

Solutions to Nuclear, Particle and Medical Physics – Test A

- (from 0.5) up to 3 fm / 3×10^{-15} m [1]
- Rest mass is the same [1]
Charge is opposite (or both zero) [1]
 - Annihilation [1]
Photons produced [1]
- $${}_{92}^{238}\text{U} \rightarrow {}_{90}^{234}\text{Th} + 2\alpha$$

Accept He in place of α [1]
Uranium written correctly [1]
as a [1]
at 190 correctly written with product
tool (even if not Th) [1]
4 and 2 correctly written with other product [1]
 - charge = $1.6 \times 10^{-19} \times 2 = 3.2 \times 10^{-19}$ (C) [1]
mass = $1.67 \times 10^{-27} \times 4 = 6.68 \times 10^{-27}$ (kg) [1]
charge : mass ratio = $\frac{3.2 \times 10^{-19}}{6.68 \times 10^{-27}} = 4.79 \times 10^7$ [1]
- alpha particles are strongly ionising [1]
alpha particles have short range in air [1]
Then any two from:

 - point the source away from people [1]
 - keep the source as far from the body as possible [1]
 - stand behind a screen [1]
 - put the source back inside a lead container when not in use [1]
- alpha particles are fired at a thin metal foil [1]
most alpha particles pass straight through [1]
shows the atom is mostly empty [1]
some alpha particles are deflected through small angles, which shows there is charge present [1]
very few alpha particles were deflected through angles $> 90^\circ$, which shows that there is a strong charge density at the centre of the atom [1]
- use a gamma source [1]
use a Geiger counter / GM tube and counter [1]
measure the count rate at different distances in a straight line from the source [1]
correct for background radiation [1]
- background count rate is significant / contributes to a large percentage of the count rate [1]
results will be more accurate [1]
- $\ln 4$ calculated [1]
all values correct and to the same number of s.f. and no unit [1]

t / s	A / s ⁻¹	$\ln A$
0	852	2.95
60	402	2.67
120	287	2.46
180	148	2.17
240	91	1.96
300	42	1.62
360	29	1.46
420	19	1.28

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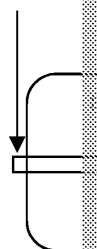
9. $\lambda = 0.693$
 7.60×10^{-7}
 $100 = 65 \times$
 $\ln 65 - \ln 100$
56.7 hours

10. charge on a
 $= 3.2 \times 10^{-19}$
charge on a
 $= 2.08 \times 10^{-19}$
 8.99×10^9

95.7 mN

11. reactants 2
products 2
difference
 0.191×1.6
 3.17×10^{-19}
 3.17×10^{-19}
 2.86×10^{-19}
 2.86×10^{-19}
 1.72×10^{-19}

12. (a)



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- (b) high-speed electrons / electrons with high kinetic energy [1]
 (electrons) arrive at / hit the anode [1]
 loss of k.e. converted to X-ray photon(s) [1]
Must refer to photons not just X-rays

13. (a) loss of intensity [1]
 (b) Compton effect [1]
 X-ray photon makes inelastic interaction with an electron [1]
 X-ray photon loses energy / increases in wavelength [1]

14. (a) acoustic impedance [1]
 (b) (i) bone is more dense / solids are more dense than gases [1]
 particles are packed more tightly [1]
 (ii) compressions in longitudinal wave more easily passed on
 λ in air = 6.6×10^{-5} m [1]
 λ in bone 8.2×10^{-4} m [1]
 difference = $7.5(4) \times 10^{-4}$ [1]
 $\left(\frac{7.5 \times 10^{-4}}{6.6 \times 10^{-5}} \times 100 = \right) 1100 \%$ [1]



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