nervous system. It has two areas which deal with the sympathetic and parasympathetic nervous systems. It is responsible for many functions, including behavioural responses such as sleeping, eating and drinking. It also maintains homeostasis through negative feedback loops, including blood glucose levels, and it releases hormones which trigger hormonal production in other parts of the endocrine system, including the pituitary gland, which is situated directly next to it. It therefore has both nervous and hormonal functions, which enables the efficient communication between these two systems.

The adrenaline response

The adrenaline response is a sympathetic nervous system response leading to a fight or flight response. When a situation is encountered which triggers the adrenaline response, such as an emergency or an unpredictable situation, the hypothalamus triggers a range of nervous and hormonal responses.

One set of responses involves the activation of the sympathetic nervous system which causes the contraction of smooth muscles and also activates the adrenal medulla which causes the release of adrenaline hormones into the blood.

A second set of responses involves the hypothalamus activating the adrenal-cortical axis. This causes the pituitary gland to release a hormone which travels to the adrenal cortex which causes the release of cortisol hormones into the blood.

The adrenal glands are found on the top of the kidneys and are responsible for the production and release of hormones. Those hormones released through the adrenaline response cause many physiological responses in the body, including:

- an increase in heart and breathing rate, to ensure maximum oxygen saturation;
- a suppression of digestion and the redirection of blood from the digestive system to the muscles (such as those required for running);
- the dilation of pupils, to maximise visual acuity;
- diversion of brain function away from managing the situation to solely concentrating on the response to the situation;
- a release of glucose from storage (liver and muscle cells), to enable high energy levels for maximum ATP production.

This complex, multifaceted response involves the nervous and endocrine systems working together in the body, in a rapidly produced response which can last a reasonably long time.

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Test your knowledge

- Create a tree diagram to show how the nervous system is divided into its sections.
- Create a flow diagram to show how the adrenaline response occurs through the nervous and hormonal responses.
- List the physiological responses of the adrenaline response.

Disturbing homeostasis

If homeostasis is not maintained, ill health will be experienced. The ability to maintain homeostasis is not constant and can be affected by a number of factors. As the body's ability to maintain homeostasis effectively deteriorates, as nervous pathways, receptors and effectors become less responsive. Glands which produce hormones can produce less hormone, which means feedback loops become less effective because less hormone is released initially, meaning less of a response happens as a result. Disturbances in thyroid hormone levels are common in older people and can affect the ability to maintain blood pressure. This can lead to damage to the heart which, when coupled with other conditions such as diet and exercise levels, can lead to life-critical conditions such as diabetes, and stroke.

Drug and alcohol abuse cause a great range of damage to the body, including by causing damage to heart and cardiovascular tissue, affecting heart rate and rhythm control, including to the liver, lungs and digestive system, and impairment in brain function. This can lead to loss of memory and other conscious functions such as reasoning, so that impairment of core functions such as homeostasis and sleep.

These effects build up gradually over time, and do not require excessive use of cocaine, to be visible. Occasional cigarette smoking, and drinking small but regular amounts of alcohol, can have these impacts if sustained over a long period of time.

Maintenance of a well-balanced diet, regular and varied exercise and only small amounts of alcohol consumption will enable the body to maintain optimum health conditions for as long as possible, with major events excluded.

Test your knowledge

- List three factors which can cause a reduction in the body's ability to maintain homeostasis.
- State two symptoms of poor homeostatic control.



Recap questions: Homeostasis

- 1 A student drinks a large quantity of a very sugary drink. Describe and explain the effects on the body. (8 marks)
- 2 Another student is about to deliver their first major presentation to their school. They are feeling nervous and have sweaty palms, a high heart rate, and are aware of their effects have been triggered. (5 marks)
- 3 Plants respond to hormones through positive feedback loops, too. The hormone ethylene promotes fruit ripening via a positive feedback loop. Describe how this response occurs, based on your understanding of human positive feedback loops. (5 marks)

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Answers for Recap Questions

A: Structure and function of cells and tissues

Structure and function of cells and tissues

1. Plasma membrane: controls entry and exit of substances from the cell (1); cell capsule provides protection from external environment (1)
Each part provides a different form of control or protection (1)
Without which the bacteria would be more susceptible to damage from external environment (1)
2. Cell wall provides structural support and strength (1); the vacuole provides turgor pressure (1)
3. Stained with crystal violet (1); fixed with iodine (1); washed with alcohol (1); applied to slide (1); viewed under microscope (1)
4. Prevent the lens and slide/specimen from touching each other (1); which could damage the specimen (1)
5. Correctly arranged equation: $A = \text{image size} \div \text{magnification}$ (1)
Correct use of equation: $A = 143.56 \text{ mm} \div 148\,000$ (1)
Conversion of mm to μm , either before the division or after (1)
Correct answer in micrometres: $0.97 \mu\text{m}$ (1)

Structure and function of specialised cells in multicellular organisms

1. Two potential symptoms from: difficulty stopping bleeding; bruising under the skin (2)
Dangerous if left untreated because could lead to significant blood loss (1)
2. Damage will be removal of hair-like protrusions from root hair cells (1)
This will significantly reduce the surface area of the root (1)
And reduce the uptake of water and mineral ions into the plant (1)
3. Lignin required for xylem walls (1)
Makes walls impervious to water (1)
Lack of lignin may prevent water and mineral ions from being transported up the plant (1)
This would slow growth, causing dwarfism (1)

Structure and function of biological tissues

1. Cells are in one layer thick (1) and closely grouped together (1) to form a membrane (1)
Epithelial tissue forms linings of lungs and other organs (1)
Endothelial tissue forms lining of blood vessels (1)
2. Myofibrils are aligned end to end (1) and fibrils form fibres (1) which form muscles (1)
Fibres are aligned in the same direction (1) so that when muscle fibres contract they create a larger effect (1)
3. During resting potential, the membrane is impervious to sodium ions because it is closed (1), partially permeable to potassium ions because the potassium ion gateway is open (1).
On activation, the sodium ion channel opens and the membrane becomes permeable to sodium ions (1).
This increases the membrane potential (1) as sodium ions diffuse into the axon (1).
The potassium ion gateway opens, increasing the permeability of the membrane to potassium ions (1).
This decreases the membrane potential (1) as potassium ions diffuse out of the axon (1).
After hyperpolarisation, membrane potential returns to the resting potential and the axon returns to its normal state (1).

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