

Solutions to Nuclear, Particle and Medical Physics – Test B

1. (a) ${}_{90}^{234}\text{Th} \rightarrow {}_{91}^{234}\text{Pa} + {}_{-1}^0\beta + \bar{\nu}$
 Accept e^- in place of beta; accept subscript
 e on antineutrino and A / Z values of 0
 on antineutrino
 Thorium written correctly [1]
 Pa as a product [1]
 234 and 91 correctly written with product
 symbol (even if not Pa) [1]
 0 and -1 correctly written with product [1]
 (b) charge = $1.6 \times 10^{-19} \times 1.44 \times 10^{-17}$ (C) [1]
 $1.6 \times 10^{-19} \times 234 = 3.91 \times 10^{-25}$ (kg) [1]
 mass ratio = $\frac{1.44 \times 10^{-17}}{3.91 \times 10^{-25}} = 3.68 \times 10^7$ [1]
 C kg^{-1} [1]

2. Up to 0.5 fm / 5×10^{-16} m [1]
 3. A photon [1]
 Incident on a nucleus [1]
 Produces a particle [1]
 and its equivalent antiparticle [1]
 4. Momentum [1]
 Energy [1]
 5. gamma radiation is ionising [1]
 gamma radiation has long range in air / is highly
 penetrating [1]
 Then any two from:
 • point the source away from people
 • keep the source as far from you as possible [1]
 • stand behind a lead shield [1]
 • never store the source inside a lead container when
 not in use [1]
 6. (before Rutherford) plum pudding model / atoms
 were solid balls [1]
 Rutherford model showed atoms were mostly empty
 space [1]
 nucleus known to contain positive charge and be at
 the centre [1]
 electrons in shells / different energy levels / orbitals [1]
 7. intensity on the vertical axis [1]
 $1 / \text{distance}^2$ on horizontal axis [1]
 graph should be a straight line [1]
 with positive gradient / gradient is k in $I = \frac{k}{x^2}$ [1]
 8. background count rate is less significant / count rate
 to a small percentage of the count rate [1]
 accuracy not significantly affected [1]

9. (a) $\ln A$ calculated
 all values
 and no units

t / s
0
60
120
180
240
300
360
420

- (b) graph with
 horizontal
 scaled so
 the grid
 all points
 straight line
 (c) gradient
 the line
 all values
 calculation
 negative
 if all correct
 (d) numerical
 positive
 unit as s

10. $\lambda = 0.693 / 1.9$
 3.59×10^{-4} years
 $100 = 77 \times e^{-\lambda t}$
 $\ln 77 - \ln 100 = -\lambda t$
 728 years [1]

11. charge on alpha
 $= 3.2 \times 10^{-19}$
 charge on nucleon
 $= 1.26 \times 10^{-14}$
 $\frac{8.99 \times 10^9 \times 3.2 \times 10^{-19}}{(3.59 \times 10^{-4})^2}$
 296 N [1]

12. reactants 5.02×10^{-12}
 products 5.01×10^{-12}
 difference 0.01×10^{-12}
 $0.019431 \times 1.6 \times 10^{-19}$
 3.227×10^{-29}
 3.227×10^{-29}
 2.90×10^{-12}
 $2.90 \times 10^{-12} \times 3.227 \times 10^{-29}$
 3.49×10^{12} J

INSPECTION COPY

COPYRIGHT
 PROTECTED



13. (a) *Diagram/description to include:*
 Collimator [1]
 Scintillator [1]
 Detector [1]
 computer / software / circuitry / display [1]
Functions to include two from:
 collimator selects gamma rays / photons from one direction only [1]
 scintillator amplifies signal / converts one gamma photon to many visible photons
 detector detects visible photons [1]
14. (a) attenuation [1]
 (b) photoelectric effect [1]
 X-ray photon absorbed by electron/atom [1]
 (photo) electron emitted [1]
15. (a) acoustic impedance [1]
 (b) (i) blood is denser / liquids are denser than gases [1]
 particles/molecules packed more tightly [1]
 vibrations/compressions in longitudinal wave more easily passed on [1]
 (ii) λ in air = 3.3×10^{-5} m [1]
 λ in blood 1.5×10^{-4} m [1]
 difference = 1.17×10^{-4} [1]

$$\left(\frac{1.17 \times 10^{-4}}{3.3 \times 10^{-5}} \times 100 \right) = 355 \%$$
 [1]

COPYRIGHT
PROTECTED



Preview of Answers Ends Here

This is a limited inspection copy. Sample of answers ends here to stop students looking up answers to their assessments. See contents page for details of the rest of the resource.